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1 Nature, distribution and origin of CO₂ on Enceladus

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52 Abstract

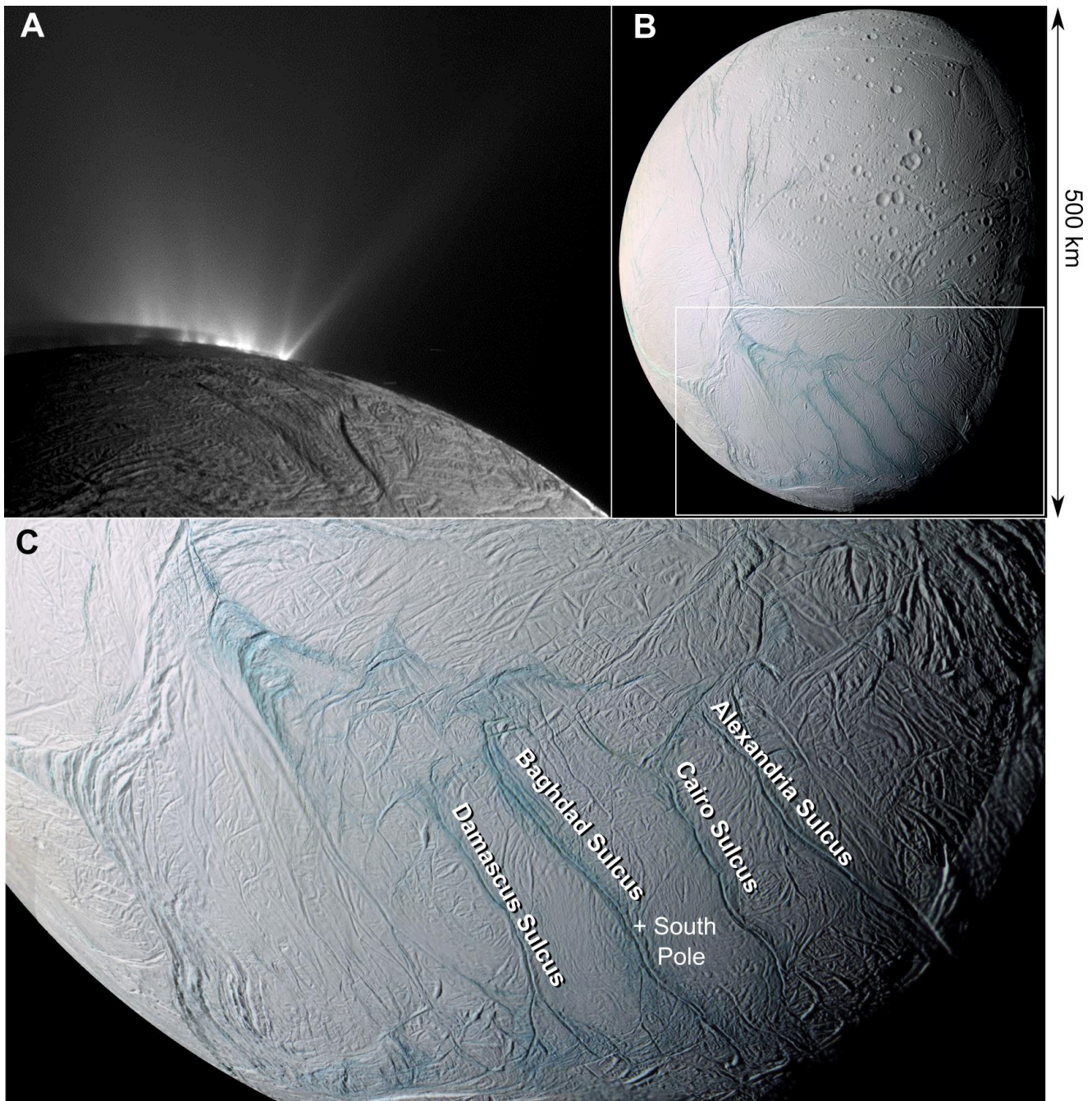
53 We present the first map of CO₂ at the surface of Enceladus using data obtained by the *Cassini*
54 Visible-Infrared Mapping Spectrometer (VIMS). In order to measure the weak and narrow CO₂
55 absorption band depths around 4.26 μm, we improved: (1) the calibration of VIMS spectra; (2) the
56 calculation of geographic coordinates; and (3) the projection techniques. We averaged multiple
57 observations of a given area to obtain a signal to noise ratio high enough to map the CO₂ abundances.
58 CO₂ is reliably detected mostly in the South Polar Region. This region includes active faults (Tiger
59 Stripes), the highest observed surface temperatures, and locations of active plume eruptions. The
60 occurrence here of CO₂ is consistent with an endogenic origin controlled by tectonics. Both pure CO₂
61 ice and complexed CO₂ are detected from the positions of absorption bands between 4.27 and 4.24
62 μm. The highest concentrations of CO₂ are between the main active faults of the South Polar Region,
63 where the surface temperature is low. Pure CO₂ ice deposits at the surface of Enceladus are best
64 modeled by the formation of gas pockets below the icy crust and above the surface of the internal
65 ocean. These pockets eventually release cold CO₂ gas (~70 - ~119 K) at low-velocity (seeping)
66 between the Tiger Stripes [Matson et al., 2018, *Icarus*, 302, 18-26]. CO₂ clathrate hydrates may form
67 in the ocean and may be subsequently released when a CO₂ gas pocket blows out and erupts. Other
68 mechanisms may contribute to reinforcing the anti-correlation of the CO₂ distribution (of any type)
69 with respect to the location of the Tiger Stripes, such as successive sublimation of CO₂ and
70 condensation on colder areas, and partial frost cover by H₂O releases from plume eruptions.

71 Highlights

- 72 • We present the first map of CO₂ at the surface of Enceladus by *Cassini* VIMS.
- 73 • The distribution of CO₂ in the South Polar Region suggests an endogenic origin and control
74 by tectonic activity.
- 75 • High surface temperature likely explains why the highest concentrations of CO₂ are measured
76 between the Tiger Stripes and not directly on them.
- 77 • Pure surface CO₂ could be vented at low-velocity (seeping) as a cold gas (~70 - ~119 K) from
78 gas pockets that form between the internal ocean and the ice lid along the main active faults.
- 79 • Complexed CO₂ is detected.
- 80 • CO₂ clathrate hydrates, which form crystal structures, may originate in the internal ocean and
81 may be eventually released to the surface by gas pocket eruptions.

82 1. Introduction

83 Plumes of material are observed erupting from four linear depressions in the South Polar Region
84 (SPR) of the Saturnian satellite Enceladus (Dougherty et al., 2006; Hansen et al., 2006; Porco et al.,
85 2006), which is evidence for geophysical processes operating today within Enceladus (Fig. 1). These
86 depressions have distinctive albedo (darker) and color (bluer) than most of Enceladus' surface and are
87 known by the colloquial name of Tiger Stripes (Porco et al., 2006; Johnson, 2005). The Tiger Stripes
88 consist of four subparallel active faults ~130 km long, ~2 km wide, separated by ~35 km on average.
89 Each fault is about 500 m deep and bordered by a 100 m high ridge (Porco et al., 2006). Since the
90 beginning of the *Cassini* mission, the Ion Neutral Mass Spectrometer (INMS, Waite et al., 2004) and
91 the Visual and Infrared Imaging Spectrometer (VIMS, Brown et al., 2004) have been used to determine
92 the composition of the erupting material. Both INMS (Waite et al., 2006, 2009; Glain et al., 2015;
93 Bouquet et al., 2015) and VIMS (Hansen et al., 2006, 2011) indicate that water vapor is the main
94 component. The current quantitative compositional estimates (Waite et al., 2017) are ~96-99% H₂O,
95 with the remaining few percent comprising of a mixture of H₂ (0.4-1.4%), CO₂ (0.3-0.8%), CH₄ (0.1-
96 0.3%) and NH₃ (0.4-1.3%). From Earth-based telescopic observations, the surface of Enceladus is
97 known to be rich in water ice (e.g. Grundy et al., 1999; Cruikshank et al., 2005; Emery et al., 2005),
98 but the *Cassini* Visible-Infrared Mapping Spectrometer (VIMS) also identified CO₂ ice on the surface
99 (Brown et al., 2006; Hansen and Romain, 2010), especially in the SPR. It is important to understand
100 the distribution of CO₂ on Enceladus because of its association with and likely role in generating
101 plume activity.



102

103 **Fig. 1: Enceladus observed by the *Cassini* Imaging Science Subsystem (ISS) acquired on December**
 104 **1, 2010. A – Plumes from the active faults in the South Polar region. B – Enhanced colors in the visible**
 105 **where the active faults appear in blue-green. C – Close-up view of the South Polar regions; the four**
 106 **active faults are labeled.**

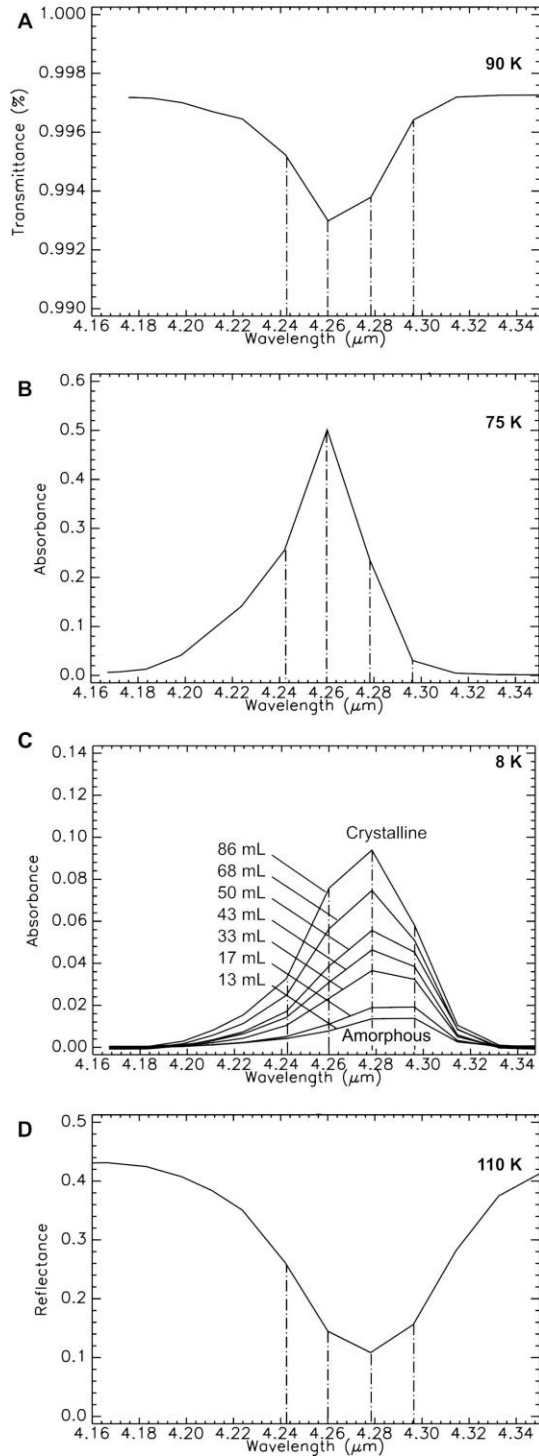
107 CO_2 is common in the outer Solar System and can exist as part of the body formation material; it
 108 can also be manufactured by chemistry using radiation or other energy sources (e.g., [Hudson and](#)
 109 [Moore, 2001](#)). CO_2 was found on the Galilean satellites by the *Galileo* Near Infrared Mapping
 110 Spectrometer (NIMS), apparently due to both interior outgassing ([Carlson, 1999](#)) and radiolysis
 111 ([McCord et al., 1997; 1998](#)), and also in a thin atmosphere around Callisto ([Carlson, 1999](#)). CO_2 was

112 also detected on the Uranian satellites (Grundy et al., 2006; Cartwright et al., 2015) and Triton (e.g.,
113 Cruikshank et al., 1993; Holler et al., 2016).

114 CO₂ has been reported by VIMS on the surface of other icy Saturn satellites (e.g. Cruikshank et al.,
115 2010; Filacchione et al., 2010). On Enceladus, CO₂ is likely part of the original condensates of the
116 satellites of Saturn (McCord et al., 1998). CO₂ trapped in amorphous water ice below 120 K possibly
117 formed in the outer regions of the solar nebula. Here, solid CO₂ in ice mixtures in molecular clouds
118 at 10-30 K has been observed [De Graauw et al., 1996], also explaining its presence in comets and
119 interstellar grains [e.g., Delsemme, 1982; Sandford and Allamandola, 1990; Whittet et al., 1996]. CO₂
120 also may be formed exogenically by irradiation of organic materials on the surface (Mennella et al.,
121 2004), such as those found in carbonaceous chondrite meteorites (Carlson, 1999), or above the
122 surface, in the plumes, although organics are not abundant in the plumes (Waite et al., 2017) or on the
123 surface of Enceladus. Endogenically, CO₂ could exist as clathrates or dissolved in the Enceladus
124 ocean water [e.g. Blake et al., 1991; Bouquet et al. 2015]. CO₂ is certainly being released in the plumes
125 that emanate from the South Polar Region of Enceladus. Matson et al. (2012) proposed that the origin
126 of the CO₂ was from degassing ocean water beneath the icy Enceladus crust. Due to the relatively
127 high surface temperatures (100-140 K) of the South Polar Region [Spencer et al. 2006; Abramov and
128 Spencer 2009; Taffin et al., 2012], CO₂ ice is not stable once exposed at the surface [McCord et al.,
129 1998; Palmer and Brown, 2008]. However, CO₂ could be stable if trapped or complexed, such as in
130 inclusions or clathrates [Brown et al., 2011]. Thus, any surficial CO₂ ice must be resupplied, and the
131 form and abundance of the remaining CO₂ may be altered with time.

