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Seeds of Life in Space (SOLIS)

V. Methanol and acetaldehyde in the protostellar jet-driven shocks L1157-B0 and B1

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ABSTRACT

Context. It is nowadays clear that a rich organic chemistry takes place in protostellar regions. However, the processes responsible for it, that is, the dominant formation routes to interstellar complex organic molecules, are still a source of debate. Two paradigms have been evoked: the formation of these molecules on interstellar dust mantles and their formation in the gas phase from simpler species previously synthesised on the dust mantles.

Aims. In the past, observations of protostellar shocks have been used to set constraints on the formation route of formamide (NH_2CHO), exploiting its observed spatial distribution and comparison with astrochemical model predictions. In this work, we follow the same strategy to study the case of acetaldehyde (CH_3CHO).

Methods. To this end, we used the data obtained with the IRAM-NOEMA interferometer in the framework of the Large Program SOLIS to image the B0 and B1 shocks along the L1157 blueshifted outflow in methanol (CH_3OH) and acetaldehyde line emission.

Results. We imaged six CH_3OH and eight CH_3CHO lines which cover upper-level energies up to ~ 30 K. Both species trace the B0 molecular cavity as well as the northern B1 portion, that is, the regions where the youngest shocks (~ 1000 yr) occurred. The CH_3OH and CH_3CHO emission peaks towards the B1b clump, where we measured the following column densities and relative abundances: $1.3 \times 10^{16} \text{ cm}^{-2}$ and 6.5×10^{-6} (methanol), and $7 \times 10^{13} \text{ cm}^{-2}$ and 3.5×10^{-8} (acetaldehyde). We carried out a non-LTE (non-Local Thermodynamic Equilibrium) Large Velocity Gradient (LVG) analysis of the observed CH_3OH line: the average kinetic temperature and density of the emitting gas are $T_{\text{kin}} \sim 90$ K and $n_{\text{H}_2} \sim 4 \times 10^5 \text{ cm}^{-3}$, respectively. The CH_3OH and CH_3CHO abundance ratio towards B1b is 190, varying by less than a factor three throughout the whole B0–B1 structure.

Conclusions. Comparison of astrochemical model predictions with the observed methanol and acetaldehyde spatial distribution does not allow us to distinguish whether acetaldehyde is formed on the grain mantles or in the gas phase, as its gas-phase formation, which is dominated by the reaction of ethyl radical (CH_3CH_2) with atomic oxygen, is very fast. Observations of acetaldehyde in younger shocks, for example those of $\sim 10^2$ yr old, and/or of the ethyl radical, whose frequencies are not presently available, are necessary to settle the issue.

Key words. stars: formation – ISM: jets and outflows – ISM: molecules – ISM: individual objects: L1157

1. Introduction

One of the open questions in astrochemistry refers to how chemistry richness develops and evolves in star forming regions. More specifically, the molecular complexity during the evolutionary stages that lead a molecular cloud to form a solar-type star and its planetary system has yet to be elucidated. Particularly relevant are the abundances of the so-called interstellar complex organic molecules (iCOMs: molecules with at least six atoms: Herbst & van Dishoeck 2009; Ceccarelli et al. 2017), as they may be considered as small blocks from which complex prebiotic species can build up (e.g. Caselli & Ceccarelli 2012;

Belloche et al. 2014). The present consensus is that after an initial cold phase in which icy mantles enriched with hydrogenated species form on the dust grains (e.g. Tielens & Hagen 1982; Watanabe & Kouchi 2002; Rimola et al. 2014), the iCOMs synthesis can either occur on the grain mantle surfaces initiated by energetic processing (e.g. Garrod & Herbst 2006; Garrod et al. 2008; Ruaud et al. 2015; Vasyunin et al. 2017) or in the gas phase initiated by the sublimation of the mantles themselves (e.g. Charnley et al. 1992; Balucani et al. 2015; Codella et al. 2017; Skouteris et al. 2018). In both cases, the composition of the grain mantles therefore plays a paramount role in the gaseous, observable abundance of iCOMs (see e.g. Burkhardt et al. 2019) – in the

former possible scenario, because the mantle species are directly injected from the mantles into the gas; and in the latter possible scenario, because they feed gas reactions leading to iCOMs.

