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### Abstract

Laboratory spectral measurements of relevant analogues materials were performed in the framework of the Rosetta mission in order to explain the surface spectral properties of comet 67P. Fine powders of coal, iron sulphides, silicates and their mixtures were prepared and measured in the VIS/IR range. These spectra are compared to a reference spectrum of 67P nucleus obtained with the VIRTIS/Rosetta instrument up to 2.7  $\mu\text{m}$ , excluding the organics band centred at 3.2  $\mu\text{m}$ . The species used are known to be analogues for cometary materials which could be present at the surface of 67P. Grain sizes of the powders range from tens of nanometres to hundreds of micrometres. Some of the mixtures studied here actually reach the very low reflectance level observed by VIRTIS on 67P. The best match is provided by a mixture of sub-micron coal, pyrrhotite, and silicates. Grain sizes are in agreement with the sizes of the dust particles detected by the GIADA, MIDAS and COSIMA instruments on board Rosetta. The coal used in the experiment is responsible for the spectral slope in the visible and infrared ranges. Pyrrhotite, which is strongly absorbing, is responsible for the low albedo observed in the NIR. The darkest components dominate the spectra, especially within intimate mixtures. Depending on sample preparation, pyrrhotite can coat the coal and silicate aggregates. Such coating effects can affect the spectra as much as particle size. In contrast, silicates seem to play a minor role.

<b>Keywords</b>	Comets ; Comets, nucleus ; Comets, composition ; Spectroscopy ; Experimental techniques
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# Laboratory simulations of the Vis-NIR spectra of comet 67P using sub- $\mu\text{m}$ sized cosmochemical analogues

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## Abstract

Laboratory spectral measurements of relevant analogues materials were performed in the framework of the Rosetta mission in order to explain the surface spectral properties of comet 67P. Fine powders of coal, iron sulphides, silicates and their mixtures were prepared and measured in the VIS/IR range. These spectra are compared to a reference spectrum of 67P nucleus obtained with the VIRTIS/Rosetta instrument up to 2.7  $\mu\text{m}$ , excluding the organics band centred at 3.2  $\mu\text{m}$ . The species used are known to be analogues for cometary materials which could be present at the surface of 67P. Grain sizes of the powders range from tens of nanometres to hundreds of micrometres. Some of the mixtures studied here actually reach the very low reflectance level observed by VIRTIS on 67P. The best match is provided by a mixture of sub-micron coal, pyrrhotite, and silicates. Grain sizes are in agreement with the sizes of the dust particles detected by the GIADA, MIDAS and COSIMA instruments on board Rosetta. The coal used in the experiment is responsible for the spectral slope in the visible and infrared ranges. Pyrrhotite, which is strongly absorbing, is responsible for the low albedo observed in the NIR. The darkest components dominate the spectra, especially within intimate mixtures. Depending on sample preparation, pyrrhotite can coat the coal and silicate aggregates. Such coating effects can affect the spectra as much as particle size. In contrast, silicates seem to play a minor role.

## 1 Introduction

After spending about 2 years in the vicinity of comet 67P, the Rosetta spacecraft has provided unprecedented amounts of information on the formation of comets and the surface processes at play during a revolution. Among the experiments aboard the Rosetta spacecraft, the Visible and Infrared Thermal Imaging Spectrometer (VIRTIS, [Coradini \*et al.\* \(2007\)](#)) instrument was able to determine the spectral signature of cometary dust and its mixture with CO<sub>2</sub> and H<sub>2</sub>O ices. VIRTIS observations have revealed that the surface dust is generally homogeneous across the nucleus (with no obvious differences between the two lobes), is extremely dark (6% geometric albedo in the visible, [Ciarniello \*et al.\* \(2015\)](#); For-

33 *nasier et al. (2015)*, comprised in the 0.04-0.07 range of other comets, *Ciarniello*  
34 *et al. (2015)*), and that the surface material is spectrally red across the 0.3-2.7  
35  $\mu\text{m}$  range. While the presence of a 3.2  $\mu\text{m}$  feature suggests the presence of some  
36 organic component within the dust, significant efforts remain to be done to un-  
37 derstand the compositional controls on the spectral signature of 67P and more  
38 generally comets and related D-type asteroids (*Vernazza & Beck, 2016*).

39 Instruments onboard Rosetta have provided information on the physics of 67P  
40 dust, including dust size and velocity distribution. The Grain Impact Analyser  
41 and Dust Accumulator (GIADA, (*Colangeli et al., 2007; Della Corte et al., 2014*))  
42 measured sizes from 100  $\mu\text{m}$  to millimetres and characterised the dust flux (*Ro-*  
43 *tundi et al., 2015; Della Corte et al., 2016; Fulle et al., 2016*). The Cometary  
44 Secondary Ion Mass Analyser (COSIMA, *Henkel et al. (2003)*) measured sizes  
45 from tens of micrometres to millimetres (*Langevin et al., 2016; Merouane et al.,*  
46 *2016*). Last, the Micro-Imaging Dust Analysis System (MIDAS, *Riedler et al.*  
47 *(2007)*), which had the capability to observe grain-size at the smallest scale within  
48 the Rosetta payload, was able to image particles from nanometres to micrometres  
49 (*Bentley et al., 2016; Mannel et al., 2016*). The latter results reveal that the build-  
50 ing blocks of cometary dust are fine-grained, which was expected from thermal  
51 emission spectroscopy of cometary dust tails (*Kolokolova et al., 2004*) and coma  
52 (*Bockelée-Morvan et al., submitted*). Of importance, here is the fact that the ma-  
53 jor part the building blocks are sub- $\mu\text{m}$ , therefore of the order of the wavelength  
54 when observed in the Vis-NIR: a good cometary analogue requires the right grain  
55 size, in order to be in a realistic regime of radiative transfer.

56 The COSIMA instrument has provided some information on the bulk composition  
57 of cometary grains, that appear to be roughly chondritic (*Paquette et al., 2016;*  
58 *Hilchenbach et al., 2016*). However, there were no instruments onboard Rosetta  
59 able to directly measure the mineralogy of cometary dust. In this respect, measure-  
60 ments on micrometeorite and Interplanetary Dust Particles (IDPs) often described  
61 as originating from comets, are of paramount interest to understand their miner-  
62 alogy. However, note that a fraction of IDPs and micrometeorites could originate  
63 from asteroidal sources (*Vernazza et al., 2015; Battandier et al., 2016*).

64 The findings summarized above help in the selection of cometary analogue mate-  
65 rials, in terms of their particle size and composition. Here, by means of laboratory

66 measurements of the Vis-NIR reflectance properties of these materials and their  
67 mixtures, we aim to reproduce the spectral signature of 67P as measured by the  
68 VIRTIS instrument. In particular, our approach is to use sub- $\mu\text{m}$  powders, ana-  
69 logue to cometary dust, and to build on the vast knowledge derived on IDP from  
70 laboratory studies in order to define the constituents and composition of our ana-  
71 logue material.

## 72 **2 Methods**

### 73 **2.1 Compositional endmembers: rationale**

74 The composition of the refractory component of 67P was inferred from measure-  
75 ments of the VIRTIS and COSIMA instruments aboard Rosetta (Capaccioni *et al.*,  
76 2015; Quirico *et al.*, 2016; Fray *et al.*, 2016), from ground and satellite observa-  
77 tions (Crovisier *et al.*, 1997; Sugita *et al.*, 2005; Harker *et al.*, 2005; Lisse *et al.*,  
78 2006), and from the composition of STARDUST grains, chondritic porous strato-  
79 spheric IDPs and micrometeorites (Keller *et al.*, 2006; Zolensky *et al.*, 2006;  
80 Dobrica & Brearley, 2011; Dobrica *et al.*, 2011, 2012). The presence of mafic  
81 minerals (olivine, pyroxene) and glasses has been evidenced by spectroscopic ob-  
82 servations and laboratory analysis of cosmomaterials (Dobrica & Brearley, 2011;  
83 Dobrica *et al.*, 2012; Engrand *et al.*, 2016). Amorphous silicates is also deter-  
84 mined by the analysis in the laboratory of presumed cometary grains and the most  
85 primitive chondrites, which contain the so-called GEMS (Glass with Embedded  
86 Metals and Sulphides) (Leroux *et al.*, 2015). GEMS comprise a Mg-rich silicate  
87 glass phase that hosts very small inclusions of Fe metal and Fe sulphides. The  
88 contribution of these sub- $\mu\text{m}$  Fe-rich opaques to the spectral signature of 67P has  
89 been discussed in Quirico *et al.* (2016) and the presence of sulphides was sug-  
90 gested in the spectral modelling of VIRTIS data (Capaccioni *et al.*, 2015). Com-  
91 plex macromolecular organics have been observed in situ on Halley and 67P, and  
92 they are ubiquitous in presumed cometary grains as IDPs and micrometeorites  
93 (Fomenkova *et al.* (1994); Fray *et al.* (2016); and see discussion in Quirico *et al.*  
94 (2016)). This organic component is abundant in stratospheric IDPs (15 wt% in  
95 average, up to 90 wt%) and a significant fraction of organics in 67P dust is needed