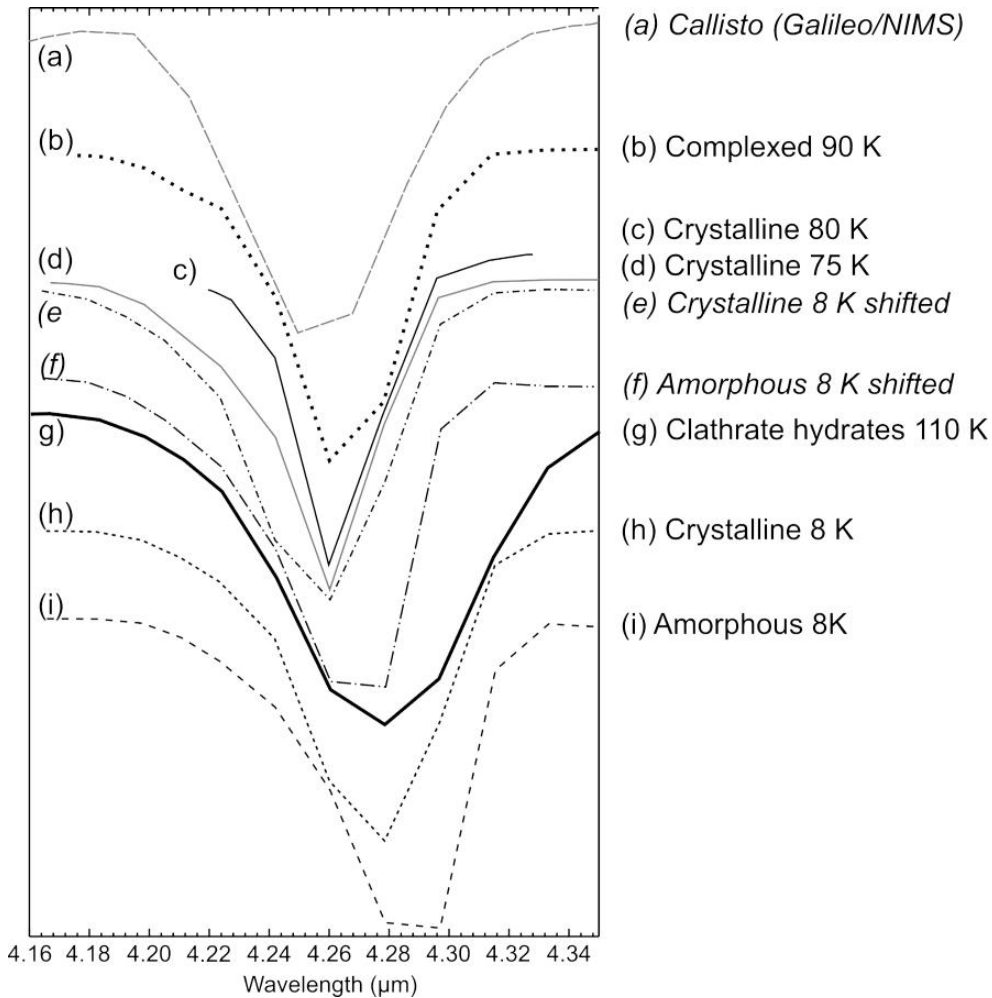
132 Near-infrared spectra of CO₂ (Fig. 2 and Fig. 3) exhibit a strong absorption band in the region 4.24-
133 4.27 μm , due to the asymmetric stretching mode of the CO₂ molecule, and a weaker absorption near
134 2.7 μm (Fig. 4), due to $\nu_2+\nu_3$ vibration transition modes (e.g. Chaban et al., 2007). The longer
135 wavelength absorption band position is sensitive to the nature and strength of molecular associations
136 between CO₂ and its neighbors. Free CO₂ ice absorbs at 4.268 μm (Sandford and Allamandola, 1990).
137 The difference between crystalline and amorphous CO₂ is small at VIMS spectral resolution, however
138 the center position of the absorption band of amorphous CO₂ seem to be slightly shifted to longer
139 wavelength compared to crystalline CO₂ (Fig. 2 and Fig. 3). Fig. 3 and Fig. 4 illustrate also that
140 complexed CO₂ absorbs near 4.247 μm (Chaban et al., 2007), whereas CO₂ clathrate hydrates absorb
141 between 4.28 μm and 4.30 μm (Oancea et al., 2012). This wavelength shift was observed, for example,
142 for Callisto's surface (McCord et al., 1997, 1998).

143 Here, we investigate the presence, type and spatial distribution of CO₂ on Enceladus' surface using
144 nine years of *Cassini* VIMS observations. We identify the forms of CO₂ present and their associations
145 with surface features and phenomena, and thereby provide a basis for testing models of internal
146 processes that may produce the plumes.



147

148 **Fig. 2: Laboratory spectra of CO₂ resampled at VIMS wavelengths showing the absorption band**
 149 **position of the asymmetric stretching mode of the CO₂ molecule as a function of the type of the**
 150 **physical structure and type of bonds. (A) Complexed CO₂ at 90 K (Bernstein et al., 2005). (B)**
 151 **Crystalline CO₂ ice at 75 K (Isokoski et al., 2013). (C) Transition from CO₂ amorphous ice to CO₂**
 152 **crystalline ice at 8 K (Escribano et al., 2013). (D) CO₂ clathrate hydrates at 110 K (Oancea et al., 2012).**

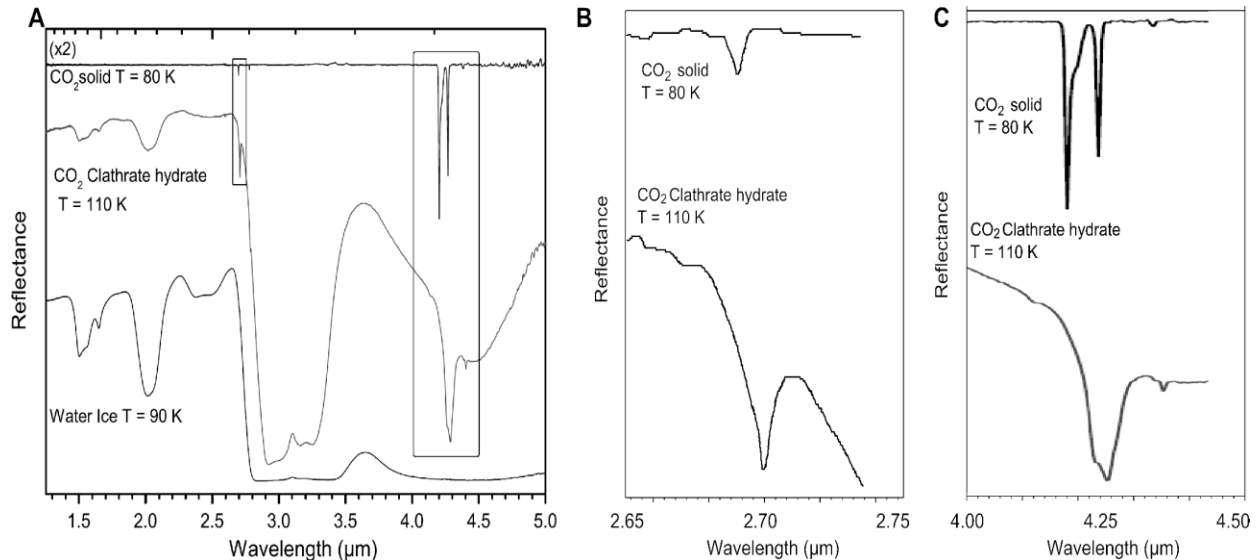


153

154 **Fig. 3:** Normalized spectra of CO₂-rich materials resampled at VIMS wavelengths showing the
 155 absorption band position variation of the asymmetric stretching mode of the CO₂ molecule. Labels in
 156 italics indicate spectra that are not from laboratory measurements or that are empirically shifted in
 157 wavelength to account for temperature effects. (a) CO₂-rich region on Jupiter's satellite Callisto
 158 measured by *Galileo* NIMS (McCord et al., 1998), and interpreted as complexed CO₂. (b) Complexed
 159 CO₂ at 90 K (Bernstein et al., 2005). (c) Crystalline CO₂ at 80 K (Escribano et al., 2013). (d) Crystalline
 160 CO₂ at 75 K (Isokoski et al., 2013). (e) Crystalline CO₂ at 8 K (Escribano et al., 2013), shifted towards
 161 shorter wavelengths to match the wavelength position of crystalline CO₂ (c) and (d) measured at
 162 higher temperatures¹. (f) Amorphous CO₂ ice at 8 K (Escribano et al., 2013) shifted towards shorter
 163 wavelength by the same amount as (e) in order to account for the higher temperatures of the surface

¹ The shifted spectra Fig. 3e and Fig. 3f illustrate the theoretical position of the CO₂ absorption band depth of amorphous CO₂ at 75 or 80 K (between 4.26 and 4.28 μm) relevant for the surface of Enceladus. It is only theoretical because there is no spectrum of amorphous CO₂ ice available at the temperature of Enceladus. The only spectrum of amorphous CO₂ ice that we could find was acquired at 8 K (with an absorption centered between 4.28 and 4.30 μm), alongside a spectrum of crystalline CO₂ ice. Thus, the spectrum of crystalline CO₂ ice is used as a reference to prove that a temperature change from 8 K to 75 or 80 K shifts the CO₂ absorption band center of crystalline CO₂ ice from 4.28 to 4.26 μm. Assuming that for amorphous CO₂ ice a temperature increase from 8 K to 75 or 80 K results in the same shift of 0.02 μm in the absorption band center position as for crystalline CO₂ ice, the plot indicates that amorphous CO₂ is expected to have an absorption centered between 4.26 and 4.28 μm at 75 or 80 K instead of between 4.28 and 4.30 μm at 8 K.

164 of Enceladus¹. (g) CO₂ clathrate hydrates at 110 K (Oancea et al., 2012). (h) Crystalline CO₂ at 8 K
 165 (Escribano et al., 2013). (i) Amorphous CO₂ ice at 8 K (Escribano et al., 2013).



166

167 **Fig. 4: High-resolution laboratory spectra of CO₂ ice and CO₂ clathrate hydrates by Oancea et al.**
 168 **(2012). A – General view. B – Close-up view of the 2.7 μm absorption band of CO₂. C – Close-up view**
 169 **of the 4.25 μm absorption bands of CO₂.**

170 2. Spectral analysis of VIMS data to detect and map CO₂

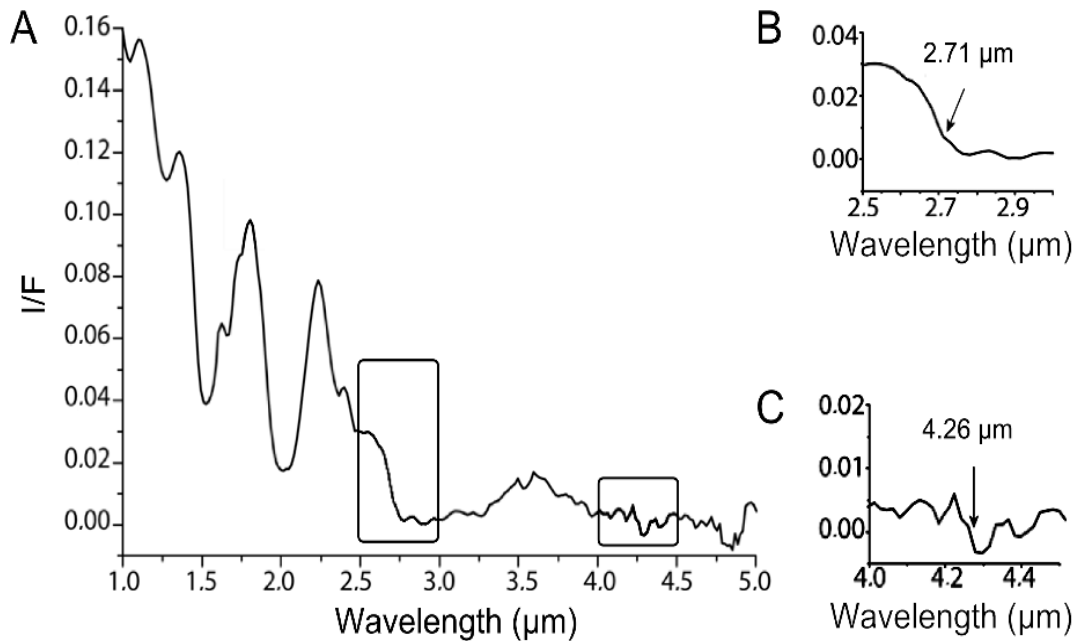
171 2.1 Instrument specifications

172 VIMS (Brown et al. 2004), on board the *Cassini* spacecraft, observed the Saturn system between
 173 February 7, 2004 (orbit insertion occurred on July 1, 2004), and June 30, 2017 (Cassini dove into
 174 Saturn on September 15, 2017), according to archives on the Planetary Data System. For our analysis,
 175 we used 1760 VIMS observations from 31 *Cassini* Enceladus flybys (Table 1 in the Appendix). The
 176 infrared part of VIMS had a linear array detector: one exposure acquired data in whiskbroom mode
 177 for a single spatial pixel. From the dispersion of the image of this pixel on the focal plane, each detector
 178 received photons from the same field of view in a different contiguous waveband. The two-
 179 dimensional image of any scene was acquired sample by sample and line by line.