In this context, protostellar shocks are particularly precious because they can provide not only the observed iCOM abundances but also the approximate times at which they are formed. Indeed, at the passage of the shock, molecules present in the grain mantles are injected into the gas phase either because of sputtering (gas–grain collisions) or shattering (grain–grain) processes (e.g. Flower & Pineau des Forêts 1994; Caselli et al. 1997; Schilke et al. 1997; Gusdorf et al. 2008a,b). Once in the gas phase, the injected mantle species undergo reactions that destroy and/or form iCOMs (unless multiple shocks occur at the very same place, which is unlikely): therefore, the gaseous iCOM abundances are intrinsically time dependent. Fortunately, the chemical evolution time, which is a few thousand years at most, is of the order of the age of young protostellar shocks (e.g. Herbst & van Dishoeck 2009; Ceccarelli et al. 2017). Thus, sufficiently high spatial resolution observations towards these objects allow us to decipher the different time-dependent abundances of iCOMs and consequently provide extremely strong constraints on the formation and destruction of iCOMs. For example, this method was applied to constrain the formation route of formamide in the jet-driven protostellar shock L1157-B1 (Codella et al. 2017; hereinafter CC17). Previous observations have shown that this latter shock is extremely rich in iCOMs (Arce et al. 2008; Lefloch et al. 2017), suggesting that it may be an ideal case to look for others with the same method as that used by Codella et al. (2017).

In this article, we continue the systematic study of the formation and destruction routes for iCOMs in the L1157-B1 shock started in CC17. Here we focus on the acetaldehyde (CH_3CHO) and methanol (CH_3OH). As in CC17, we make use of the observations from the IRAM/NOEMA (NORthern Extended Millimeter Array¹) Large Project SOLIS (Seeds Of Life In Space; Ceccarelli et al. 2017), whose goal is to investigate the formation and destruction of iCOMs during the early stages of solar-type star formation. The article is organised as follows: in Sect. 2 we describe the source and summarise the previous studies of its astrochemistry. In Sect. 3 we describe the observations, in Sect. 4 we report the results of the analysis of these observations, in Sect. 5 we discuss the impact of those results, and in Sect. 6 we present our conclusions.

2. L1157-B1: an astrochemical laboratory

The chemically rich molecular outflow driven by the L1157-mm Class 0 protostar ($d = 352$ pc; Zucker et al. 2019) has been extensively used to investigate how the gas chemistry is altered by the injection of the species frozen onto the dust mantles. L1157-mm drives an episodic and precessing jet (Gueth et al. 1996; Podio et al. 2016), which produced several cavities well observed using both single-dish antennas and interferometers (e.g. Gueth et al. 1998; Lefloch et al. 2012). In particular, the southern blueshifted outflow is associated with several shocked regions produced by impacts between jets and cavities called B0, B1, and B2 (see also Fig. 1). The brightest shocked region, B1, is one of the targets of the IRAM NOEMA SOLIS Large Program. In particular, B1 consists of a series of shocks caused by different episodes of ejection hitting the cavity wall. These shocks produced a clumpy structure (identified using several molecular species, see e.g. Tafalla & Bachiller 1995; Codella et al. 2009;