96 to explain the dielectric properties of the cometary nucleus (Herique *et al.*, 2016).  
97 According to these results, three main groups of materials can be finally identified  
98 on cometary nuclei:

- 99 (i) anhydrous silicates such as Mg-rich olivines, pyroxenes and glasses;
- 100 (ii) opaque minerals as sulphides (pyrrhotite  $\text{Fe}_{1-x}\text{S}$  with  $0 < x < 0.20$ ) and Ni-  
101 bearing iron  $\text{Fe}^0$ ;
- 102 (iii) a refractory polyaromatic carbonaceous material.

103 Below, the analogue materials selected for each group are reported.

104 **Group (i):** a terrestrial silicate was used as analogue (Tab. 1): a dunitite from an  
105 Oman ophiolite rock, which contains 95% of olivine ( $[\text{Fe}_{0.2}\text{Mg}_{0.8}]_2\text{SiO}_4$ ),  
106 and minor amounts of plagioclase and pyroxene. This sample displays sig-  
107 natures of  $\text{H}_2\text{O}/\text{OH}$  in its reflectance spectrum (Fig. 6), due to the presence  
108 of small amounts of phyllosilicates as alteration products. We did not use  
109 glass samples, but we expect this component to behave as a bright phase  
110 due to the lack of iron.

111 **Group (ii):** A pyrrhotite sample was used as an analogue of opaque minerals  
112 (provided by Museum National d'Histoire Naturelle, Paris, France). Its  
113 composition was estimated to be  $\text{Fe}_{0.35}\text{S}$  by X-ray fluorescence. Pyrrhotite  
114 is sensitive to oxidation and was stored in a desiccator maintained under  
115 primary vacuum. Nevertheless, oxidation became very critical for ground  
116 samples containing submicrometer-sized grains. In one case, we observed  
117 an exothermic reaction during grinding that led to a brownish sample. X-  
118 ray diffraction measurements showed that this sample contained pyrrhotite  
119 46%, szomolnokite ( $\text{FeSO}_4 \cdot \text{H}_2\text{O}$ ) 38%, elemental sulfur 13% and mag-  
120 netite 3%. In other cases, a black powder was recovered, with a reflectance  
121 level lower than that of the brownish sulphate-rich sample. No iron metal  
122 was used due to the critical instability of iron nanoparticles exposed to air,  
123 which immediately got oxidised. We consider here that the sulphide com-  
124 ponent mimic the whole content of opaque materials.

Sample	Species	Chemical composition
Coal	Lignite (PSOC1532)	$C_{70.62}H_{5.32}O_{23.12}N_{0.94}$
Iron sulphide	Pyrrhotite	$Fe_{1-x}S$ with $0 < x < 0.20$
Silicate	Dunite (95% olivine)	$[Fe_{0.2}Mg_{0.8}]_2SiO_4$

Table 1: Material endmembers used in the synthesis of cometary analogues.

125 **Group (iii):** A lignite (PSOC 1532, provided by the Penn State University Data  
126 Bank) was used as an analogue of cometary refractory carbonaceous mate-  
127 rial. This immature coal has a composition fairly similar to chondritic IOM  
128 from primitive chondrites (Alexander *et al.*, 2007), a disordered polyaro-  
129 matic structure similar to that of refractory organics in stratospheric IDPs  
130 of presumed cometary origin (see discussion in Quirico *et al.* (2016)). The  
131 elemental composition of PSOC 1532 is  $C_{70.62}H_{5.32}O_{23.12}N_{0.94}$ , which is  
132 fairly similar to that of Insoluble Organic Matter extracted from primitive  
133 chondrites. This composition may be slightly different than that of cometary  
134 dust (Aléon *et al.*, 2001; Fray *et al.*, 2016), but the critical point here is to get  
135 a material with a polyaromatic structure, which mostly controls the optical  
136 properties (see discussion in Quirico *et al.* (2016)).

137

## 138 2.2 Sample preparation and characterisation

139 The endmembers defined above (Tab. 1) were ground and sieved. Raw materi-  
140 als were first hand-crushed and ground in a mortar then sieved to sort out grains  
141  $< 200 \mu m$ . Hand crushing was however not very suitable for soft and sticky coal  
142 samples, which usually led to large compact agglomerates. In this case, a dry 90  
143 minutes grinding was conducted with a planetary grinder Retsch PM100, mounted  
144 with a  $ZrO_2$  50 ml bowl and 2 mm balls. The second step consisted in a colloidal  
145 (in ethanol) grinding of this size fraction using the same apparatus with  $500 \mu m$   
146 balls, a 550 rpm speed rotation, and stopping periods of 30 s every 10 minutes.  
147 The colloidal solution was then sieved again ( $< 25 \mu m$ ) and transferred into a des-  
148iccator ( $60^\circ C$ , primary vacuum) to remove ethanol and the large clusters formed  
149 by re-agglomeration, or isolated big grains that might have escaped the grinding

150 process. Finally, a crust composed of a very fine-grained material was recovered,  
151 which we hand-crushed to obtain free submicrometer-sized grains mixed up with  
152 agglomerates (Fig. 1).

153 Binary (coal + pyrrhotite or coal + silicate) and ternary (coal + pyrrhotite + sil-  
154 icate) mixtures were prepared through two protocols. The first protocol was  
155 achieved with a MM200 grinder (Retsch). An agate mortar without milling balls  
156 was filled with several powders and shaken at 10-15 Hz during 5 minutes. The  
157 second protocol was represented by hand-mixing with a mortar and pestle. It was  
158 found much more efficient to reach fully intimate mixtures at a sub-micrometric  
159 scale (Fig. 2).

160

161

### 162 **2.3 IR reflectance spectroscopy**

163 Reflectance measurements were performed with a spectro-gonio-radiometer at the  
164 Institut de Planétologie et d'Astrophysique de Grenoble (IPAG). A full descrip-  
165 tion of this instrument can be found in Bonnefoy (2001) and Brissaud *et al.* (2004).  
166 The instrument is made of two pivoting arms, one carrying the visible and infrared  
167 detectors, the other carrying the illumination source. Contrary to Bonnefoy (2001)  
168 and Brissaud *et al.* (2004), we used a new measurement mode of the spectro-  
169 gonio-radiometer. The initial concept (Brissaud *et al.*, 2004) was to measure the  
170 reflected signal from a 2 cm diameter circular area within a 20 cm diameter illumi-  
171 nated area. In the new mode, the illumination is focused onto a 7.5 mm diameter  
172 area, without changing the size of the observed area. This new measurement mode  
173 provides an increased signal-to-noise ratio (x30), which is well adapted to studies  
174 of dark materials.