180 In the wavelength region of the CO₂ spectral features studied here (4.24–4.28 μm and 2.70 μm),
 181 VIMS data have a spectral sampling width of about 0.0166 ± 0.0006 μm. This spectral sampling is
 182 sufficient to detect all the CO₂ spectral features. However, VIMS spectra cannot resolve the CO₂
 183 absorption band at 2.70 μm. Moreover, the band position shifts due to differences in molecule
 184 neighbor sites (e.g. complexed CO₂ vs. pure CO₂ ice, or CO₂ clathrate hydrates vs. crystalline CO₂
 185 ice).

186 2.2 Calculation of CO₂ relative absorption band depths at 4.24-4.30 μm

187 One objective of this study is to characterize CO₂ absorptions bands (measurement of depth, width
 188 and position) on Enceladus. The focus of this section is to optimize the spectral analysis within a range
 189 of wavelengths that has a relatively low signal-to-noise ratio due to the broad and ubiquitous H₂O
 190 absorption band, the low solar flux of photons, and an instrument sensitivity near the edge of the
 191 detector that is less than in the center.
 192



193

194 **Fig. 5: VIMS average spectrum of the CO₂-rich part of the South Polar Region of Enceladus**
 195 **showing an absorption band at 4.26 μm and an inflection at 2.7 μm , both attributable to CO₂ (Oancea**
 196 **et al., 2012). A – General view, which indicates the low reflectance between 4.0 and 4.5 μm**
 197 **comparatively to wavelengths shorter than 2.5 μm , and the relative band depth of CO₂ compared to**
 198 **the H₂O ice absorption bands at 1.28, 1.5 2.0 and 3.0 μm . B – Close-up view of the 2.7 μm absorption**
 199 **band of CO₂. C – Close-up view of the 4.25 μm absorption band of CO₂.**

200 2.2.1 Preliminary step: Scaling and ratioing of VIMS spectra

201 The stretching mode of the CO₂ molecule creates an absorption band in a range of wavelength that
 202 has a low solar photon flux and where the ubiquitous H₂O ice absorbs most of the remaining light.
 203 As a consequence, the signal-to-noise ratio in the spectra is never optimum for the measurement of
 204 CO₂ absorption bands and positions. In particular, any photometric effect of calibration artifact may
 205 have an amplitude that is comparable with the CO₂ absorptions. In order to minimize the uncontrolled
 206 variations of the spectra, we performed scaling and ratioing of VIMS spectra.

207 Scaling is the division of each spectrum by its median value in the intervals 4.0 μm and 5.12 μm .
 208 This range avoids the effects of large variations in the overtone absorption bands of H₂O ice and of
 209 the Fresnel peak of crystalline H₂O ice that occur at shorter wavelengths. Therefore, the median value
 210 over this entire range of wavelengths is less sensitive to noise in the data than the value at one given
 211 wavelength. This is a common way to minimize the effects of albedo, illumination variations and
 212 possible remaining systematic calibration artifacts.

213 Ratioing is the division of each scaled VIMS spectrum by the average scaled spectrum of Enceladus.
 214 This processing step is a way to minimize the effects of mixture of local components (here, CO₂
 215 materials) with the average surface composition, which is mostly H₂O ice. Since the surface
 216 composition of Enceladus is very homogenous and is globally dominated by H₂O ice, and since plume
 217 fallout add fresh H₂O ice deposits that may mix with any CO₂ component, all VIMS spectra have deep
 218 and broad spectral features of H₂O. After ratioing, the absorption bands of the local component are
 219 more likely to be detected by the calculation of a simple band depth parameter. This preparation step
 220 was successfully applied on spectra of Ceres by the Visible and InfraRed mapping spectrometer (VIR)
 221 of Dawn mission for the search of H₂O ice (Combe et al., 2016; 2018).

2.2.2 Calculation of CO₂ band depths that vary in position between 4.24 and 4.30 μm

Free CO₂ ice and complexed CO₂ may be mixed within a VIMS pixel: the expected CO₂ absorptions may vary in width and in center position between 4.24 and 4.30 μm. When only one form of CO₂ is present, the absorption band is narrow enough that it only involves one VIMS wavelength channel (Fig. 5). On the other hand, when mixtures occur, up to four VIMS wavelength channels may be within the CO₂ absorption band, as illustrated in Fig. 4. The absorption band depth is calculated using the average of three channels between 4.18 and 4.22 μm and three channels between 4.32 and 4.36 μm, so as to define the shoulders of the absorption. This operation minimizes the effects of illumination and albedo variations, and the average value over three VIMS wavelength channels is less sensitive to random noise in the data or occasional spikes in the spectra than the pixel value taken at a single wavelength. In order to calculate the CO₂ absorptions band depths, we used VIMS wavelength channels centered at 4.24, 4.26, 4.28 and 4.30 μm.

Any possible detection of any CO₂ absorption band has to meet the following criteria: 1) a detection threshold is set to equal the standard deviation of the absorption band depth values, and 2) the pixels that meet the detection criterion must define spatially-coherent areas (i.e. several contiguous pixels). When values lower than one occur at more than one VIMS wavelength channel (4.24, 4.26, 4.28 and 4.30 μm), this is interpreted as mixture of CO₂ compounds that may include complexed CO₂, CO₂ ice and CO₂ clathrate hydrates.

2.3 Calculation of an inflection at 2.7 μm caused by a CO₂ absorption

All forms of solid CO₂ have a narrow absorption band at 2.7 μm (Fig. 4A). At VIMS spectral resolution and sampling, this theoretically creates an inflection in the spectra (Fig. 4B). Since the 2.7 μm absorption of CO₂ is weak and not fully resolved by VIMS, we cannot use the calculation of a 2.7 μm absorption band depth to quantify the CO₂. However, at this wavelength, the solar flux is one order of magnitude higher than at 4.26 μm, which is adequate to make use of this band as a detection criterion. This inflection, therefore, can be measured using a traditional absorption band depth calculation (Clark and Roush, 1984), assuming a linear continuum between surrounding wavelength channels.

2.4 Optimization procedure of the signal-to-noise ratio with VIMS data

CO₂ absorption bands at 4.24–4.28 μm are faint and their variations are close to the instrument detection limit. Since the signal-to-noise ratio of VIMS spectra is the main challenge to study absorption bands of CO₂, we made an optimization in four steps:

- 1- Selecting VIMS pixels that cover less than 20 km² (the area covered by each pixel is calculated from the pixel corner coordinates)
- 2- Selecting VIMS images based on the spectral signal-to-noise ratio (measured by the standard deviation of the dark current)
- 3- Projecting and spatially merging VIMS data that meet the two quality criteria (points 1 and 2) where the main faults on Enceladus are spatially resolved.
- 4- Binning all VIMS observations with a spatial sampling of 4.4 km × 4.4 km (19.36 km², or 1° per pixel) in order to increase the signal-to-noise ratio of VIMS spectra.

3. Results: Distribution and nature of CO₂ deposits on Enceladus

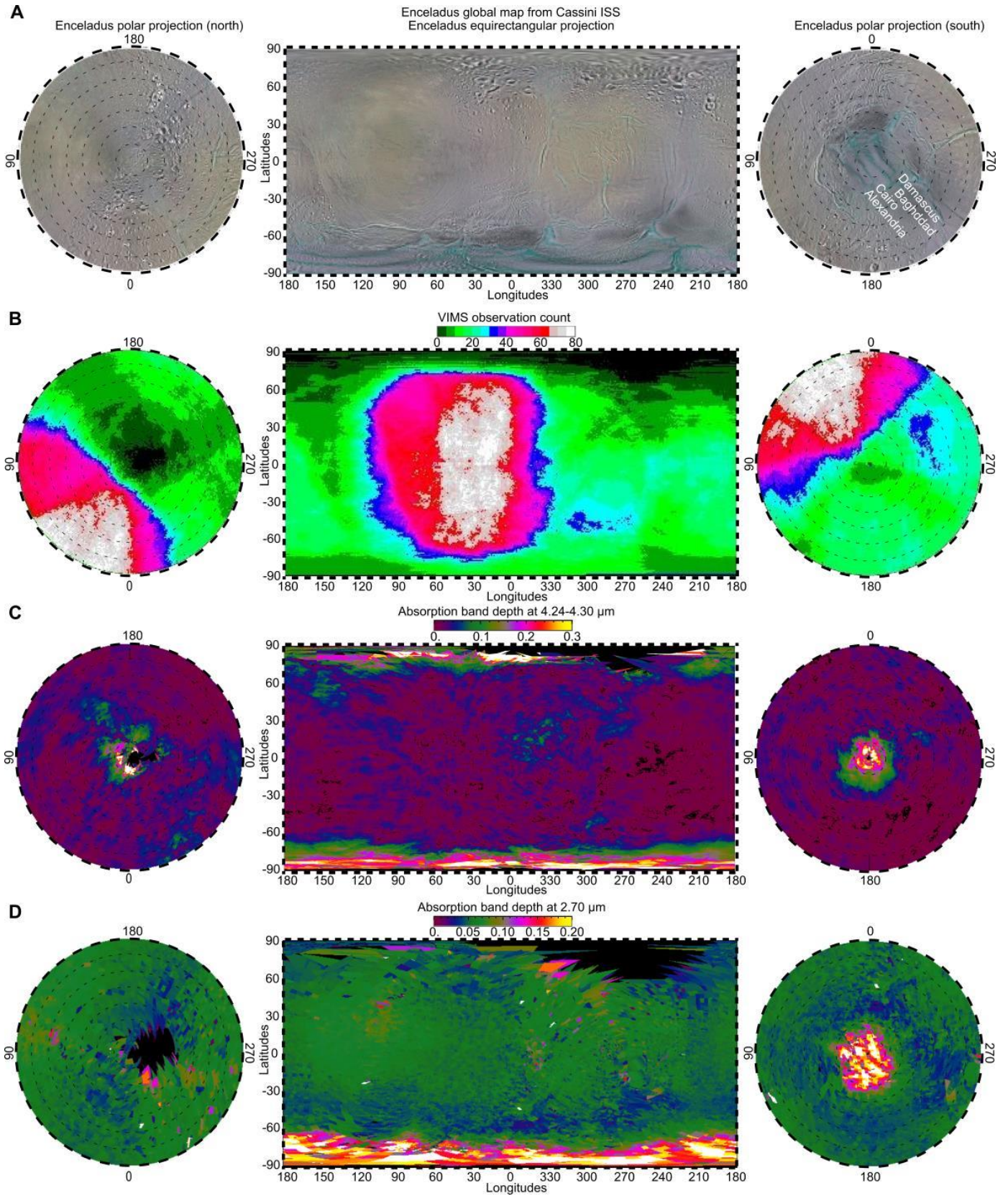
3.1 Deeper CO₂ absorption bands in the South Polar Region

3.1.1 Comparison of absorption band distributions at 2.70 μm and 4.24–4.30 μm

264 Fig. 6 shows maps of Enceladus from the *Cassini* mission in cylindrical and polar projections. Fig.
265 6A is a contextual color mosaic of ISS images in the visible. Fig. 6B shows the coverage of VIMS data
266 used in this analysis. In order to maximize the signal-to-noise ratio, multiple observations of an area
267 are combined. As a consequence, the maps of CO₂ absorption band depth are more reliable where the
268 number of observations is high. The highest redundancy of VIMS observation occurs on the area
269 between 110°W and 20°W and between -70° and +70° in latitude. The North Polar Region has the
270 smallest number of observations. In first order, mapping of CO₂-rich compounds is estimated by the
271 absorption band depth between 4.24 and 4.30 μm (Fig. 6C). Measurements of high values of the CO₂
272 absorption band depth that correspond to the most spatially-coherent area (no gap in the detection
273 and well-defined contours) occur near the South Pole. High values in the northern hemisphere around
274 60°N, 270°W, as well as around the North Pole between 70°N and 80°N correspond to a small (<20)
275 number of observations. At mid-latitudes (between 60°S and 60°N), low values (<0.1) cover a large
276 between 120°W and 300°W, whereas moderately low values (0.1-0.2) cover an equally large area on
277 the opposite hemisphere. In general, when detected values are low (<0.05), or result from a low
278 number of observations (<10), they have to be interpreted with caution.

279 For the 2.70 μm absorption (Fig. 6D), the highest values that cover a spatially-coherent area are
280 also observed in the South Polar Region, similar to the 4.24-4.30 μm absorption. The maximum value
281 is measured on the fault that forms the base of the Tiger Stripes and connects them perpendicularly
282 on the upper half of the polar view. High values are also measured in the northern hemisphere between
283 270°W and 330°W, but not around 240°W, which is spectrally different from the 4.24-4.28 μm
284 absorption.

285 The fact that the distribution of absorption bands at 2.70 μm and at 4.24-4.30 μm present the most
286 obvious similarities over the South Polar Region is an indication that they are both sensitive to the
287 same surface component (CO₂). Because the 2.70 μm absorption in VIMS spectra appears only as an
288 inflection in a range of wavelengths that is sensitive to broad and strong absorption processes, the
289 observed variations of the 2.70 μm absorption band depth cannot be interpreted alone for quantifying
290 CO₂.