¹ <http://iram-institute.org/EN/noema-project.php>

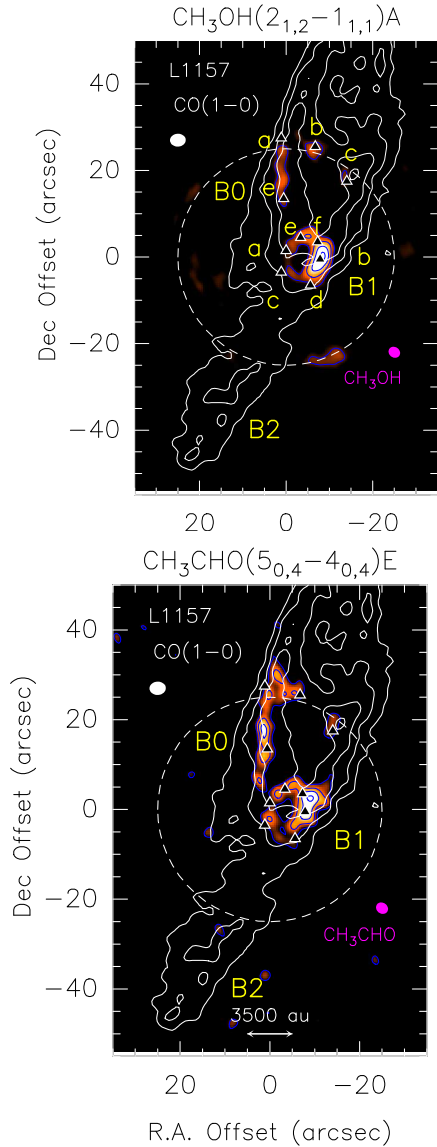


Fig. 1. L1157 southern blueshifted lobe in CO (1–0) (white contours; Gueth et al. 1996). The precessing jet ejected by the central object L1157-mm (outside the frame and toward the northwest) excavated several clumpy cavities, named B0, B1, and B2, respectively. The maps are centred at $\alpha(\text{J2000}) = 20^{\text{h}}39^{\text{m}}10^{\text{s}}.2$, $\delta(\text{J2000}) = +68^{\circ}01'10''.5$ ($\Delta\alpha = +25''$ and $\Delta\delta = -63''.5$ from the L1157-mm protostar). *Top panel:* map (in colour scale) of the emission due to the $\text{CH}_3\text{OH}(2_{1,2}-1_{1,1})$ A transition (integrated over the whole velocity range). The first contour and step are 3σ ($32 \text{ mJy beam}^{-1} \text{ km s}^{-1}$). For the CO image, the first contour and step are 6σ ($1\sigma = 0.5 \text{ Jy beam}^{-1} \text{ km s}^{-1}$) and 4σ , respectively. The dashed circle shows the primary beam of the CH_3OH image ($64''$). Yellow labels are for the B0 and B1 clumps (black triangles) previously identified using molecular tracers (see text, and e.g. Benedettini et al. 2007; Codella et al. 2009, 2017). The white and magenta ellipses depict the synthesised beams of the CO ($3''.65 \times 2''.96$, $\text{PA} = +88^\circ$), and CH_3OH ($2''.97 \times 2''.26$, $\text{PA} = -155^\circ$) observations, respectively. *Bottom panel:* same as *top panel* but for $\text{CH}_3\text{CHO}(5_{0,4}-4_{0,4})$ E. The first contour and step are 3σ ($7 \text{ mJy beam}^{-1} \text{ km s}^{-1}$). The synthesised beams of the CH_3CHO image are similar to those of the CH_3OH map.

Lefloch et al. 2012; Benedettini et al. 2013), which can be classified as follows: (i) a northern arch-like structure composed of the clumps B1a-e-f-b (see Fig. 1), plus (ii) two clumps, labelled B1c and B1d, which are the oldest ones (kinematical age ~ 1400 yr, from Podio et al. 2016, corrected for the new distance of 352 pc

Table 1. CH₃OH and CH₃CHO emission lines observed using NOEMA and the WideX backend towards the L1157-B1b peak (see Fig. 2).