175 Measurements were performed at 0° incidence, 30° emergence, 0° azimuth  
176 angles and under ambient atmosphere. The calibrated data, given in REFF (re-  
177 flectance factor), were obtained by dividing the radiance signal coming from the  
178 detectors by the radiance from reference surfaces (Spectralon<sup>TM</sup>-SR99 for the 0.4-

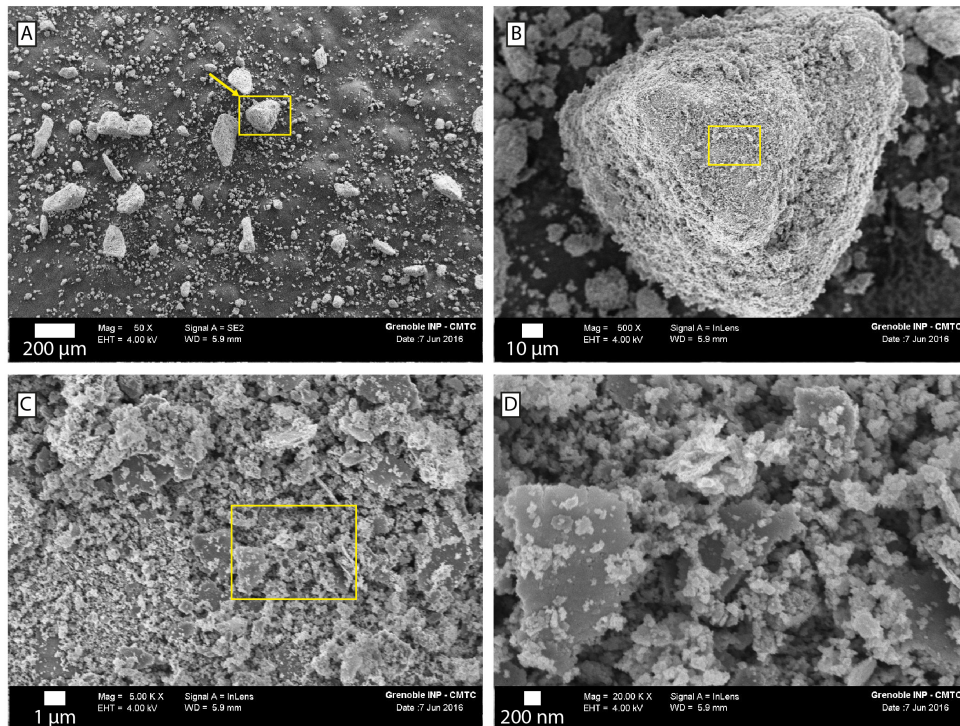


Figure 1: Scanning Electron Microscopy images of pyrrhotite after hand-crushing, 200  $\mu\text{m}$  sieving and colloidal grinding with a planetary grinder PM100 (Retsch). (a) The sample consists in free and agglomerated submicrometer-sized grains. The large clusters are fragile and easily dispersed with a needle under a binocular microscope. (b) Zoom on the agglomerate outlined by the yellow square and arrow in (a). (c, d) The same agglomerate with different magnifications. Grains are submicrometer-sized with a broad range of sizes. Note the presence of platelets, which likely point to re-agglomeration during grinding and/or desiccation.

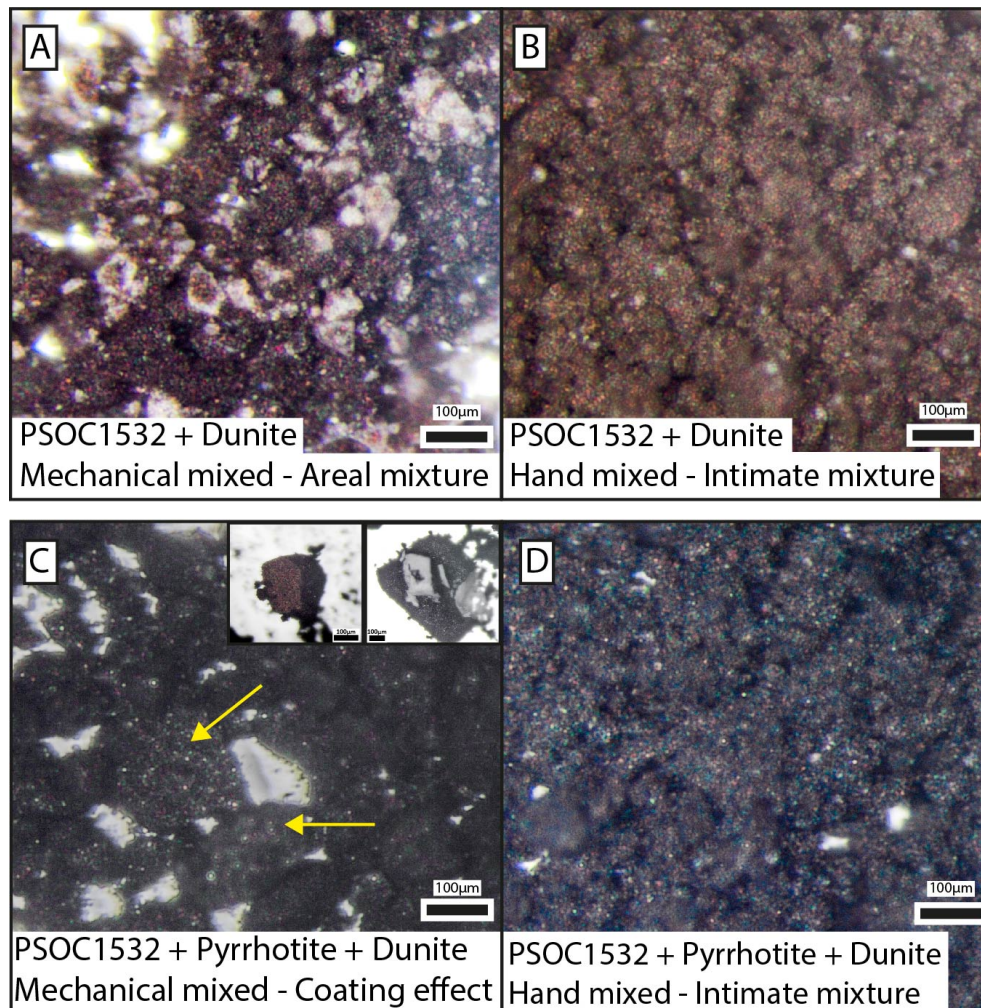


Figure 2: Optical binocular images of different mixtures sieved  $< 25 \mu\text{m}$  after a colloidal grinding with a planetary grinder. A) Areal mixture of PSOC 1532 and dunite mixed with grinder MM200 without milling balls. B) Intimate mixture of PSOC 1532 and dunite hand-mixed into a mortar with a pestle. C) PSOC 1532, dunite and pyrrhotite are mixed with the grinder MM200 as sample A. Pyrrhotite coats unbroken aggregates of dunite and organics (yellow arrow). Boxes in the upper right corner show broken aggregates of PSOC 1532 and dunite. D) Intimate mixture of PSOC 1532, dunite and pyrrhotite, hand mixed into a mortar.

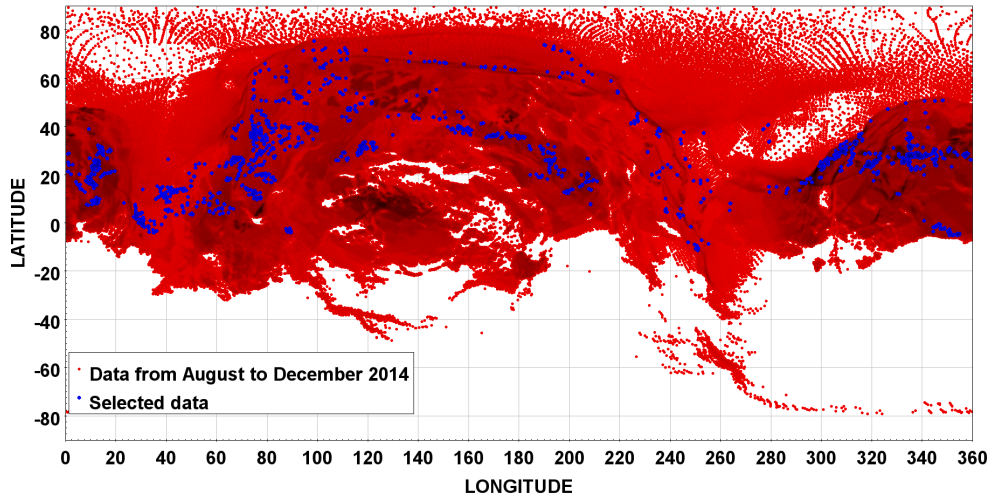


Figure 3: Partial map of the nucleus of 67P (observations from August to December 2014, corresponding to MTP006-MTP011 (MTP means Medium Term Plan). Each coloured point indicates the footprint of a spectrum. Pixels selected to compute the median spectrum of Fig. 4 are displayed in blue. The nucleus surface coverage is partial since the southern hemisphere was not illuminated at this time period.