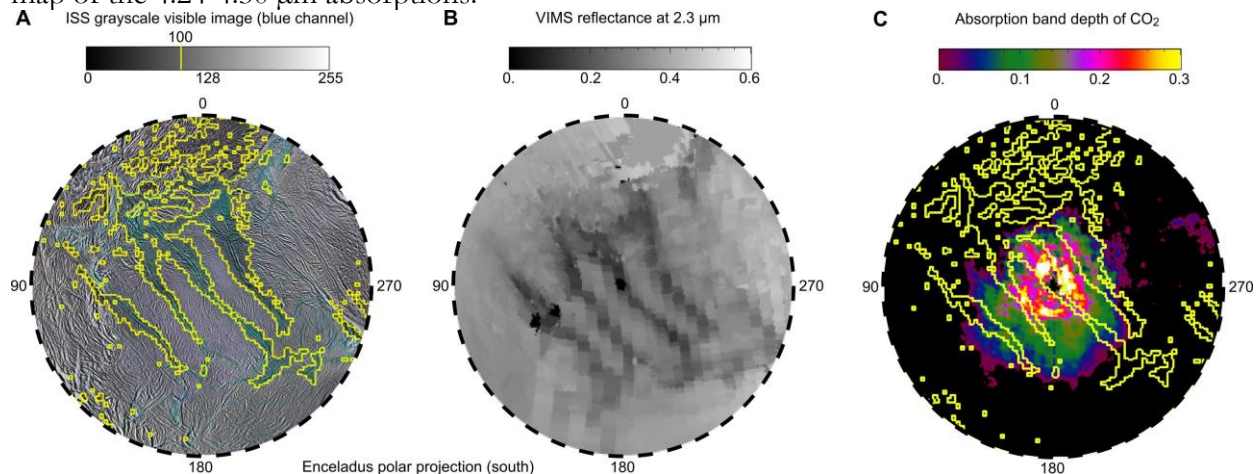


292 **Fig. 6: Enceladus global and polar maps calculated with data from the entire Cassini mission. The**
 293 **leading hemisphere (centered on 90°W) is on the left side of each map, whereas the trailing**
 294 **hemisphere (centered on 270 °W) is on the right. The polar projections extend from the pole to 20°**
 295 **latitude. (A) ISS color mosaic. The four major faults of the South Polar Region are labeled. (B) Count**
 296 **of VIMS observations of the surface of Enceladus with spatial resolution better than or equal to 5**
 297 **km/pixel for the entire *Cassini* mission up to March 2016. Most areas have been observed multiple**
 298 **times, except the North Polar Region. (C) Combined CO₂ absorptions between 4.24 and 4.30 μm. It is**
 299 **an intermediate step in the making of the CO₂ distribution maps shown in Fig. 7. The median value**
 300 **of all VIMS observations is displayed where data overlap. (D) Inflection at 2.70 μm in VIMS spectra,**
 301 **calculated as an absorption band depth. The highest resolution observations are displayed on top of**
 302 **the map.**

303 3.1.2 Combined analysis of the absorption bands at 2.70 μm and 4.24-4.30 μm

304 In this part of the study, the 2.70 μm absorption band depth is a conditional criterion for the
 305 detection of any form of surficial CO₂. The rationale behind this assumption is that the 2.70 μm
 306 absorption band should always exist in a CO₂-rich material. In particular, on the South Polar region,
 307 where several studies have confirmed CO₂ ice deposits (Brown et al., 2006; Palmer and Brown, 2008;
 308 Brown et al., 2011) both absorption bands are strong. According to laboratory experiments, although
 309 the stretching mode of the CO₂ molecule produces a much stronger absorption band at 4.24-4.30 μm
 310 than the vibration transition mode at 2.70 μm, the 2.70 μm absorption band exists regardless of the
 311 physical structure of CO₂, whether it is crystalline CO₂ ice (Isokoski et al., 2013), complexed CO₂
 312 (Bernstein et al., 2005) or CO₂ clathrate hydrate (Oancea et al., 2012). Furthermore the absorption
 313 feature is theoretically observable at VIMS spectral resolution (Oancea et al., 2012). We did not find
 314 measurements or documentation for amorphous CO₂ ice (Escribano et al., 2013 provide only spectra
 315 for the stretching mode). In addition, minerals that exhibits an absorption band near 4.25 μm such as
 316 beryl and cordierite (McCord et al., 1998; Clark et al., 2003) have also an absorption feature near 2.70
 317 μm.

318 The combined detection of an absorption between 4.24 and 4.30 μm and an inflection at 2.70 μm
 319 is mapped in Fig. 7. In this map, the absence of a 2.70 μm absorption band is used as a mask on the
 320 map of the 4.24-4.30 μm absorptions.



321 **Fig. 7: Maps of Enceladus obtained from *Cassini* observations in southern polar projection between**
 322 **50°S and 90°S. (A) Surface albedo as observed by ISS in the visible. Yellow contour lines indicate**
 323 **contrast in brightness in the ISS blue channel between the Tiger Stripes (blueish) and the surrounding**
 324 **terrains (gray). (B) Surface reflectance as observed by VIMS at 2.3 μm. The highest resolution**
 325 **observations are displayed on top of the map. (C) Consolidated map of CO₂ absorption bands on**
 326 **Enceladus. The median value of all VIMS observations is displayed where data overlap. The presence**
 327 **of CO₂ is indicated by the color scale.**

328 of CO₂ is confirmed only where both 2.70 μm absorption band depth and 4.24–4.30 μm absorption band
329 depth are greater than 0.07. The other regions of Enceladus are not shown here because the maps
330 would be entirely black.

331 **3.2 How to reconcile results from absorption features at 4.24-4.30 μm and** 332 **2.70 μm in VIMS spectra of Enceladus**

333 In section 3.1, maps of the CO₂ absorption band depth at 4.24-4.30 μm and 2.70 μm indicate
334 different distributions across the surface of Enceladus. Although the inflexion at 2.70 μm is detected
335 in most VIMS spectra, the South Polar Region above 70°N is the only location where it reaches a
336 maximum (>0.12). In contrast, the absorption band at 4.24-4.30 μm has several maxima in the
337 northern hemisphere, above 50°N and in the South Polar Region below 70°S. At this point of the
338 analysis, the question is whether the two absorption features indicate contradictory results or if they
339 complement each other.

340 **3.2.1 Arguments in favor of interpreting any absorption feature at 4.24-4.30 μm as** 341 **evidence for CO₂**

- 342 - In laboratory spectra of CO₂-rich materials, the transition vibration absorption of CO₂ at 2.70
343 μm is always weak and not spectrally resolved by VIMS.
- 344 - Although the absorption band depth at 4.24-4.30 μm and 2.70 μm are correlated in the South
345 Polar Region of Enceladus, many VIMS spectra of the northern hemisphere exhibit only a
346 strong absorption feature at 4.24-4.30 μm and a weak one at 2.70 μm, which indicates that
347 the two types of absorption are not always correlated.
- 348 - CO₂ is known to be present at the surface and in the plumes that result from eruptions on
349 the active faults of the South Polar Region.

350 **3.2.2 Arguments in favor of a detection of CO₂ only when both absorption features at** 351 **4.24-4.30 μm and 2.70 μm are present**

- 352 - All laboratory measurements of any form of CO₂ (crystalline ice, complexed and clathrate
353 hydrates) and of CO₂ inclusions in minerals (McCord et al., 1998; Clark et al., 2003) such as
354 beryl (Wood and Nassau, 1967) and cordierite (Farrell and Newnham, 1967) confirm the
355 presence of the two absorption bands.
- 356 - From laboratory experiments, the presence of a strong 4.24-4.30 μm CO₂ absorption band
357 means that a strong 2.70 micron band is systematically present.
- 358 - A few minerals have an absorption band 4.24-4.30 μm that is not related to CO₂ such as
359 hydroxyl-apatite and hydroxyl-goethite (McCord et al., 1998; Clark et al., 2003), thus any other
360 absorption feature that is known to be present in CO₂-rich materials, such as the 2.70 μm
361 absorption band, should be used as an additional detection criterion.

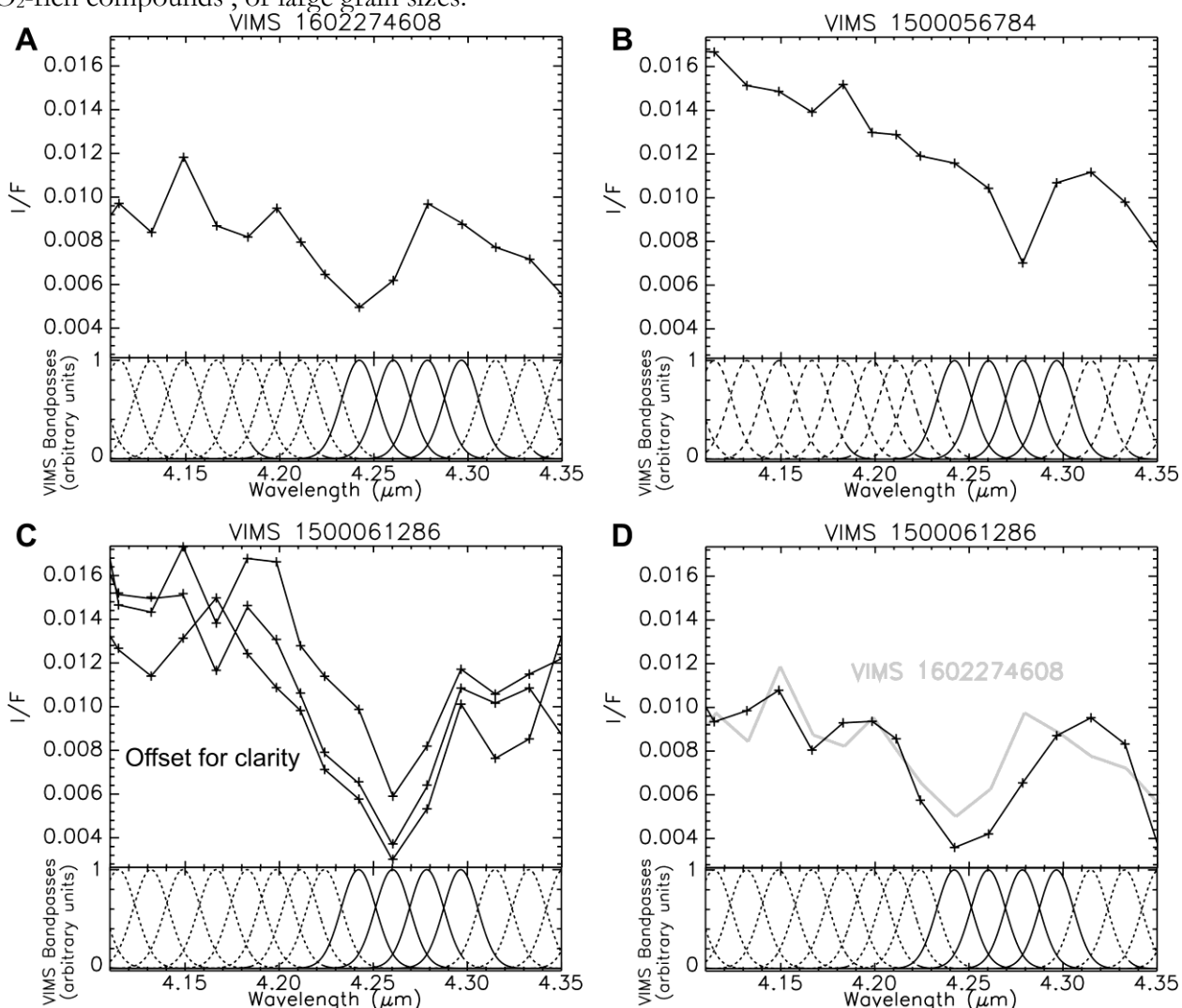
362 **3.2.3 Synthesis: absorption features at 4.24-4.30 μm and 2.70 μm may be sensitive to** 363 **different forms of CO₂**

364 The two subsections above show that ignoring the 2.70 μm absorption (subsection 3.2.1) or
365 combining absorption features at 4.24-4.30 μm and 2.70 (subsection 3.2.2) both result in contradictions
366 that are difficult to justify about CO₂ spectra. In order to reconcile the two results, we adopt the
367 hypothesis that the two CO₂ absorption features do not carry the same type of information, resulting
368 in non-correlated variations of the absorption band depths. This assumption implies that the inflexion
369 at 2.70 μm cannot be ignored, but that it should not be used as the sole detection criterion for CO₂ or
370 as a parameter for CO₂ abundance estimates. Subsequently, Fig. 7 may just illustrate the distribution
371 of one form of CO₂ compound, which is characterized by strong absorptions at 4.24-4.30 μm and at