Transition ^(a)	ν_0 ^(a) (GHz)	E_u ^(a) (K)	$S\mu^2$ ^(a) (D ²)	$\log(A_{ij})$ ^(a)	rms (mK)	I_{int} ^(b) (K km s ⁻¹)	T_{rot} (K)	N_{tot} (cm ⁻²)
CH ₃ OH							10.0(1.1)	$1.3(0.3) \times 10^{16}$
2 _{1,2} –1 _{1,1} A	95.91431	21	1.2	–5.6	14	9.05(0.15)		
2 _{–1,2} –1 _{–1,1} E ^(c)	96.73936	13	1.2	–4.6	9	40.5(1.5) ^(d)		
2 _{0,2} –1 _{0,1} A ^(c)	96.74138	7	1.6	–5.6	9	57.2(1.5) ^(d)		
2 _{0,2} –1 _{0,1} E ^(c)	96.74455	20	1.6	–5.5	9	19.0(0.5) ^(d)		
2 _{1,1} –1 _{1,0} E	96.75551	28	1.2	–5.5	9	6.69(0.09)		
2 _{1,1} –1 _{1,0} A	97.58280	22	1.2	–5.6	11	12.82(0.13)		
CH ₃ CHO							8(1)	$7(3) \times 10^{13}$
5 _{0,5} –4 _{0,4} E	95.94744	14	63.2	–4.5	5	1.71(0.14)		
5 _{0,5} –4 _{0,4} A	95.96346	14	63.2	–4.5	7	1.65(0.15)		
5 _{2,4} –4 _{2,3} A	96.27425	23	53.1	–4.6	8	0.65(0.15)		
5 _{2,4} –4 _{2,3} E	96.42561	23	52.8	–4.6	9	0.40(0.08)		
5 _{2,3} –4 _{2,2} E	96.47552	23	52.8	–4.6	11	0.44(0.16)		
5 _{2,3} –4 _{2,2} A	96.63266	23	53.1	–4.6	12	0.42(0.09)		
5 _{1,4} –4 _{1,3} E	98.86331	17	60.7	–4.5	12	1.42(0.19)		
5 _{1,4} –4 _{1,3} A	98.90094	17	60.7	–4.5	9	1.50(0.12)		

Notes. ^(a)Frequencies and spectroscopic parameters were extracted from the Jet Propulsion Laboratory molecular database (Pickett et al. 1998) for CH₃CHO and CH₃OCHO. Upper level energies refer to the ground state of each symmetry. ^(b)Integrated over the whole velocity emission range from the WideX spectra extracted at the CH₃CHO and CH₃OH emission peak (B1b; see text). In case of non-detections we report the 3σ limit. We note that the errors on the integrated areas I_{int} are propagated using the rms and the spectral resolution of the spectra. In addition, the uncertainty on the absolute flux scale due to calibration is $\leq 15\%$ (see Sect. 3). ^(c)The three lines at 96.74 GHz have been deblended using the 0.5 km s⁻¹ spectral resolution (see Sect. 3.2 and Fig. 4).

by Zucker et al. 2019), being the farthest away from the source, close to the B1 apex.

Within the context of the study of iCOMs, the advantages of observing L1157-B1 are that: (i) the observations are not polluted by the protostar emission, as it is located at a projected distance of 0.11 pc, and (ii) several previous studies have shown that grain-mantle species have been injected, due to shock-induced sputtering and shattering, into the gas phase (e.g. Bachiller et al. 2001; Arce et al. 2008; Codella et al. 2010; Fontani et al. 2014; Lefloch et al. 2017).

In the first SOLIS paper on L1157-B1 (CC17), we compared the spatial distribution of the acetaldehyde (CH₃CHO) line emission with that from formamide (NH₂CHO), finding a striking chemical stratification. This implies a NH₂CHO/CH₃COH abundance ratio which varies within the B1 structure: specifically, the formamide line emission is associated with older shocked gas with respect to acetaldehyde. Thanks to the comparison of the observations with astrochemical model predictions, we provided evidence that formamide is formed in L1157-B1 by gas-phase chemical reactions. However, we were unable to provide any indication on the formation route of acetaldehyde.