179 2.5  $\mu\text{m}$  range and Infragold<sup>TM</sup> for the 2.5-4.0  $\mu\text{m}$  range). The spectral sampling  
 180 of the measurements was 20 nm.

## 181 2.4 VIRTIS data extraction

182

183

184 The VIRTIS spectrometer (Coradini *et al.*, 2007) has acquired reflectance  
 185 spectra from two channels: VIRTIS-M, an imaging spectrometer, ranging from  
 186 0.25  $\mu\text{m}$  to 5.1  $\mu\text{m}$  with  $\sim 2$  nm and  $\sim 10$  nm spectral sampling (in the visible and  
 187 infrared range, respectively) and VIRTIS-H, a point spectrometer ranging from  
 188 1.9  $\mu\text{m}$  to 5.1  $\mu\text{m}$  with a  $\sim 10$  times higher spectral resolution than VIRTIS-M.

189 A selection of spectra acquired by VIRTIS-M in August 2014 (see Fig. 3  
 190 and Appendix) is used in this study. Selecting this short time period minimises

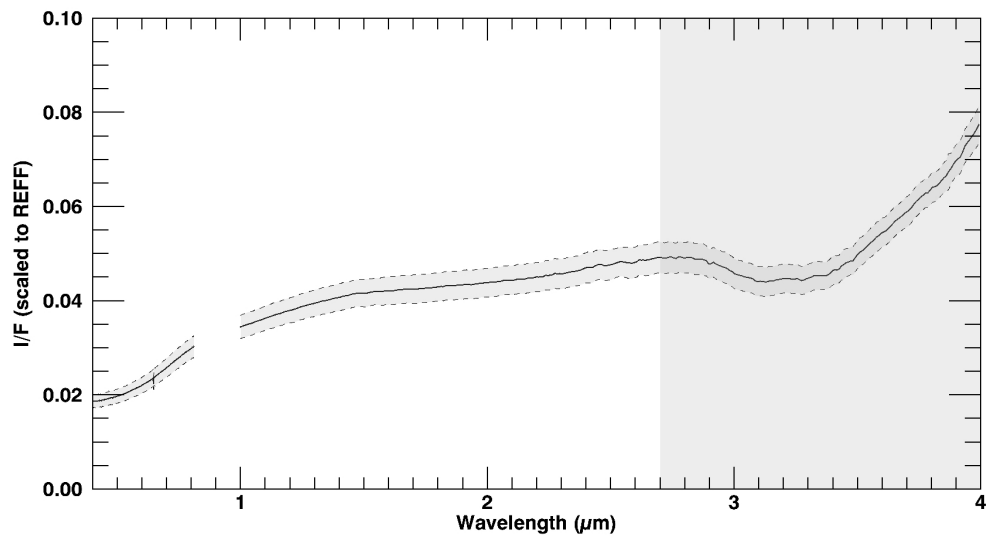


Figure 4: VIRTIS median spectrum of the surface of comet 67P, taken as the reference for this study and plotted in the range of our experimental data (0.4-4.0  $\mu\text{m}$ ). Thermal emission dominates longward of 4.0  $\mu\text{m}$  (not represented). The lack of data between 0.8  $\mu\text{m}$  and 1.0  $\mu\text{m}$  is due to a calibration artefact. Very small variations present at 1.5  $\mu\text{m}$  and 2.5  $\mu\text{m}$  are due to sorting filter junctions. The grey box on the right highlights the spectral range which is not studied because of the presence of the 3.2  $\mu\text{m}$  absorption band, the contribution of the thermal emission and measurement conditions of our mixtures.

191 activity-related variations of the surface spectral properties (Filacchione *et al.*,  
 192 2016; Ciarniello *et al.*, 2016). In addition, since the cometary heliocentric dis-  
 193 tance was 3.5 AU, the thermal emission relatively was weaker compared to later  
 194 mission stages. Fig. 4 shows the median of these spectra, calculated from 900  
 195 spectra. The selected data were restricted to incidence angles between 0° and 5°  
 196 and emergence angles between 28° and 32°. For each wavelength, we applied a  
 197 sigma clipping procedure, which allowed us to select the best observations, then  
 198 we computed the median spectrum. The error margins in Fig. 4 are given by the  
 199 standard deviation. In Fig. 3, the footprints of spectra that contribute to the me-  
 200 dian spectrum are indicated in blue. They are distributed over different regions  
 201 and therefore encompass the little spectral variability of the surface (Ciarniello  
 202 *et al.*, 2015; Fornasier *et al.*, 2015; Filacchione *et al.*, 2016; Quirico *et al.*, 2016).  
 203 The map includes acquisitions recorded between August and December 2014 (red  
 204 dots).

205 The radiance values  $I$  measured by VIRTIS are converted to reflectance factors

$$REFF(i, e, g, \lambda) = \frac{I(i, e, g, \lambda)}{F} \frac{1}{\cos(i)}$$

206 where  $F$  is the solar irradiance at 3.5 AU,  $i$ ,  $e$  and  $g$  are the incidence, emergence  
 207 and phase angles, respectively, and  $\lambda$  is the wavelength. This correction allows  
 208 one to compare VIRTIS data with data acquired in the laboratory. We observe  
 209 two different spectral slopes in the 67P nucleus spectra: one in the visible range  
 210 and the second in the infrared (Capaccioni *et al.*, 2015; Ciarniello *et al.*, 2015;  
 211 Filacchione *et al.*, 2016). A complex absorption band is visible, from 2.8  $\mu\text{m}$  to  
 212 3.5  $\mu\text{m}$  (Capaccioni *et al.*, 2015). This broad feature cannot be attributed to a  
 213 single compound. According to Quirico *et al.* (2016), organic  $-\text{OH}$ , aromatic  
 214  $\text{C}-\text{H}$ ,  $\text{CH}_2$  and  $\text{CH}_3$ -groups,  $\text{NH}_4^+$  ion are plausible species contributing to this  
 215 feature. The part of the spectrum longward of 3.5  $\mu\text{m}$  is dominated by the thermal  
 216 emissions from the surface, and even after correction for this thermal emission  
 217 no clear absorption features have been identified in this range. The present study  
 218 focuses on the part of the spectrum from 0.4  $\mu\text{m}$  to 2.7  $\mu\text{m}$ .

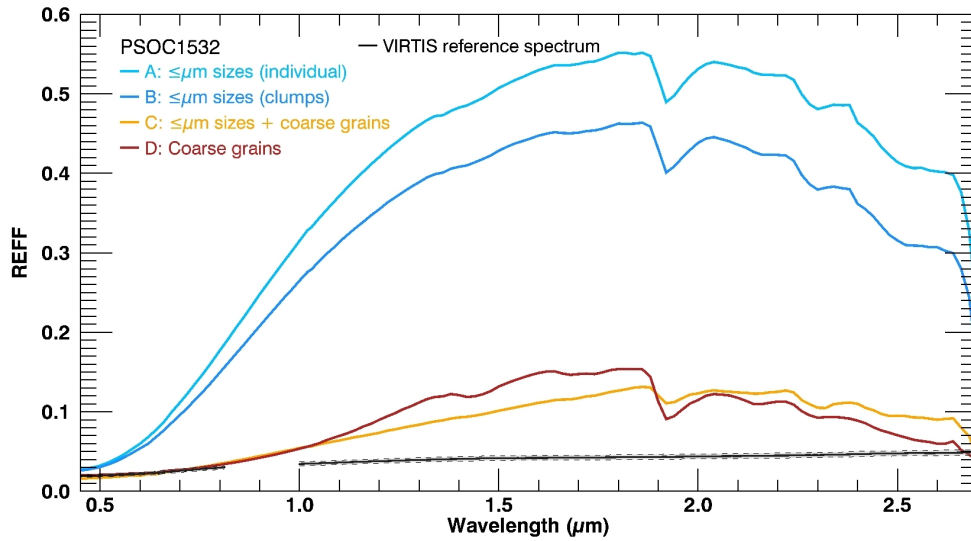


Figure 5: Spectra of PSOC1532 with various grain sizes compared to the VIRTIS spectrum. Spectra A to C correspond to samples after 180 min of colloidal grinding with the planetary grinder. Samples A and B (top) are sieved  $< 25\mu\text{m}$  after the grinding. The sample corresponding to spectrum B contains clumps, unlike sample A. Sample C consists of sub-micrometric particles, clumps and coarse unground grains (not sieved after planetary grinding). The spectrum D corresponds to the original PSOC1532 sample before grinding and sieving.