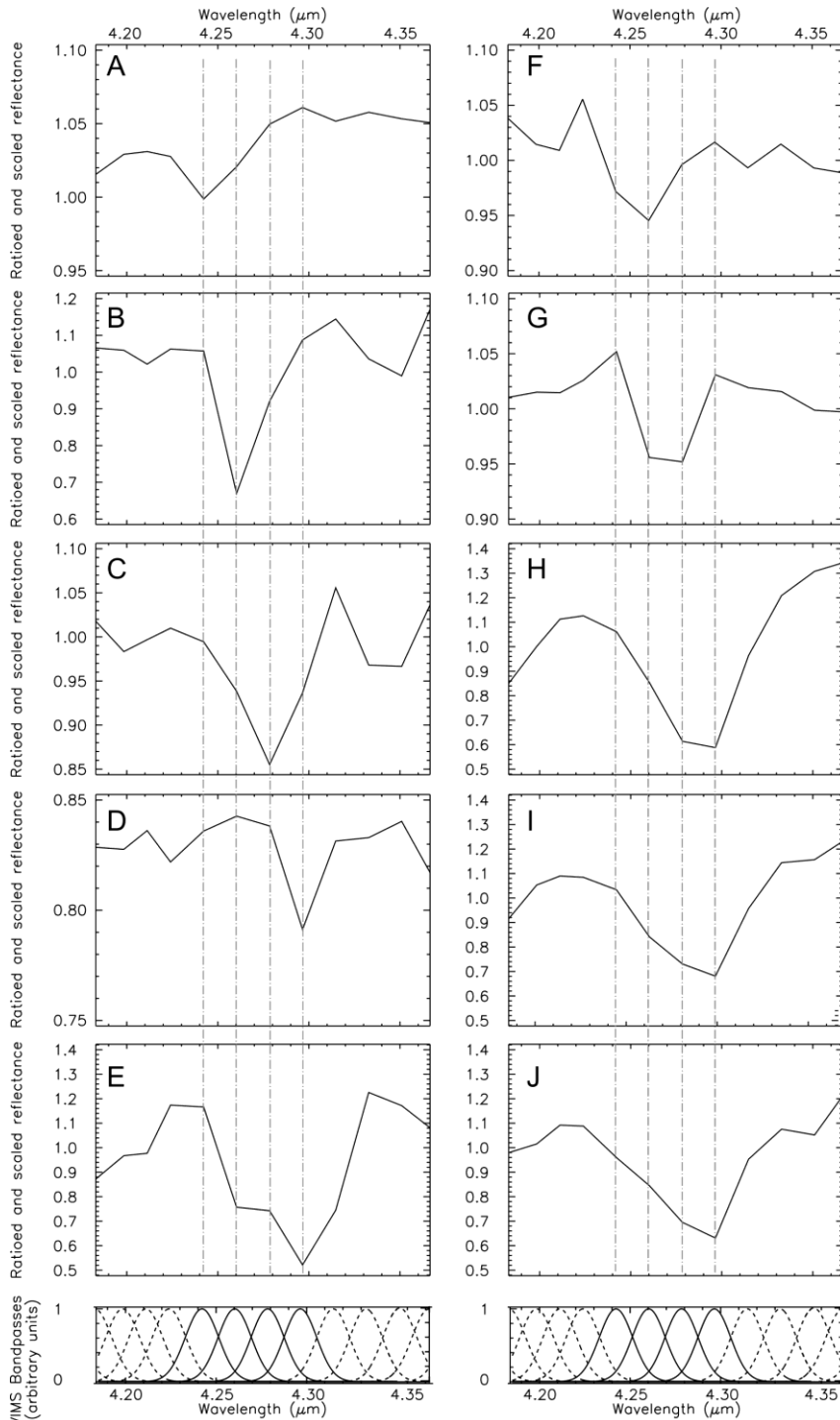
372 2.70 μm , and that is may not be representative of other forms of CO₂ compounds. The analysis of
 373 absorption band position may help clarify this assumption and determine whether the distribution of
 374 various types of CO₂ is consistent with the differences observed between the absorption band depths
 375 at 4.24-4.30 μm and 2.70 μm .

3.3 Variations of the CO₂ absorption band shape and position

376 One objective of this study is to discriminate between various types of CO₂ compounds (Fig. 8 and
 377 Fig. 9) by analyzing the absorption band depth measured at the expected position for free, amorphous
 378 CO₂ ice (4.28 μm , Fig. 8A and Fig. 9C), crystalline CO₂ ice (4.26 μm , Fig. 8B and Fig. 9B), CO₂
 379 clathrate hydrates (and Fig. 9D) and complexed CO₂ (4.24 μm , Fig. 8C and Fig. 9A). A broad band
 380 that covers more than one VIMS channel (Fig. 8D and Fig. 9E-J) represents either a mixture of various
 381 CO₂-rich compounds, or large grain sizes.
 382



383
 384 **Fig. 8: Example of VIMS spectra showing CO₂ absorption bands of various shapes and positions,**
 385 **as expected for deposits of complexed CO₂ (A), free CO₂ ice (B), and combinations of the two (C and**
 386 **D). The three solid line Gaussian curves correspond to the VIMS bandpasses that are used to measure**
 387 **the CO₂ absorptions. The number above each plot indicate the VIMS file.**

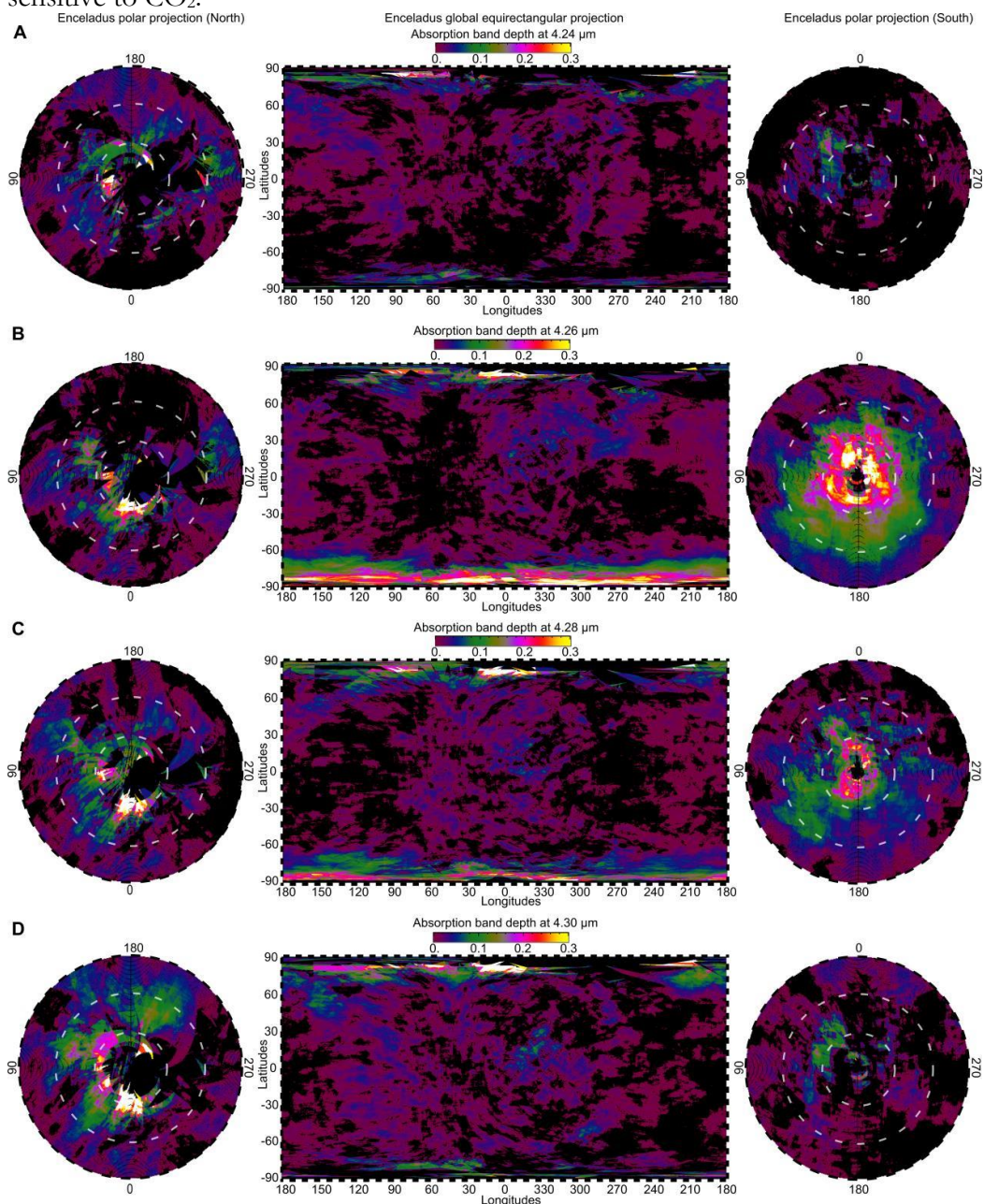


388

389 **Fig. 9** Example of VIMS spectra that are averaged over multiple observations, scaled and ratioed
 390 (for details, see text in section 2.2.1) in order to enhance the CO₂ absorption bands. Uncertainties are
 391 estimated to be on the order of ± 0.01 . Various shapes and positions of the CO₂ absorption bands are
 392 observed, as expected for deposits of complexed CO₂ (A), free CO₂ ice (B), possible amorphous ice
 393 (C) and possible CO₂ clathrate hydrates (D). Combinations of two or more types of CO₂ components
 394 may create absorption features with more than one minimum (E). Several types of CO₂ components,
 395 or rain size variations or temperature variations may create broad bands and different wavelength
 396 position of the band center (F-J) The Gaussian curves correspond to the VIMS bandpasses.

397 **3.3.1 Comparative absorption band depth distributions at 4.24, 4.26, 4.28 and 4.30 μm**

398 Although laboratory experiments show no evidence that the stretching mode absorption band of
 399 the CO₂ molecule at 4.24-4.30 μm could exist without the vibration transition mode at 2.70 μm , a large
 400 number of VIMS spectra of Enceladus show a strong absorption at 4.24-4.30 μm (>0.3) and a weak
 401 (<0.07) absorption feature at 2.70 μm . In Fig. 10A-D, we present the global distributions of absorption
 402 band depths measured at four consecutive VIMS wavelength channels (in increasing order), which are
 403 all sensitive to CO₂.



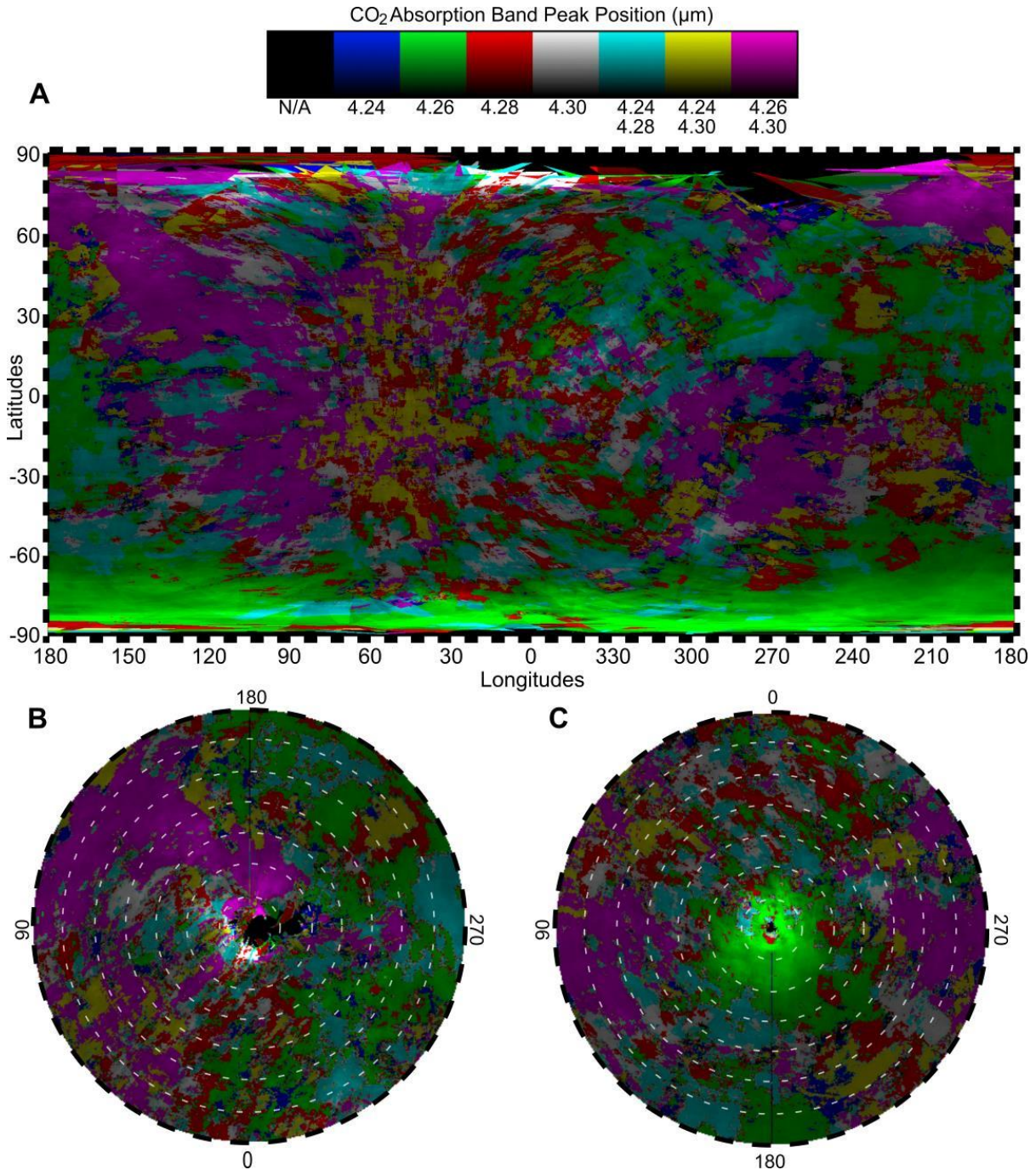
404
 405 **Fig. 10: Maps of the CO₂ absorption band depth on the South Polar Region of Enceladus from 60°S**
 406 **to 90°S. The median value of all VIMS observations is displayed where data overlap. From top to bottom, the**
 407 **spectral parameters represent absorptions at 4.24 μm (A), 4.26 μm (B), 4.28 μm (C), and 4.30 μm (D)**
 408 **respectively.**

409 Over the South Polar Region (60°S-90°S), the absorption band measured at 4.26 μm (Fig. 10B) is
410 clearly the strongest of all. At this wavelength, high values (>0.2) represent a larger area than the band
411 depth measured at other wavelengths. In addition, these high absorption band depth values at 4.26
412 μm occur mostly, between 70°S and the South Pole, whereas the absorption band depth measured at
413 other wavelength have mostly moderate values (<0.1) in that region. In all four maps of absorption
414 band depths, several areas with values ~ 0.3 exist in the North Polar Region around 80°N between 0°
415 and 90°W.