3. Observations

The L1157-B0/B1 shocks were observed at 3 mm with the IRAM NOEMA array during two tracks in October 2016 (7 antennas), and three tracks in January 2017 (8 antennas) using both the C and D configurations. The shortest and longest baselines are 24 and 304 m, respectively. These configurations allowed us to recover emission at scales (with a full efficiency) up to $\sim 14''$. The primary beam was $64''$. The phase centre is $\alpha(\text{J2000}) = 20^{\text{h}}39^{\text{m}}10^{\text{s}}.2$, $\delta(\text{J2000}) = +68^{\circ}01'10''.5$. The CH₃OH and

CH₃CHO lines listed in Table 1 were observed using the WideX backends, with a total spectral band of ~ 4 GHz, and a spectral resolution of 1.95 MHz (~ 6 km s⁻¹). Several CH₃OH lines were also observed with 80 MHz backends with a spectral resolution of 156 kHz (~ 0.48 km s⁻¹). Calibration was carried out following standard procedures using GILDAS-CLIC². The bandpass was calibrated on 3C454.3, while the absolute flux was fixed by observing MWC349, 2200+420, 2013+370, and 2007+659, the latter also being used to set the gains in phase and amplitude. The final uncertainty on the absolute flux scale is $\leq 15\%$. The phase rms was $\leq 50^\circ$, the typical precipitable water vapour (PWV) was from 2 to 20 mm, and the system temperatures were ~ 50 –100 K (D) and ~ 50 –250 K (C). The rms noise in the 1.95 MHz channels was 4–20 mJy beam⁻¹, depending on the frequency (see Table 1). Images were produced using natural weighting, and restored with a clean beam of $2''.97 \times 2''.26$ (PA = -155°).

4. Results

We imaged six methanol emission lines towards the L1157 blueshifted outflow, covering upper level excitations (E_u) from 7 to 28 K (see Table 1). Figure 1 (top) shows, as an example, the imaging of the CH₃OH(2_{1,2}–1_{1,1}) A line emission on top of the CO (1–0) image (Gueth et al. 1996), which accurately outlines the B0, B1, and B2 cavities opened by the precessing jet driven by L1157-mm, which is located at $\Delta\alpha = -25''$ and $\Delta\delta = +63''.5$ with respect to the centre of Fig. 1 map. Figure A.1 shows the images of the whole of the CH₃OH line emission reported in Table 1. In addition, we detected and imaged eight acetaldehyde lines ($E_u = 14$ –23 K), which are also reported in Table 1.

² <http://www.iram.fr/IRAMFR/GILDAS>

Figure 1 (bottom) shows the $\text{CH}_3\text{CHO}(5_{0,4}-4_{0,4})$ E map, while Fig. A.2 is for all the other CH_3CHO images.

Using the ASAI (Astrochemical Surveys At IRAM³; Lefloch et al. 2017, 2018) unbiased spectral survey obtained with the IRAM 30 m antenna, we evaluated the missing flux of the present NOEMA dataset. Comparison between the ASAI and NOEMA spectra indicates that a large fraction (about 60%) of the CH_3OH and CH_3CHO emission is filtered out, indicating that their emission is extended across structures larger than $\sim 14''$. Indeed, this matches the goal of the present project, which is the analysis of the chemical content of the small structures associated with multiple shocked regions, and without any contribution due to more extended emission. In Sects. 4.3 and 5.1 the present results will be compared with those derived by the analysis based on single-dish observations, showing there is no bias introduced by the missing flux.