## 219 3 Results

### 220 3.1 Endmembers, the impact of sub- $\mu\text{m}$ grain

#### 221 3.1.1 PSOC 1532

222

223 The coal PSOC1532 shows a spectrum with a low reflectance in the visible  
 224 (less than 0.05 at 0.55  $\mu\text{m}$ , see Fig. 5), which increases towards longer wave-  
 225 lengths. The maximum reflectance value is reached around 1.9  $\mu\text{m}$ . A series of  
 226 overlapping absorption bands is observed between 2.2 and 2.6  $\mu\text{m}$ . These bands  
 227 are mostly due to combinations of various organic fundamental vibrational modes  
 228 and metal-OH in clay minerals. Absorption bands due to water and structural OH  
 229 are visible at 1.9  $\mu\text{m}$  and also at 1.4  $\mu\text{m}$  although less evident.

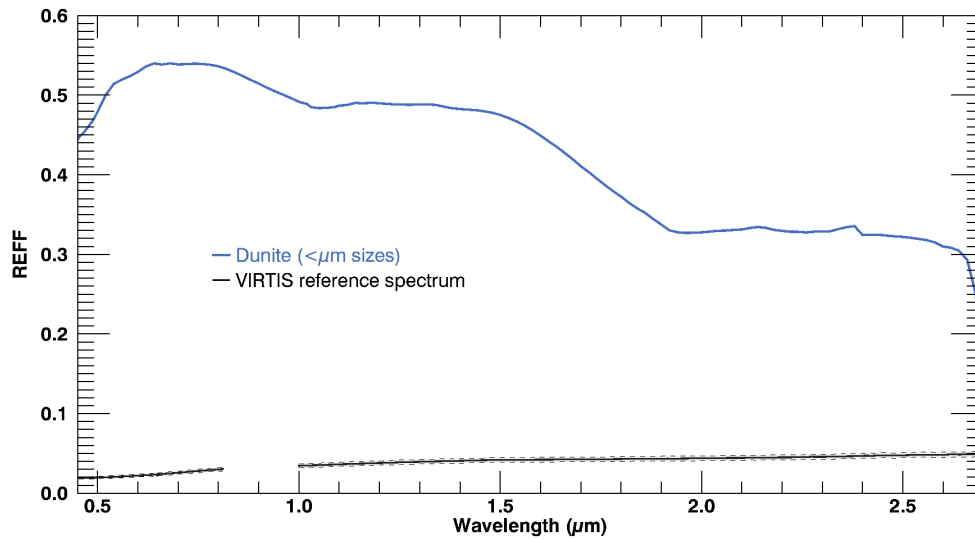


Figure 6: Dunite spectrum compared to VIRTIS reference spectrum. The sample of dunite is sieved < 25 μm after a 180 min colloidal grinding in the planetary grinder.

230 The spectra of the original material (before grinding or sieving) has a low re-  
 231 flectance in the visible (0.02 at 0.55 μm) and a maximum reflectance around 0.15  
 232 at 1.9 μm. The latter value is higher by a factor of 3 with respect to the cometary  
 233 nuclei. When ground to sub-μm grain size, the overall reflectance is increased by  
 234 a factor of 4. In a previous study (Quirico *et al.*, 2016) the impact of decreasing  
 235 grain size on the reflectance was determined for grain size down to 25-50 μm,  
 236 revealing a similar increase in the overall reflectance

### 237 3.1.2 Silicates

238

239 The reflectance spectrum of the dunite powder is presented in Fig. 6, where the  
 240 sample was ground to sub-μm grain size using the planetary grinder. The sample  
 241 remains relatively bright in the whole spectral range (reflectance above 0.3), with  
 242 a blue slope starting at ~0.7 μm, which is more pronounced than a typical dunite.  
 243 Absorption features are present around 1 and 2 μm. These absorptions are related  
 244 to the presence of Fe in the silicate (olivine and pyroxene solid solutions) (Burns,  
 245 1989). Note that this dunite is natural, and its Fe/(Fe + Mg) is around 20%.

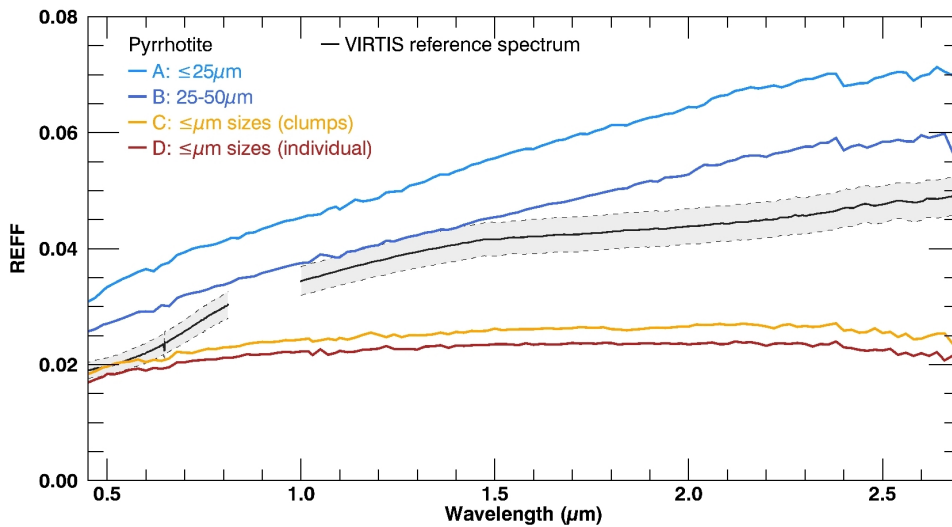


Figure 7: Spectra of ground pyrrhotite. The spectra A and B correspond to hand-ground and sieved samples. The sample C and D underwent 180 min colloidal grinding with the planetary grinding and contain only grains  $< 25 \mu\text{m}$ . The sample C contains clumps because it has not been hand mixed after drying unlike the sample of the spectrum D.

### 246 3.1.3 Iron sulphide (pyrrhotite)

247

248 The spectra obtained on the pyrrhotite with various grain sizes are presented  
 249 in Fig. 7. The spectra obtained for the hand-sieved fraction 25-50  $\mu\text{m}$  and  $< 25\mu\text{m}$   
 250 show a low reflectance (0.35 and 0.45 at 1  $\mu\text{m}$  respectively) slightly higher than the  
 251 value that was derived here for the surface of 67P. These two spectra have a strong  
 252 red slope across the whole spectral range. This spectral slope is typical of metals  
 253 when there is a contribution of specular reflection to the measured reflectance  
 254 (Cloutis *et al.*, 2010).

255 The spectrum of the sulphide that was ground to sub- $\mu\text{m}$  size with the planetary  
 256 grinder has a lower reflectance (0.02 at 1  $\mu\text{m}$ ), even lower than that measured  
 257 for comet 67P. The spectrum is also much flatter than that of sulphide with a  
 258 larger grain size. This difference is not due to a composition effect (the pyrrhotite  
 259 structure is preserved) but it is instead a pure grain size effect.