416 3.3.2 Position of the asymmetric stretching mode absorption maxima of the CO₂ 417 molecule

418 In order to determine the nature of the CO₂ detected on the surface, we performed an analysis of
419 the absorption band position of the CO₂ molecule stretching mode. As shown in Fig. 8, this absorption
420 band may exhibit one or two absorption maxima, which likely indicates a possible mixture of CO₂-
421 rich compounds rather than a homogenous composition. Ideally, the analysis of the absorption band
422 position would be performed by fitting one or more Gaussian curves to the CO₂ absorption feature
423 as a way to achieve accurate measurements. However, since the number of VIMS wavelength channels
424 between 4.24 and 4.30 μm is only four, the fitting of two or more Gaussian curves is an under-
425 determined problem (with more unknowns than equations). As a consequence, the absorption band
426 analysis is limited to measurements of absorption band depths, position of the absorption maximum
427 (or maxima when two peaks are detected), and absorption band area.

428 Fig. 11 combines the information of the multiple peak position with the absorption band area.
429 Peaks at all wavelengths between 4.24 and 4.30 μm are observed globally on Enceladus, and all
430 combinations of two peaks (4.24 μm and 4.28 μm , or 4.24 μm and 4.30 μm , or 4.26 μm and 4.30 μm)
431 are also observed. Most of Enceladus' South Polar Region between 70°S and 90°S is characterized by
432 an absorption band with a maximum centered at 4.26 μm (green) that is characteristic of crystalline
433 CO₂ ice. The North Polar Region exhibits several homogenous large areas near 80°N with strong,
434 absorptions at 4.30 μm (white) that may indicate CO₂ clathrate hydrates. Overall, the occurrence of
435 maximal absorptions at 4.26 μm and 4.30 μm correspond to the largest spatially-coherent zones on
436 Enceladus. In the North Polar Region, few, small areas show a strong double-peak absorption at 4.26
437 and 4.30 μm (magenta) which may indicate crystalline CO₂ ice and CO₂ clathrate hydrates, and at 4.24
438 and 4.28 μm (cyan), which may indicate complex CO₂ and possibly CO₂ clathrate hydrates or CO₂ ice.
439 At latitudes below 70° in both hemisphere, the values of the absorption band depth are <0.07 , which
440 indicate a low probability of detection.



441
 442 **Fig. 11: Wavelength position of the CO₂ absorption local maxima between 4.24 and 4.30 μm. The**
 443 **hue illustrates the position, and the brightness is sensitive to the absorption band area (brighter means**
 444 **a larger area). (A) Global map in equirectangular projection. (B) Polar map between 20°N and 90°N.**
 445 **(C) Polar map between 20°S and 90°S.**

446 3.3.3 Sensitivity of absorption band position to the type of CO₂

447 In the spectra of many minerals and ices (e.g. H₂O), the position of absorption bands shifts as a
 448 function of temperature. On Enceladus, variations of >100 K in the surface temperature are inferred
 449 from *Cassini* Composite Infrared Spectrometer (CIRS) measurements on the Tiger Stripes and
 450 between them [Abramov and Spencer, 2009; Goguen et al., 2013]. In VIMS spectra of Enceladus,

451 there is no apparent correlation between the high temperature of the main active faults and the
452 position of the CO₂ absorption bands between 4.24 and 4.30 μm. As a result, the wavelength shift
453 observed on Enceladus must be caused by mixture in a chemical compound [Mastrapa et al., 2005].

454 **3.3.4 Result summary: Nature and distribution of CO₂ on Enceladus**

455 Although most of Enceladus surface exhibits spectral features at 2.7 μm and in the range 4.24-4.30
456 μm, the relative strengths of these two absorption bands may be sensitive to the different states of
457 CO₂ exposed. According to laboratory measurements, measurements of the absorption band position
458 between 4.24 and 4.30 μm help determine whether CO₂ samples are predominantly free, complexed
459 (Chaban et al., 2007) or present as CO₂ clathrate hydrates (Oancea et al., 2012). In our maps of
460 Enceladus, we observe variations in absorption band position, similar to observations on Callisto
461 (McCord et al., 1997, 1998) or Ganymede, that are consistent with the presence of CO₂ ice and
462 complexed CO₂, and possible CO₂ clathrate hydrates. Mixture of CO₂ ice, complexed CO₂, and CO₂
463 clathrate hydrates likely explains the observed lateral variations of the CO₂ absorption band width on
464 Enceladus. Even though the grain size distribution and surface temperatures may also affect the CO₂
465 absorption band width in some extent, the effects are not as strong, and thus less likely to be measured
466 by VIMS. As illustrated in Fig. 7D, the combination of absorption bands of crystalline CO₂ ice,
467 amorphous CO₂ ice, complexed CO₂ and CO₂ clathrate hydrates may produce a single, wide
468 absorption. On first order, from our analysis, four main types of compositional units can be identified
469 globally:

- 470 - A strong absorption band at 2.7 μm is found exclusively on the South Polar Region delimited
471 by the Tiger Stripes, and it is associated with a strong absorption band centered at 4.26 μm,
472 characteristic of crystalline CO₂ ice. The deepest absorption bands at 4.26 μm occur between
473 the main active faults, although the correlation is not very clear.
- 474 - A weak or moderate absorption band at 2.7 μm and a strong absorption which may occur
475 anywhere in the range 4.26-4.30 μm may be due to CO₂ clathrate hydrates, and occur on the
476 North Polar Region only.
- 477 - Moderate absorption bands at 2.7 μm and in the range 4.24-4.30 μm are measured at mid-
478 latitudes (60°S-60°N) on the anti-Saturn side (300°W-120°W). The high variability in the
479 position of the absorption stretching mode of the CO₂ molecule in that region suggest
480 mixtures of different CO₂ compounds on the surface.
- 481 - Weak absorption bands at 2.7 and 4.26 μm are measured at mid-latitudes (60°S-60°N) on the
482 sub-Saturn side (120°W-300°W). Although the absorption band depths are close to the noise
483 level in the spectra, they may be sensitive to small amounts of crystalline CO₂ ice on the
484 surface.

485 **4. Interpretation: sources and transport of CO₂ on Enceladus**

486 The presence and distribution of one or several forms of CO₂ is relevant to the interpretation of
487 geophysical processes that occur on Enceladus and the possible interactions with the specific
488 environment created by the vicinity of Saturn, such as the magnetic field. Maps of the three absorption
489 bands of CO₂ on Enceladus support two main results.

490 First, the distribution of CO₂ can help identify its origin. In order to interpret the origin of surface
491 materials on Enceladus, three hypotheses may be considered: endogenic (i.e. release by venting
492 through the crust), exogenic (i.e. delivery by collision with a CO₂-rich object), and in-situ (i.e.
493 formation by chemical reaction or parent molecules and particles).

494 Second, the different forms of CO₂ detected on the surface may have implications for the
495 formation, transport and deposition mechanisms for this constituent. As a follow-up of section 3.2.3,

496 we observed that where the 2.7 μm band is mostly correlated with the 4.26 μm band, which is
497 consistent with a higher abundance of free CO₂ ice. In locations where the 4.24-4.30 μm band is
498 present, but the 2.7 μm band is weak or absent, other CO₂ compounds are more abundant. In
499 particular, the fact that the 4.30 μm sensitive to CO₂clathrate hydrates is strong on the northern polar
500 region is consistent with a more stable storage of CO₂ at Enceladus northern polar latitudes than on
501 the South Polar Region. As a corollary, the strong 4.26 μm band around the south pole indicates
502 greater recent exposure of the less stable free CO₂ ice at Enceladus, along the Tiger Stripes.

503 **4.1 CO₂ ice on the South Polar Region could be from local sources**

504 Absorption bands of CO₂ measured by VIMS are stronger in a zone that includes the main active
505 faults. Both higher abundances and larger grain sizes may produce deeper absorptions, whereas larger
506 grains may also cause wider absorption bands. With CO₂, there is no known way to evaluate the
507 relative role of abundance vs. grain size from near-infrared absorption bands alone because the
508 absorption bands are too narrow for a reliable analysis of bandwidth (they are not spectrally resolved
509 by VIMS). In the following subsections, we assume that the observed absorption band depths are
510 mostly sensitive to the abundance of CO₂.

511 **4.1.1 Crystalline CO₂ ice on the main active faults region: In situ deposition of CO₂ from** 512 **gas pockets**

513 The strongest detection of exposed CO₂ on the surface of Enceladus is observed in the South
514 Polar Region, within the limits of the main active faults (the Tiger Stripes), which indicates that
515 tectonic activity has some control on the release of CO₂, and therefore suggesting a source at the
516 surface connected to an internal reservoir.

517 When endogenic materials reach the surface of a planet, the highest concentrations might be
518 expected, intuitively, to be found near the source, and with concentrations decreasing as function of
519 distance from the points of origin. The distribution observed in VIMS data is consistent with release
520 of CO₂ from gas pockets between the faults that host plume eruptions [Matson et al., 2018]. The
521 presence of CO₂ ice and possible CO₂ clathrate hydrates is compatible with the gas pocket model: gas
522 seeping may deliver pure, cold CO₂ at low velocity, whereas blow-out eruption of gas pockets may
523 vent both pure CO₂ (most of which would escape) and CO₂ clathrate hydrates.

524 ***CO₂ ice deposits may come from seeping of CO₂ gas through narrow fractures***

525 Pure CO₂ gas can be released by seeping through a narrow fissure that connects to a gas pocket
526 between the internal ocean and the H₂O ice crust, between the main active faults [Matson et al., 2018].
527 During its seep through the ice its temperature is reduced. By the time it exits, it is in the range of ~70
528 to ~119 K. This constrains the deposition of CO₂ adjacent to the nozzle of the fissure, and there is
529 relatively little transport of CO₂ away from these vents.

530 Plumes, on the other hand, release primarily H₂O and a fraction of a percent of CO₂ at speeds that
531 exceed the escape velocity for Enceladus. For these reasons, the detection of pure CO₂ ice strongly
532 supports the gas pocket model. Any deposit of amorphous CO₂ ice may crystallize eventually. On
533 surfaces where the temperature exceeds 120 K, CO₂ ice sublimates rapidly over geologically-relevant
534 timescales [James et al., 1992]; consequently, observation of CO₂ ice on Enceladus implies a recent
535 deposition, which is consistent with contemporary activity.

536 4.1.2 Deeper CO₂ absorption bands between the Tiger Stripes: Possible role of H₂O and 537 surface temperature

538 Because plume eruptions on the Tiger Stripes are one source of CO₂ (Waite et al., 2006) mixed with
539 H₂O (dominant) and other minor species, they may produce deposits where CO₂ is mixed as well. Gas
540 seeping through the ice lid (Matson et al., 2018) are sources of pure CO₂, distinct from the Tiger
541 Stripes. The concentration of CO₂ in the deposits from gas seeping is expected to decrease as a
542 function of the distance from their source. This may explain why the highest concentrations of CO₂
543 ice (Fig. 7) may not be located near the Tiger Stripes.

544 *Partial masking of surface CO₂ by more recent H₂O deposits*

545 Absorption bands of H₂O at 1.5 and 2 μm are ~25% stronger on the Tiger Stripes (e.g. Jaumann
546 et al., 2008; Scipioni et al., 2017) than in between. In addition, H₂O absorption bands are very deep
547 everywhere on Enceladus (between 42 and 75% of the signal at 1.5 μm, according to Jaumann et al.,
548 2008), including where CO₂ is the most abundant (Brown et al., 2006). Changes in the depth and width
549 of the H₂O absorption bands, as well as the strength of the Fresnel peak of H₂O ice at 3.1 microns
550 are interpreted as variations of grain sizes (Jaumann et al., 2008; Taffin et al., 2012; Scipioni et al.,
551 2017) and surface temperature (Taffin et al., 2012). In all these studies, the hypothesis for possible
552 variation in H₂O abundance across the South Polar Region is never considered. However, assuming
553 the distribution of the H₂O absorption band depth is representative of the abundance of H₂O at the
554 surface, its anti-correlation with CO₂ band depth favors the hypothesis of independent events,
555 implying that the CO₂-rich units are not likely plume fallout.

556 H₂O-rich plume eruptions on Enceladus have been observed multiple times by *Cassini* (Hansen et
557 al., 2006); CO₂ has been detected also in smaller proportions (Waite et al., 2006). To date, however,
558 eruptions or releases of pure CO₂ have not been observed. As a consequence, if H₂O-rich eruptions
559 are more frequent or have more discharge than CO₂-rich venting, H₂O deposits on the surface of
560 active faults may partially mask deposits of CO₂.