4.1. CH_3OH and CH_3CHO spatial distribution

The present NOEMA CH_3OH data improve the images and spectra based on IRAM-PdBI observations of the methanol spectral pattern at 97.94 GHz performed in 1997 using five antennas, and with a synthesised beam of $\sim 5''$, as well as those of the $2_{1,1}-1_{1,0}$ A line, with a spectral resolution of 0.5 km s^{-1} (Benedettini et al. 2007). The eight CH_3CHO images also allow us to perform a deeper analysis with respect to the spatial distributions obtained by Codella et al. (2015), who stacked the $\text{CH}_3\text{CHO}(7_{0,7}-6_{0,6})$ E and A lines with an angular resolution of $\sim 2''.5$.

The present results are consistent with the previous observations that show that the CH_3OH and CH_3CHO spatial distributions are in good agreement; both trace the B0 eastern molecular cavity as well as the northern arch-like structure containing B1a-e-f-b. In particular, the peak emission of both species is clearly associated with the B1b clump. In other words, although emission is observed also towards B1c, the bulk of the CH_3OH and CH_3CHO emission does not come from the southern B1 apex (Codella et al. 2017). Instead, they preferentially trace the region associated with the youngest B1 shock occurring next to the B1a position, where HDCO, a selective tracer of dust mantle release, has been observed (Fontani et al. 2014), as well as high-velocity SiO emission (Gueth et al. 1998; Podio et al. 2016), locating the likely current sputtering of the dust refractory cores due to the shock. These findings are not surprising for methanol, which is efficiently produced on the grain surfaces by CO hydrogenation. On the other hand, they indicate that acetaldehyde is either formed on the mantles as well or that it is quickly (in less than $\sim 1000 \text{ yr}$) produced in the gas phase using simpler mantle products (as suggested by Codella et al. 2015). This topic is further discussed in Sect. 5.2.

4.2. Line spectra

Figure 2 shows the spectra extracted from the CH_3OH and CH_3CHO emission peaks at the B1b position. In addition to methanol and acetaldehyde, in B1b we observe emission at 99.3 GHz due to another iCOM, dimethyl ether (CH_3OCH_3). In particular, the detected line is due to the $4_{1,4}-3_{0,3}$ transitions of the EA, AE, EE, and AA isomers, blended at the $\sim 2 \text{ MHz}$ spectral resolution (see Fig. 2). Finally, we detect several S-bearing species (CS, SO, and OCS isotopologues), which will be analysed in a forthcoming paper on sulphur chemistry (Feng et al., in prep.).