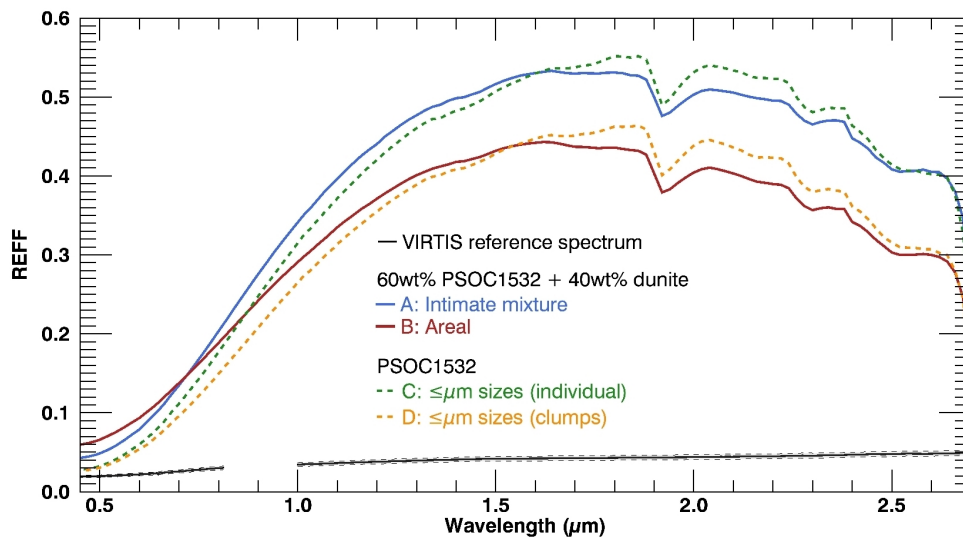


Figure 8: REFF spectra of: (A) intimate (see Fig. 2.B) and (B) areal mixtures (see Fig. 2.A) of PSOC 1532 and dunite compared to VIRTIS reference spectrum and pure PSOC 1532 spectra. Samples A and C are produced by hand mix while samples B is produced by mechanical mix (preserving clumps). Note that no coating occurs for the areal mixture.

## 260 3.2 Binary mixtures

261

262

263 In order to assess the spectral behaviour of mixtures with sub- $\mu\text{m}$  grains, two  
 264 different experiments were performed. In a first step, grains of dunite were inti-  
 265 mately mixed with PSOC 1532 and in a second step, a mixture between dunite  
 266 and sulphide was prepared. The intimate mixture was obtained by hand-mixing.  
 267 The silicate+coal mixture (Fig. 8) shows a non-linear behaviour. The spectrum  
 268 of the mixture is almost identical to the coal spectrum in the region where the  
 269 coal is strongly absorbing ( $< 1.5 \mu\text{m}$ ). In the region where the two endmembers  
 270 are bright (longward of  $1.5 \mu\text{m}$ ), the mixture spectra are somehow intermediate  
 271 between those two. Note that the mixture spectrum is not a good match to the  
 272 cometary spectrum and a darkening agent is needed in addition to dunite and the

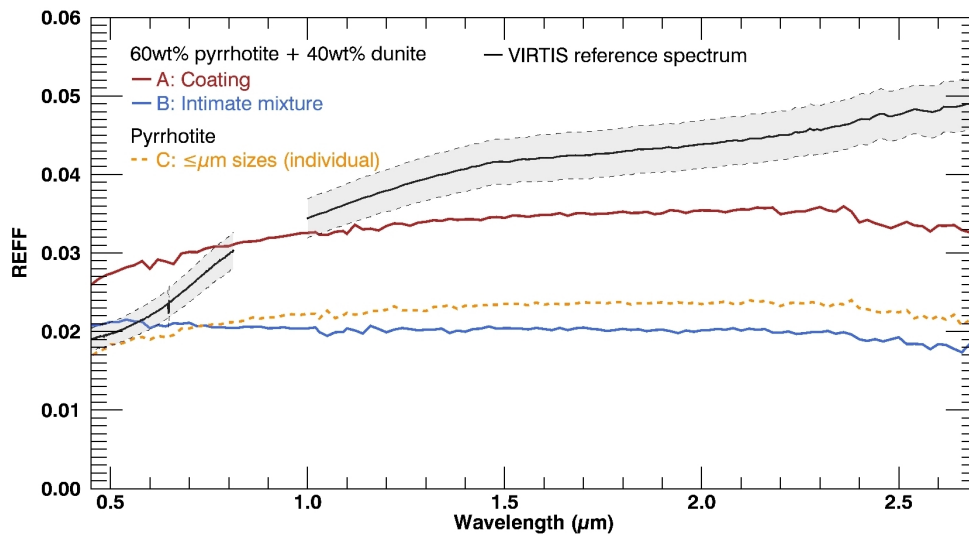


Figure 9: REFF spectra of (A) mixture with coating and (B) intimate mixture of 60wt% pyrrhotite and 40wt% dunite compared to VIRTIS reference spectrum and a pure sample of pyrrhotite (C). The sample C does not contain clumps and has the same texture of mixture B.

273 organic phase in order to explain the low reflectance of the nucleus from the VIS  
 274 to the NIR.

275 The spectrum of the intimate mixture between sub- $\mu\text{m}$  sulphide powder (opaque)  
 276 and a dunite powder (more reflective) is shown in Fig. 9. The mixture has a low re-  
 277 flectance value, its spectral signature being almost identical to that of the sulphide.  
 278 The behaviour of the mixture is again non-linear, the silicate signature is totally  
 279 obscured by the presence of the sulphide. The spectrum of the silicate-sulphide  
 280 binary mixture has a reflectance level lower than the comet, but the red-spectral  
 281 slope is not reproduced.

### 282 3.3 Ternary mixtures

283

284 In Fig. 10, ternary mixture of sub- $\mu\text{m}$  sulphides, coals and silicates are pre-  
 285 sented. The samples were prepared using a constant sulphide to organics ratio  
 286 (1/3 vs 2/3 in the relative mass fraction) and increasing fraction of dunite (5, 10,

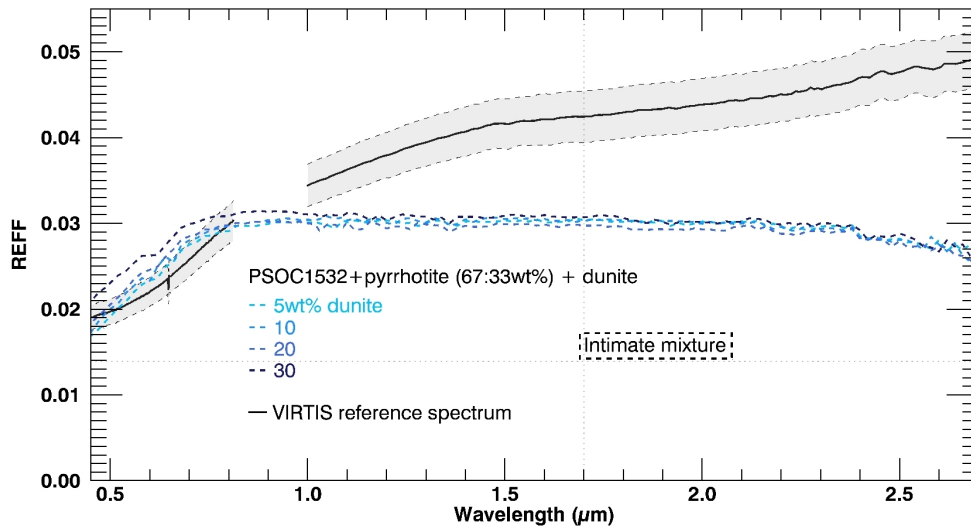


Figure 10: REFF spectra of intimate mixtures of PSOC 1532, pyrrhotite and dunite. The coal/pyrrhotite ratio is kept constant. Dunite varies from 5wt% to 30wt% but does not lead to strong variations of the spectral shape.

287 20, 30 wt.%). In these experiments, the C/Fe is roughly solar (10.5 for the mix-  
 288 ture against 9.5 for a solar composition, (Lodders, 2003). The C/Si ratio varies  
 289 according to the proportion of dunite, within a range from 6.3 (5 wt%) to 51.6 (30  
 290 wt%) against 8.3 for a solar composition (Lodders, 2003). Given that some of the  
 291 solar carbon might have been hosted by gaseous species (CO, CN, CO<sub>2</sub>, HCN)  
 292 the solar C/Si value is probably an upper limit for cometary C/Si.