561 *Local CO₂ abundance variations controlled by surface temperature heterogeneities*

562 Variations of the CO₂ absorption band depth over a given type of terrain (on or in between the
563 main active faults) can be also controlled by the local surface temperature. Although on average the
564 Tiger Stripes are warmer than the intermediate terrains, the internal convection in the ocean may create
565 temperature variations at the surface of the ocean [Stegman et al., 2009] that can be transmitted by
566 conduction at the surface of the faults or at the surface of the thick crust [Ingersoll and Pankine, 2010].
567 Furthermore, active convection may also erode the thickness of the ice lid or crust and allow more
568 heat to be conducted through the ice. As a consequence, surface temperature variations in the faults
569 or in between may lead to sublimation, transport and condensation of CO₂. This mechanism may
570 reinforce the anti-correlation of CO₂ abundance with the Tiger Stripes.

571 From *Cassini* CIRS observations acquired on July 14, 2005 at ~25 km/pixel, the temperature over
572 the average polar region southern of 65°S was ~133 ±12 K whereas the average over the Tiger Stripes
573 was 145 ±14K [Spencer et al. 2006, Abramov and Spencer, 2009]. Bright 16 - 200 μm emission at the
574 anti-Saturn end of Damascus and Baghdad Sulci may indicate excess endogenic power, either from the
575 Tiger Stripes or the inter-stripe regions (Howett and Spencer, 2011). From the analysis of VIMS
576 spectra acquired during the same flyby at ~5 km/pixel, temperature variations of crystalline water ice
577 over the Tiger Stripes were at least 20 K [Taffin et al., 2012]: this was done by measuring the strength
578 and center position of the absorption band of crystalline H₂O ice at 1.65 μm (e.g. Grundy and Schmitt,
579 1998). However, the temperatures between them could not be estimated because water ice is
580 predominantly amorphous in these locations. Measurements from a 74-km altitude flyby by CIRS on
581 April 14, 2012 provided a spatial resolution of 38 m/pixel where the average temperature of 197 K
582 was estimated to come from a 9-m wide fracture (after deconvolution of the point spread function

583 with respect to the size of the fracture) [Goguen et al., 2013]. Locally, temperatures >200K have been
584 measured on Enceladus [Abramov and Spencer, 2009; Goguen et al., 2013].

585 Temperatures well above 140 K recorded on the active faults may indicate thermodynamic
586 conditions that favor the sublimation of CO₂ ice [McCord et al., 1998]: in fact, free CO₂ ice has a
587 vapor pressure of 1.8 mbar at 140 K and 31 mbar at 160 K [James et al., 1992], whereas for surface
588 temperatures below ~72 K, it may be stable over geological timescales [Watson et al., 1963]. Over
589 time, this difference in surface temperatures between the Tiger Stripes and intermediate terrains may
590 result in the observed pattern in the abundance of CO₂, possibly as the result of migration of free CO₂
591 ice towards areas with lower surface temperatures [Brown et al., 2006]. Thus the detection of CO₂ ice
592 indicates a transient deposit, and a mechanism of CO₂ replenishment therefore exists at the surface
593 [Brown et al., 2006], similar to Europa [McCord et al., 1998].

594 4.2 CO₂ compounds on the North Polar Region could come from eruptions 595 on the Tiger Stripes

596 4.2.1 CO₂ clathrate hydrate deposits may come from blow out eruptions

597 *Formation of CO₂ clathrate hydrates*

598 Fig. 10D and Fig. 11 show relatively deeper absorption band at 4.30 μm in the North Polar Region
599 that could be consistent with CO₂ clathrate hydrates. These hydrates could be formed in Enceladus'
600 internal ocean [Bouquet et al. 2015], especially structure I (46 H₂O molecules for 1 to 8 CO₂
601 molecules). The temperature and salinity of the internal ocean likely make CO₂ clathrate sink at the
602 bottom [Safi et al., 2017], and its dissociation may trigger plume eruptions (Kieffer and Xinli, 2006)
603 that are observed over the Tiger Stripes. Given the relatively small amounts of CO₂ measured in the
604 plumes compared to H₂O, the role of CO₂ clathrate hydrates in triggering plume eruptions (which
605 CO₂:H₂O ratio is 0.022 to 0.17) could be a valid assumption.

606 *Venting and deposition mechanisms of CO₂ clathrate hydrates*

607 When a CO₂ gas pocket evacuates (“blows”), fast degassing [Matson et al., 2018] may also entrain
608 some clathrate molecules present in the liquid, which may be vented out by the eruption and be
609 released onto the surface or above the surface. This process requires high velocities at the nozzle, large
610 openings of the ceiling of the gas pocket, and sudden exposure of the surface of the liquid, which is
611 at a temperature near 273 K (Matson et al., 2012), which comes from the constraints on temperature
612 in and out of the plume chambers (Schmidt et al., 2008). However, salt in the ocean in Enceladus with
613 concentrations of NaCl (0.05–0.2 mol.k⁻¹) and the Na-carbonates (0.02–0.1 mol.kg⁻¹) (Postberg et al.,
614 2009) lowers the temperature of the freezing point compared to pure water. Matson et al. (2012)
615 assumes a composition of seawater for Enceladus' internal ocean, which brings the freezing point in
616 Enceladus 2 K lower (271 K) than pure H₂O (273 K). According to (Matson et al., 2012) “in the plume
617 chambers, water vapor is produced which makes up some ninety percent of the gas in the plumes.
618 The heat required to evaporate the water comes from the ocean water passing through the plume
619 chambers. By giving up this heat the ocean water is cooled by 2 K. Since the near-surface ice–seawater
620 interface is at 271 K, this suggests that the ocean water arriving at the plume chambers has a
621 temperature of 273 K.” The high venting velocity may allow most of the CO₂ gas to escape. However,
622 a fraction of the population of the much heavier CO₂ clathrate hydrate molecules may not escape
623 from the gravity field of Enceladus, and thus it could be distributed over a large area. CO₂ clathrate
624 hydrates may survive during the eruption and after deposition at temperature near the ocean's surface
625 conditions (~273 K). According to the particle speed probability (Maxwell-Boltzman distribution) the
626 mean speed v_p for a structure I CO₂ clathrate hydrate is $76 < v_p < 88$ m/s (depending on whether one
627 or eight cages contain CO₂ molecules), and for structure II it is $49 < v_p < 52$ m/s. For both cases,

628 these speed estimates are well below the escape velocity of Enceladus (239 m/s). Therefore it might
629 be that blow out eruptions are the source of clathrate venting that might yield deposits. These
630 velocities, however, are high enough for deposition at great distances from the source, even as far as
631 the antipodes.

632 4.2.2 Stability of CO₂ clathrate hydrates

633 In low-gravity environment, the escape rate is proportional to the vapor pressure. According to
634 [Smythe, 1975], CO₂ clathrate compounds have vapor pressure several orders of magnitude lower than
635 CO₂ gas alone, which implies that deposits observed today at the surface may correspond to relatively
636 older materials than those consisting of CO₂ ice.

637 Assuming that CO₂ in the northern hemisphere come from remote sources, the relatively higher
638 concentrations of CO₂ clathrate hydrates in the North Polar Region may indicate that
639 thermodynamical conditions at 60°N latitude and further North favor the stability of that molecular
640 structure. Deposits of CO₂ clathrate hydrates from blow-out eruptions may occur on the entire surface
641 of Enceladus. Over time, the clathrate structures may dissociate faster at lower latitudes than in the
642 polar regions. Consequently, CO₂ clathrate hydrates may accumulate above 60°N. The same
643 phenomenon may occur in the Southern Polar Region, although the average surface temperature may
644 be higher than in the north, due to higher surface thermal flux in the vicinity of active faults. In
645 addition, new deposits of fresh CO₂ ice and H₂O ice may mask any older CO₂ clathrate hydrate
646 deposits.

647 5. Conclusions

648 We have related the distribution of CO₂ to the distribution of resurfacing activity of Enceladus by
649 mapping CO₂ surface deposits with near-infrared reflectance spectra from *Cassini* VIMS. The presence
650 of CO₂ ice indicates recent deposition because, at the temperatures measured on Enceladus, CO₂
651 sublimates rapidly, requiring active replenishment of surface deposits for this constituent to persist
652 over geologically-relevant timescales. Surface CO₂ is therefore compatible with endogenous origin
653 associated with the activity observed in the active faults, from which issue plumes. Our results confirm
654 the highest concentrations of CO₂ are measured in the South Polar Region, where most of the active
655 faults (e.g., the Tiger Stripes) are located, indicating that tectonic activity has some control on the
656 release of CO₂.

657 The results obtained from our analysis of VIMS spectra support the occurrence of seeping of CO₂
658 from gas pockets through narrow fissures. The release of pure, cold CO₂ forms ice on the surface near
659 the Tiger Stripes, adjacent to the source vent. Then, local high temperatures (> 130 K on average)
660 favor the sublimation of CO₂, whereas lower temperatures in between and outside the active faults
661 controls the transport and deposition of CO₂ ice. Partial overlay by H₂O ice coming from frequent
662 plume eruption may also enhance the anti-correlation of CO₂-rich materials with respect to the Tiger
663 Stripes and to the H₂O absorption band depths.

664 In our maps, we can discriminate between CO₂ ice, complexed CO₂ and CO₂ clathrate hydrates,
665 from absorption band depth measurements at four positions between 4.24 and 4.30 μm. The
666 distributions of the four absorption band depths of CO₂ present some obvious similarities at first
667 order, such as the highest concentrations on the South Polar Region, small areas in the North Polar
668 Region, and weaker absorption band depths at mid-latitudes. Those similarities are consistent with the
669 fact that all these absorptions are related to absorbers that share common chemical properties, such
670 as the type of molecule (here it is CO₂).

671 Some obvious differences exist in the location of maxima and the overall intensity of bands, which
672 is consistent with CO₂-rich materials that differ in chemistry (pure CO₂ ice versus complexed CO₂ or

673 CO₂ clathrate hydrates). Since complexed CO₂ and CO₂ clathrate hydrates are more stable on the
674 surface of Enceladus than free CO₂ ice, the distribution of the various types of CO₂ deposits is also
675 possibly a function of their relative ages.

676 Although plume eruptions have been observed by the *Cassini* mission, relatively large amounts of
677 H₂O (96-99%) with respect to CO₂ (0.3-0.8%) have been measured in the plumes (Waite et al., 2017).
678 The strong absorption band depth (up to 40% of the signal in our study) detected on the surface by
679 VIMS suggests a much higher proportion of CO₂ on the surface than in the atmosphere, which is not
680 consistent with the composition of an eruption from a plume chamber. On the other hand, VIMS
681 results are supported by the CO₂ gas pocket model (Matson et al., 2018), because its mechanism of
682 low velocity seep-venting is the only way of delivering pure, cold, CO₂ to the surface where it can
683 form CO₂ ice. In addition, this model is also compatible with more eruptive blow-out venting of gas
684 pockets that could produce and eject complexed CO₂ away from the South Polar Region. Plume
685 eruptions of H₂O and/or sudden eruption of CO₂ from a gas pocket may drag CO₂ clathrate hydrates
686 (that form in the ocean) and eject them over a very large area around the vent, possibly spreading
687 them over the entire surface of Enceladus. Over time, the lower surface temperatures of the polar
688 regions may favor the accumulation of CO₂ compounds with respect to the mid-latitudes.

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845

846 **Appendix**

847 Like most datasets collected by spacecraft instruments, VIMS data have imperfections that require
 848 more careful treatment as time passes and the analyses become more sensitive. In this study, data
 849 calibration has been under particular scrutiny, because precise positioning of each observed pixel is
 850 critical, and because the signal-to-noise in the range 4.24–4.28 μm is generally low. We have
 851 investigated several calibration issues and developed techniques for improving them in this case. These
 852 techniques could be generalized to be used by other studies and perhaps on the entire VIMS data set.

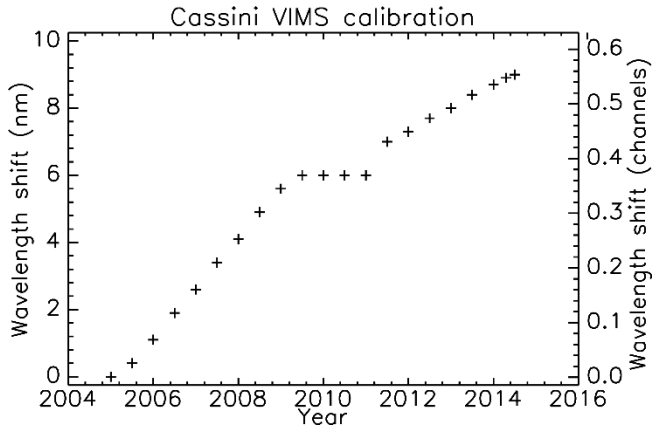
853 Of particular concern is how the VIMS infrared wavelengths have shifted over time (Fig. A1), as
 854 revealed by monitoring narrow lines of known position in Titan’s and Saturn’s atmospheres (Clark et
 855 al., 2016) (<http://wwwvims.lpl.arizona.edu/Howitworks.html>). We compensate for this variation by
 856 a two-step process. 1) A value for this wavelength shift is calculated at the time of the acquisition of
 857 any VIMS spectrum, assuming a linear interpolation between each point measured in Fig. A1. 2) VIMS
 858 spectra are resampled in the spectral dimension from the shifted wavelength to the original
 859 wavelengths.