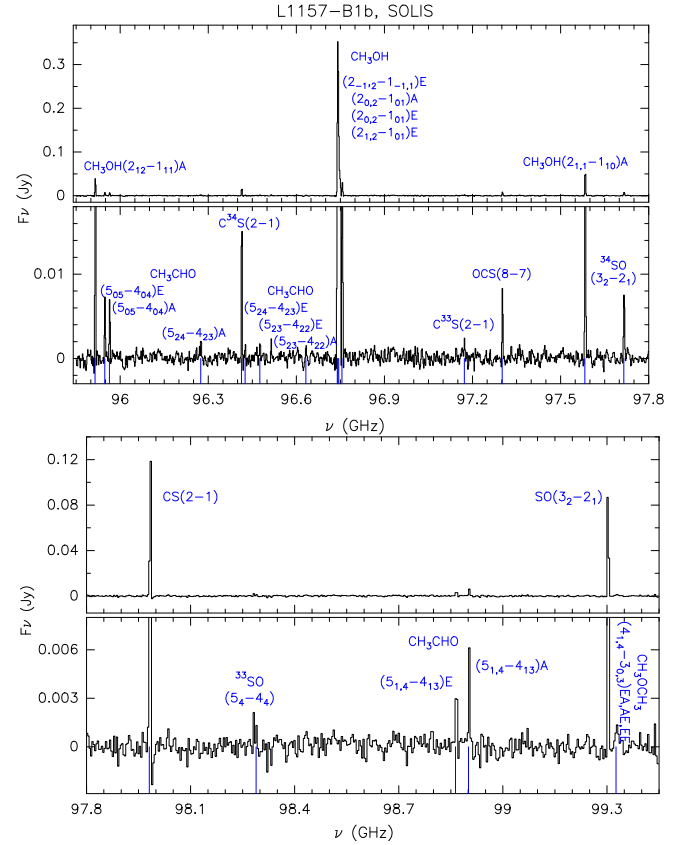


Fig. 2. Upper and lower panels: IRAM-WideX spectra (in F_ν scale) extracted at the L1157-B1b blueshifted position (see Fig. 1 and text). The spectral resolution is 1.95 MHz (see Sect. 3). A zoom-in in intensity is shown so that the weakest lines can be seen. The transitions producing the emission lines are labelled in blue, with the corresponding frequencies marked by vertical blue lines. The detected CH_3OH and CH_3CHO lines (analysed here) are reported in Table 1.

For each detected line, in addition to the spectroscopic parameters, Table 1 reports the intensity integrated over the whole velocity emission range (down to velocities blueshifted by 15 km s^{-1}). The 2 MHz spectral resolution of the WideX backend does not allow us to investigate the iCOMs kinematics. However, we can confirm that all the spectra are blueshifted, in agreement with all the molecular lines so far observed towards L1157-B1 (e.g. Bachiller et al. 2001; Lefloch et al. 2017). In addition, the CH_3OH spectral pattern at 96.7 GHz (see Table 1) has been observed with a 0.48 MHz (0.16 km s^{-1}) spectral resolution. Figure 3 reports the spectra as observed towards the B0e and B1b clumps: four transitions of the $2_{k,k}-1_{k,k}$ pattern are reported, with the corresponding frequencies marked by small vertical red lines. The spectra are centred at 96741.38 MHz , which is the frequency of the $2_{0,2}-1_{0,1}$ A line. The methanol lines extracted from the B0e position have a peak velocity of $\sim +0.5 \text{ km s}^{-1}$ ($V_{\text{LSR}} = +2.6 \text{ km s}^{-1}$), and are about $3-4 \text{ km s}^{-1}$ in width (FWHM), with evidence of bluer wings. The B1b spectrum is definitely more complex showing in addition a secondary peak at about -4 km s^{-1} . This high-velocity peak was tentatively detected by Benedettini et al. (2013) using the $\text{CH}_3\text{OH}(3_{k,k}-2_{k,k})$ emission at 2 mm . In the present dataset, a clear Gaussian-like secondary peak is emerging, definitely offset in velocity from the mean peak. These findings confirm that the northern part of the B1 structure is associated with young shocked gas, here traced by methanol molecules injected into the gas phase at high velocities.

³ www.oan.es/asai

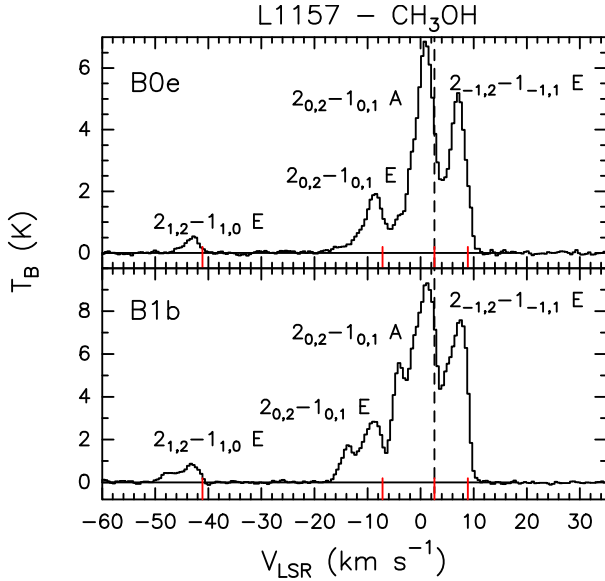


Fig. 3. CH₃OH emission spectra at 96.7 GHz (in T_B scale) extracted at the B0e and B1b positions (see Fig. 1 and text). The spectral resolution is 0.48 MHz, corresponding to 0.16 km s⁻¹. The transitions producing the emission lines (see Table 1) are reported, with the corresponding frequencies marked by small vertical red lines. The spectra are centred at the frequency of the $2_{0,2}-1_{0,1}$ A transition: 96 741.38 MHz. The dashed vertical line stands for the systemic velocity: +2.6 km s⁻¹ (e.g. Bachiller et al. 2001).