293 The spectra of the ternary mixture are similar to those of the sulphide-silicate mix-  
 294 ture. Vis slope and albedo approximate the 67P spectrum as measured by VIRTIS.  
 295 On the contrary, the slope is too flat and albedo too low in the IR. The addition  
 296 of a significant organic fraction does not help in reproducing the spectral slope  
 297 typical of cometary dust as observed by VIRTIS. The fraction of silicate has only  
 298 a marginal impact on the reflectance spectra of the mixture. The various mixture  
 299 spectra are almost identical with the exception of the visible. Note that this is the  
 300 region where grain size begins to be of the order of the wavelength and change of  
 301 scattering regime might occur.

## 302 4 Discussion

### 303 4.1 The spectral behaviour of sub- $\mu\text{m}$ particles

304 Radiative transfer in a particulate media is a combination of reflection, absorption,  
305 and diffraction. When the grain size is much higher than the wavelength, two ma-  
306 jor scattering regimes are often described. First, a surface scattering regime, when  
307 most of the light seen by the observer has underwent one, two or more reflections  
308 on grain surfaces. This is the case of strongly absorbing material (sulphides for  
309 example). Second, a volume scattering regime, which is typical of materials with  
310 low absorption coefficients in the wavelength range of interest, for which most of  
311 the light seen by the observer has gone through at least one grain, and often sev-  
312 eral ones. This is the case of Fe-poor silicates, which are translucent in the VNIR  
313 range.

314 These two regimes can be dealt with and modelled according to geometric optics.  
315 In general, for the volume scattering regime, the impact of decreasing grain size is  
316 to increase the overall reflectance and decrease absorption band depths (Adams &  
317 Filice, 1967). This effect has been observed for carbonaceous chondrites (John-  
318 son & Fanale, 1973) or coal materials (Quirico *et al.*, 2016). In the case of surface  
319 scattering, the impact of grain size is less clear, since grain shape and orientation  
320 can have a first-order impact on the reflectance level. In the case of troilite, the  
321 finest grain size studied by Britt *et al.* (1992) appeared to have the lowest overall  
322 reflectance. This is also the case of powdered iron meteorite (Cloutis *et al.*, 2015).  
323 When grain size is below the wavelength, geometric optics cannot be used and a  
324 change of scattering regime is expected to occur when grain size approaches  $\lambda/\pi$   
325 (Hapke, 2012). An isolated particle experiences a strong decrease in its scattering  
326 cross section and becomes a perfect absorber. Nevertheless, particules may be not  
327 isolated within a regolith, and the behaviour would be different (e.g. leading to  
328 a reduced contribution of the diffraction to the scattering efficiency). One would  
329 expect then to observe a decrease in the overall reflectance (for both absorbing  
330 and non-absorbing materials) when grain size becomes  $< \lambda/\pi$  as was observed  
331 for silicates in the mid-IR (Mustard & Hays, 1997).

332 These effects are clearly not observed for our sub- $\mu\text{m}$  samples of silicates and

333 coals. Both samples when ground below 0.5  $\mu\text{m}$  still remain generally bright and  
334 do not appear to have experienced a change of scattering regime. The deviation  
335 from the theoretically expected behaviour is due to the presence of inter-particle  
336 interaction. The presence of electro-magnetic coupling between grains (grains  
337 sticking was often observed under the SEM) might lead to a difference between  
338 the physical grain size ( $< 1 \mu\text{m}$ ) and the “optical grain size”. In other words, our  
339 reflectance spectra of sub- $\mu\text{m}$  silicate and coals very much resemble that of the  
340 VNIR spectra 10  $\mu\text{m}$  sized samples.

341 In the case of the sulphides, a different behaviour is observed, which appears to  
342 be in better agreement with the theoretical prediction. The reflectance spectrum  
343 obtained for the pyrrhotite sample with the smallest grain size has a different slope  
344 (less red) and a lower overall reflectance with respect to the sample with a bigger  
345 grain sizes. Such a behaviour can be interpreted by a decrease of the contribution  
346 of specular reflection. As shown by [Britt & Pieters \(1988\)](#), the reflectance spectra  
347 of metallic surfaces of low roughness measured in the specular or near specular  
348 direction show a strong red slope and a larger reflectance, while at different ge-  
349 ometries the spectrum is darker and characterised by a flat to blue slope ([Britt &  
350 Pieters, 1988](#)). Therefore, in the case of the sub- $\mu\text{m}$  sulphide, the grain size (and  
351 therefore the surface roughness) is low enough to minimise the contribution of  
352 a specular (coherent) reflection by any part of the sample. This “quenching” of  
353 the specular reflection is likely to produce the change of spectra observed for our  
354 sub- $\mu\text{m}$  sulphide. Contrary to the coal and the dunite, the pyrrhotite is less subject  
355 to formation of aggregates, resulting to a optical grain size closer to the real grain  
356 size.

## 357 **4.2 Spectral behaviour of mixtures**

358

359 The spectral behaviour of a sub- $\mu\text{m}$  grains powders was investigated also for  
360 sulphide/silicate, as well as for coal/silicate mixtures (Fig. 8, 9). The correspond-  
361 ing spectra clearly show a non-linear behaviour, i.e. the reflectance of the mix-  
362 ture is not the weighted average of the endmember spectra. In both cases, the

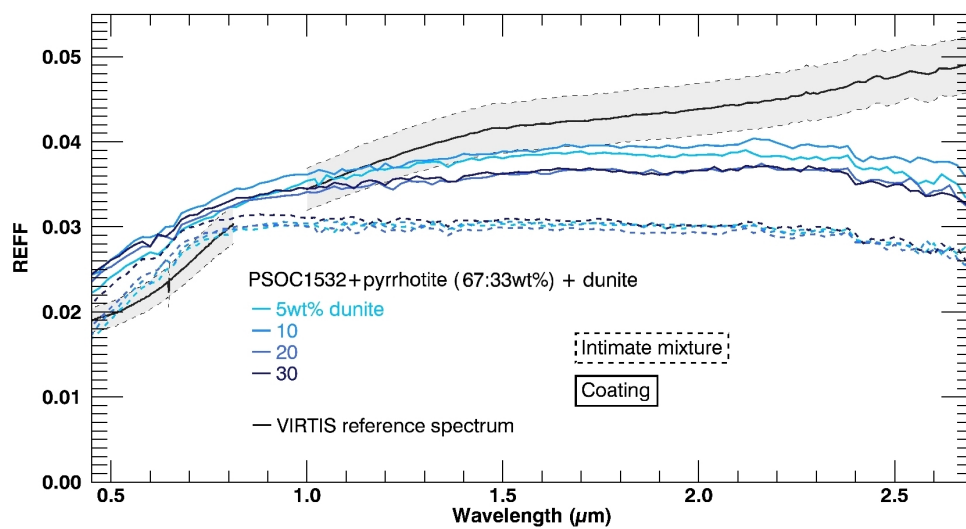


Figure 11: Same figure as Fig. 10 with mechanically mixed ternary mixtures of PSOC 1532, pyrrhotite and dunite. This protocol preserves clumps of organic and silicate which are covered by pyrrhotite (coating mixture, see text). The resulting spectra similar to the ones of the intimate mixtures in the visible while they are redder and brighter in the IR range. Various proportions of dunite do not affect strongly the spectral behaviour.

363 reflectance of the mixture appears to be dominated by the dark material when  
364 present. The coal-dunite mixture has a spectrum close to the coal spectra below  
365 1.5  $\mu\text{m}$ . In the wavelength range where both materials are relatively bright (above  
366 1.5  $\mu\text{m}$ ), the reflectance of the mixture is intermediate between the two. In the  
367 case of the dunite-sulphide mixture, the spectrum of the mixture is almost identi-  
368 cal to that of the pyrrhotite.

369 When mixing two endmembers (bright and dark) with similar grain size, the re-  
370 flectance of the mixture is non-linear and tends to be “biased” toward the dark  
371 material (i.e. [Pommerol & Schmitt \(2008\)](#)) but not to the extent observed here  
372 for the sulphide-dunite mixture or the coal-dunite mixture (the coal being quite  
373 dark in the IR). Laboratory studies have shown that an efficient darkening can be  
374 obtained only when the grain sizes are contrasted (large grains of dark materials,  
375 small grains of bright material). This work shows that a strong darkening can be  
376 obtained when mixing two materials with similar grain size, if below the micron  
377 size. This observation is reinforcing our earlier suggestion that sub- $\mu\text{m}$  sulphides  
378 and metals are prime actors in explaining the darkness of cometary surfaces ([Ca-  
379 paccioni \*et al.\*, 2015](#); [Quirico \*et al.\*, 2016](#)).