860 **Table 1: Summary of VIMS observations processed to create global maps of Enceladus.**

Orbit number	First number	file	Last file	Number of observations
7	1484523993		1484533914	6
8	1487255888		1487302144	29
9	1489028159		1489087241	34
11	1495319119		1495363657	10
12	1500040056		1500069702	85
16	1511777815		1511784548	25
17	1514140705		1516171358	70
21	1530343833		1530346698	10
23	1536531398		1536534584	10
25	1541727375		1541728777	15
31	1561728480		1561728480	1
43	1597166869		1597180353	37
44	1600379685		1602294224	95
45	1604152328		1604184588	128
46	1606886709		1606891734	10
52	1627326410		1627330202	3
54	1634223108		1635855445	47
55	1637449785		1637482463	96
60	1652848938		1652874475	73
61	1656976036		1656976650	2
62	1660420386		1660446670	52
65	1669819228		1671608635	74
66	1675144005		1675152950	27
70	1694646956		1699270828	303
71	1702359426		1702362997	17
72	1710055594		1711575441	34

73	1713110172	1713115977	155
90	1816669056	1820438640	52
91	1823501373	1824733761	53
92	1829231340	1831438809	165
93	1834200478	1834214488	42

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862

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Fig. A1: VIMS wavelength shift as a function of time.

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5.1 A.1 VIMS calibration

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5.1.1 A.1.1 Dark current correction

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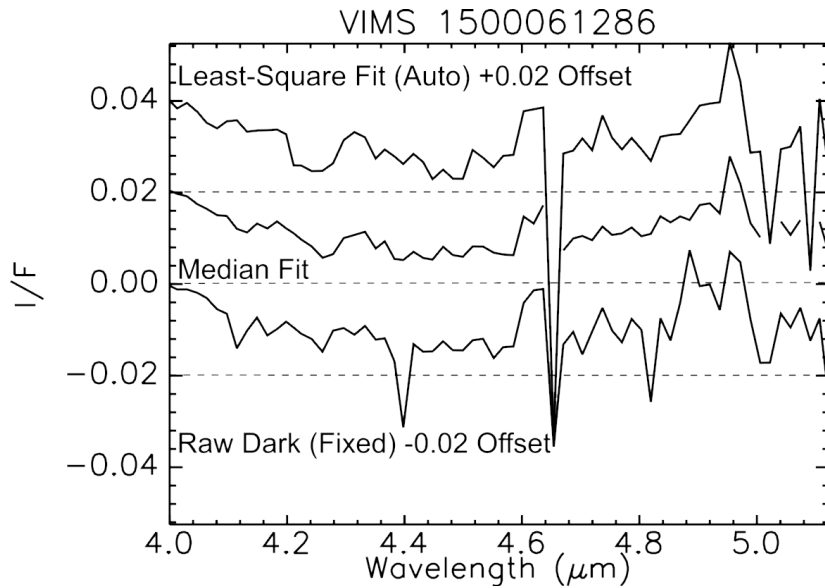
870

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873

In VIMS data, a measurement of the electronic dark current is acquired at the beginning of the sequence corresponding to one line of the final image cube. Reading and processing VIMS data can be done in a standard way using routines of the Integrated Software for Imagers and Spectrometers (ISIS, <https://isis.astrogeology.usgs.gov/>). The dark current correction for the VIMS infrared detector consists of reading the sideplane data and subtracting this from the input cube. In ISIS, the default correction from the VIMSCAL procedure is a simple subtraction from the raw DNs of the dark current values. Optionally, it is possible to apply the dark current with a linear fit instead of the default subtraction.

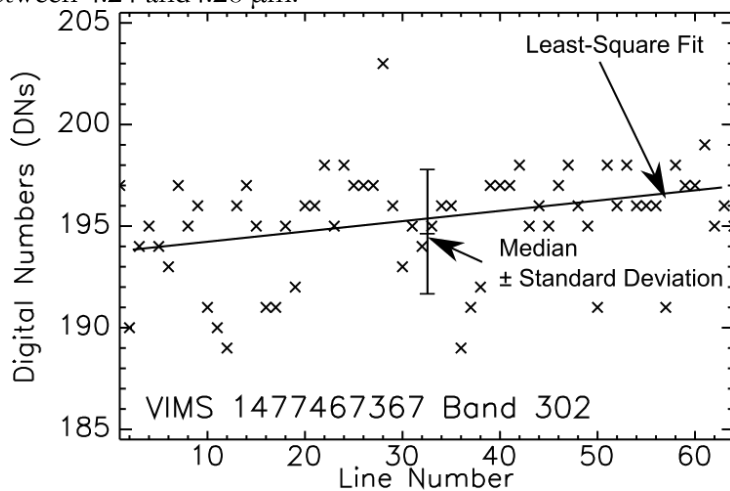


874

875 **Fig. A2:** : One example of a VIMS spectrum calibrated with three dark current corrections. In the
 876 **VIMSCAL** procedure of **ISIS**, the **Raw Dark** and **Linear Fit Dark** options may result in negative values
 877 (erroneous). The median is the one we use in this study.

878 We tested the two options and we have identified various occurrences of negative values in the
 879 data as a result (Fig. A2). Negative values in I/F are invalid, which proves errors in the calibration
 880 process. Analysis of these errors helped us understand the cause and develop a workaround (Fig. A3).

881 The VIMS dark current is affected by occasional spurious data that deviate from the expected trend
 882 between the first and last line of a given file (Fig. A3). These deviations are excessive compared to the
 883 measured DN's over the surface target. The simple subtraction of the measured dark current results
 884 in spikes in the spectra (Fig. A2) and horizontal striping of VIMS images (Fig. A4A). On the other
 885 hand, the optional linear fit of the VIMSCAL procedure is sensitive to outlying points caused by spikes
 886 in the data (Fig. A3), which may also results in excessive correction of the dark current. This
 887 overcorrection creates excessive absorption band depth (Fig. A4B) and sometimes negative I/F values
 888 between 4.24 and 4.28 μm .

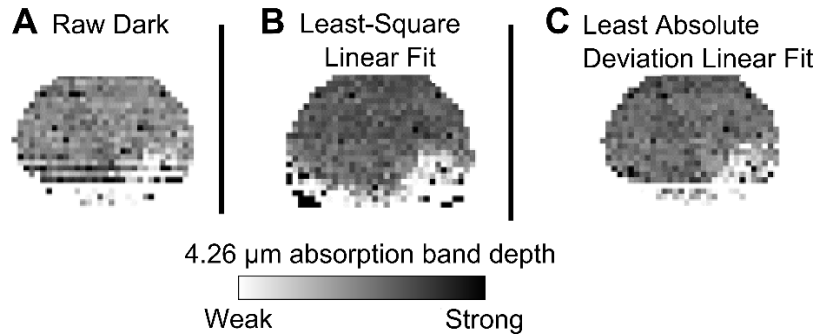


889

890 **Fig. A3:** VIMS Dark Current (DNs) as a function of line number and corrections. Calibration
 891 options publicly available either in the PDS or in **ISIS** include a subtraction of the measured dark
 892 current (default), or a least-square linear fit of the measured dark current as a function of the line

893 number. Both methods are affected by outlying points in the data, which create artifacts in the
 894 calibrated data. An improved correction proposed in this study relies on the median value of the dark
 895 current as a function of the line number.

896 Fig. 12:



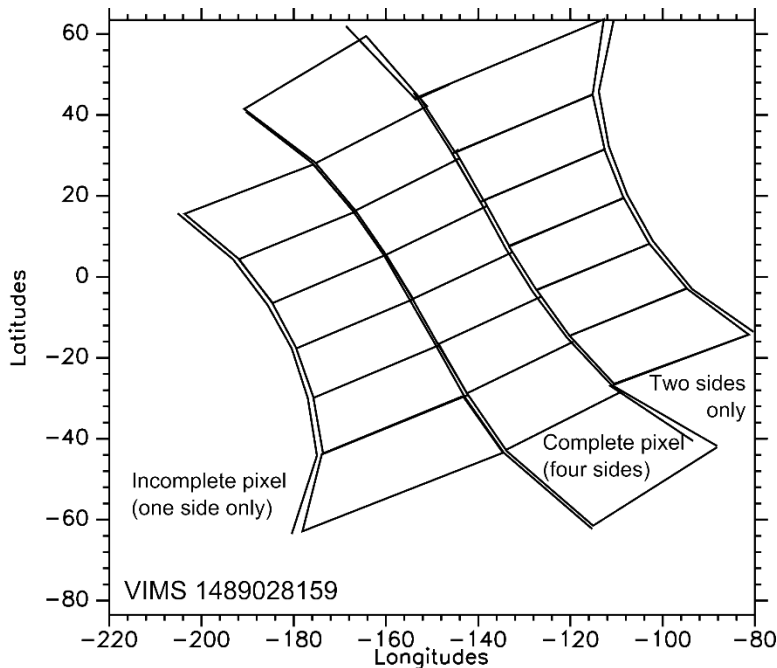
897

898 Fig. A4: VIMS images showing the effects of three different dark current corrections. A –
 899 Subtraction of the Raw Dark. B – Least-squares linear fit. C – Least absolute deviation linear fit (this
 900 study).

901 We have developed a dark current correction that relies on the median value of the measured
 902 points, which is less sensitive to outlying points due to spikes in the data (Fig. A3). As a result, negative
 903 I/F values are mostly eliminated (Fig. A2), and images have better spatial coherence (Fig. A4C).

904 5.1.2 A.1.2 Calculation of geographic coordinates

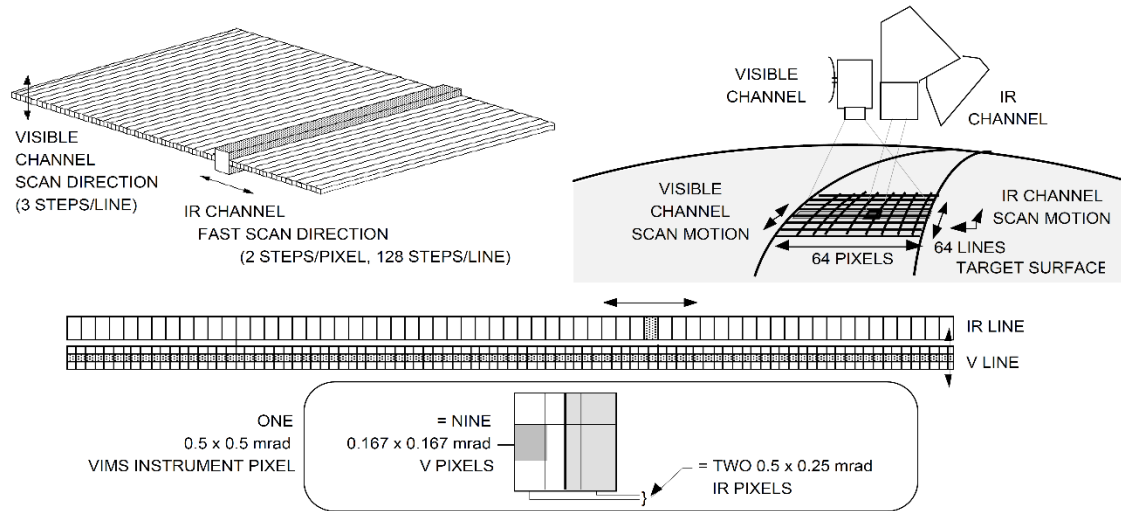
905 Although ISIS provides tools for mapping VIMS data, such as the PHOCUBE routine, the
 906 outcome is limited to one set of coordinates (pixel center latitude and longitude) for each pixel.
 907 However, each VIMS pixel can be large compared to the size of Enceladus, and the area covered by
 908 each pixel can vary a great deal in size and shape (Fig. A5). Thus, the location of the center of each
 909 pixel is not accurate enough to map the CO₂ distribution over linear features such as the Tiger Stripes.



910

911 **Fig. A5: Schematic view of the projection of VIMS pixels of one VIMS observation on Enceladus.**
 912 **It shows the geographic coordinates for each pixel is necessary for mapping accurately the surface of**
 913 **Enceladus.**

914 In order to improve the mapping of VIMS data, we calculated the coordinates of each corner of
 915 each pixel using data and routines from the Navigation and Ancillary Information Facility (NAIF)
 916 SPICE system (Acton, 1996). We used a package that is compatible with IDL (Interactive Data
 917 Language, by Harris Geospatial), which is available as a toolkit
 918 (https://naif.jpl.nasa.gov/naif/toolkit_IDL.html). The data needed to calculate the coordinates are
 919 the spacecraft ephemeris, planet ephemerides, instrument description and pointing which are also
 920 made publicly available by NAIF and have the generic name of SPICE kernels
 921 (<http://naif.jpl.nasa.gov/pub/naif/CASSINI/kernels/>). Based on the design of VIMS (Fig. A6)
 922 described in Brown et al. (2004) and in ([http://www.ifs-roma.inaf.it/cassini/downloads/VIMS-](http://www.ifs-roma.inaf.it/cassini/downloads/VIMS-paper.pdf)
 923 [paper.pdf](http://www.ifs-roma.inaf.it/cassini/downloads/VIMS-paper.pdf)), we have developed an IDL program that calls the routines of the NAIF toolkit, reads the
 924 SPICE kernels and calculates the geographic coordinates for each corner of each pixel, both at the
 925 start and stop times of the acquisition. In addition, the program calculates the incidence, emergence
 926 and phase angles and the geographic coordinates for the center of each pixel. Based on these geometric
 927 parameters, we have been able to validate our calculations against results provided by the PHOCUBE
 928 routine of ISIS.



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931

Fig. A6: VIMS pixel synthesis and image scanning (<http://www.ifsi-roma.inaf.it/cassini/downloads/VIMS-paper.pdf>).

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