380 When mixing a bright and dark materials with similar grain size  $\gg \lambda/\pi$ , photons  
381 travelling through the media are reflected or scattered by the bright material and  
382 reflected or absorbed by the dark one. The overall reflectance of the mixture is,  
383 therefore, brighter than the dark endmember due to reflections and scattering by  
384 the bright material. When grain size is below  $\lambda/\pi$ , we can suppose that reflection  
385 and scattering by the bright material become negligible and that the dark material  
386 will ultimately absorb any photon travelling through the media. The spectrum of  
387 the mixture is therefore only controlled by the absorption properties of the dark  
388 material.

### 389 **4.3 How to reproduce both the spectral slope and the low re- 390 flectance level?**

391 A first attempt to reproduce the reflectance spectra of the cometary nucleus was  
392 done through ternary mixtures of silicate, coal and sulphide (Fig. 10). The  
393 coal/sulphide ratio was kept constant (2/3 coal, 1/3 sulphide in weight) with a

394 C/Fe about solar, and increasing fractions of silicate were added to this mixture  
395 (5, 10, 20, 30 wt%). Mixtures were prepared manually by grinding in an agate  
396 mortar, which insured intimate mixing at the finest scale (Fig. 1, 2). The spectra  
397 of these mixtures are almost identical to those of the sub- $\mu\text{m}$  sulphide, confirming  
398 that intimate mixtures of sub- $\mu\text{m}$  material are strongly dominated by absorbing  
399 particles when present. Still, a small impact of the silicate content is observed  
400 around 0.7  $\mu\text{m}$ , but this could be due to a change of scattering regime since grain  
401 size becomes closer to the wavelength.

402 These ternary intimate mixtures failed to reproduce the 67P's spectrum. A first  
403 possibility is that the type of mixing used in our experiments is not adequate, and  
404 that not all compounds are intimately mixed, but that large aggregates of a given  
405 component are present. The identification of organics-rich grains in 67P (Fray  
406 *et al.*, 2016) would suggest some level of heterogeneity in the dust-grain aggregates.  
407 In order to produce some level of heterogeneity in our samples, mixtures  
408 of sub- $\mu\text{m}$  grains were prepared using a mechanical grinder. Such a preparation  
409 protocol was able to preserve clumps formed during the drying of pure phases  
410 (especially for silicate and organics). In such a sample, these 100  $\mu\text{m}$ -sized aggregates  
411 are covered by pyrrhotite (Fig. 2). If the coating is not complete, the sample  
412 has the potential to behave more like an areal mixture with the reflectance of the  
413 mixture being a linear combination of the spectral endmember weighted by the  
414 fraction of covered area. This type of mixture has the potential to produce a dark  
415 and red spectrum, more similar to 67P's (Fig. 11). Although the spectra of the  
416 mixture obtained are not yet a perfect match to the one of the comet, with the mixture  
417 reflectance levels too low in the IR, a red spectral slope is present up to 2.3  
418  $\mu\text{m}$ . This slope is likely due to the presence of a few organics-rich aggregates in  
419 the sample, which, unlike the sulphide, preserve their spectral slope when ground  
420 below the  $\mu\text{m}$ .

421 Other alternatives can be proposed to explain the spectral slope of comet 67P surface  
422 material. First, a sample that mix sub- $\mu\text{m}$  grains with other sizes (e.g. ~100-  
423 200  $\mu\text{m}$ ) might be necessary to retrieve the spectral slope of the VIRTIS spectrum  
424 (Moroz *et al.*, 2016; Rousseau *et al.*, 2016, 2017) while being more representative  
425 of a comet surface. In such a sample, however, the scattering behaviour is complex.  
426 The organics material used in these experiments might not be red enough

427 (not H-rich enough) to impact the spectral slope when intimate mixture grains of  
428 organics, sulphides and silicates are prepared (with sub- $\mu\text{m}$  grains). In situ mea-  
429 surements by the COSIMA instrument suggest that the bulk organic compounds  
430 found in 67P particles are similar to the insoluble organic matter (IOM) found in  
431 meteorite and IDP (the IOM which is relatively similar to our coal) but that it is  
432 richer in hydrogen than the IOM (Fray *et al.*, 2016). Another possibility is that our  
433 prepared mixture has a too high sulphide/organic ratio, which tends to erase the  
434 organic signatures. These experiments were designed in order to have C/Fe ratio  
435 close to solar. This value was taken as an upper bound since some of the solar car-  
436 bon should have been distributed within gaseous species (CO, CN, CO<sub>2</sub>, HCN)  
437 in addition to organic compounds, while most of the iron is expected to have been  
438 present in a condensed reduced phase in early Solar System grains (Fe-metal and  
439 Fe-sulphides). Still, some of the iron could have been present within the silicates.  
440 Further studies will include mixtures with higher organic/sulphide ratios.

## 441 **5 Conclusions**

442 From the obtained results the following conclusions can be drawn:

- 443 • A method was developed to produce large samples made of sub- $\mu\text{m}$  indi-  
444 vidual grains which enables to produce cosmochemically relevant cometary  
445 analogues. Therefore, we can study a radiative transfer regime where the  
446 physical grain size is below the wavelength. Intimate mixtures of organics  
447 (coal), sulphides and silicates were produced in which the typical grain size  
448 is a few 100s nanometres.
- 449 • When ground below the  $\mu\text{m}$  scale, a brightening is observed for moderately  
450 absorbing materials (coal and silicate) when compared to grain size  $> \mu\text{m}$ ,  
451 while one would expect a darkening given that the scattering cross section of  
452 particles with size smaller than the wavelength should decrease. This sug-  
453 gests that for this type of material the “optical” grain size is still larger than  
454 wavelength because of clumping of the particles due to electro-magnetic  
455 interactions. In the case of the opaque material (sulphides), when ground  
456 below the  $\mu\text{m}$ , a darkening is observed as well as a decrease in the spectral

457 slope. This effect is interpreted as due to the reduction of the scattering cross  
458 section for very small particles (size  $< \lambda/\pi$ ) and by a decrease of specular  
459 reflection contributions to the measured reflectance.

- 460 • Sub- $\mu\text{m}$  sulphides can efficiently darken a sample within intimate mixtures  
461 of sub- $\mu\text{m}$  materials. This reinforces earlier suggestions (Capaccioni *et al.*,  
462 2015; Quirico *et al.*, 2016) that Fe-rich opaques (sulphides and (Fe, Ni) met-  
463 als) are the major contributor to the darkness of 67P and probably comets  
464 in general.
- 465 • The prepared intimate mixtures are able to reproduce the low reflectance  
466 levels observed for comet 67P, but the spectral slope remains too small. A  
467 possibility that was explored is that a few percent of organic-rich aggregates  
468 may induce this slope. A mechanically prepared mixture where organic-rich  
469 aggregates are present (as well as sulphide-rich and dunite-rich aggregates)  
470 show a more pronounced spectral slope.
- 471 • The peculiar slope of 67P, cometary nuclei and D-type asteroids may be  
472 explained by the presence of H-rich organics (redder than the used coal)  
473 and by a larger organic/opaque ratio than used in this study.

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489

Visible	Infrared
V1_00366679119	I1_00366679117
V1_00366686318	I1_00366686316
V1_00366693519	I1_00366693517
V1_00366700719	I1_00366700717
V1_00366707918	I1_00366707916
V1_00366725919	I1_00366725917
V1_00366729519	I1_00366729517
V1_00366733119	I1_00366733117
V1_00366736719	I1_00366736717
V1_00366740319	I1_00366740317
V1_00366743919	I1_00366743917
V1_00366747519	I1_00366747517
V1_00366751119	I1_00366751117
V1_00366754719	I1_00366754717
V1_00366758319	I1_00366758317
V1_00366765519	I1_00366765517

Table 2: List of the cubes used for compiling the median VIRTIS spectrum. They were acquired at the beginning of the Rosetta mission at the comet during MTP006/STP013 (14 & 15 August 2014).

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