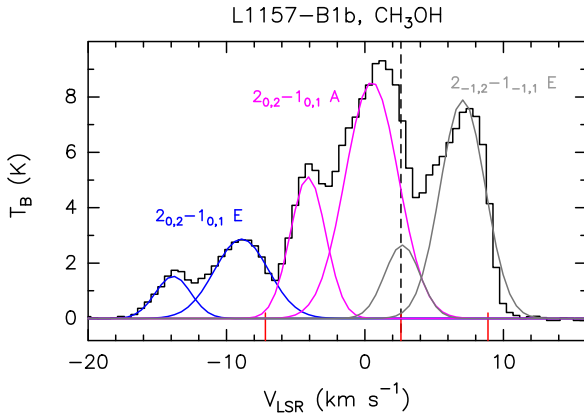


Fig. 4. Spectral fit of the $2_{-1,2}-1_{-1,1}$ E (grey lines), $2_{0,2}-1_{0,1}$ A (magenta), and $2_{0,2}-1_{0,1}$ E (blue) triplets as observed towards B1b using the NOEMA narrow-band spectrometer. The spectra are centred at the frequency of the $2_{0,2}-1_{0,1}$ A transition: 96 741.38 MHz. The transitions producing the emission lines (see Table 1) are reported, with the corresponding frequencies marked by small vertical red lines. The dashed vertical line is for the systemic velocity: +2.6 km s⁻¹ (e.g. Bachiller et al. 2001). Each transition shows a main peak at +0.5 km s⁻¹ plus a secondary peak further blueshifted by ~ 5 km s⁻¹.

Figure 4 shows the spectral fit of the $2_{-1,2}-1_{-1,1}$ E, $2_{0,2}-1_{0,1}$ A, and $2_{0,2}-1_{0,1}$ E triplets as observed towards B1b. The velocity scale is based on the rest frequency of the brighter line, $2_{0,2}-1_{0,1}$ A, at 96 741.38 MHz. In order to perform the fit, we assumed that the three low-excitation methanol transitions are associated with the same spectral profile. As a consequence, the rest frequencies of the transitions give us the expected velocity shifts among the three lines (see the red small vertical segments in Fig. 4). In addition, both $2_{0,2}-1_{0,1}$ E and A lines (blue and magenta) show a secondary peak (3.0 km s⁻¹ broad)

offset by 5 km s⁻¹ with respect the main one (which has a line width of 4.6 km s⁻¹). In conclusion, we fixed the position and the line width of the secondary peak expected for the $2_{-1,2}-1_{-1,1}$ E line (grey). The overall fit is satisfactory, with a residual with an rms of 480 mK.

4.3. Large Velocity Gradient analysis and rotational diagrams: column densities and abundances

Given the lack of CH₃CHO collisional rates in the literature, we only analysed the CH₃OH emission with the non-local thermodynamic equilibrium (NLTE) large velocity gradient (LVG) approach using the model described in Ceccarelli et al. (2003). We used the CH₃OH-H₂ collisional coefficients computed by Rabli & Flower (2010) and provided by the BASECOL database (Dubernet et al. 2013). In the computations, we assumed a H₂ ortho-to-para ratio of one (Nisini et al. 2010). We run grids of models varying the kinetic temperature (T_{kin}) from 50 to 300 K, the volume density (n_{H_2}) from 10⁴ to 10⁸ cm⁻³, and the methanol column density ($N(\text{CH}_3\text{OH})$) from 10¹⁴ to 10¹⁸ cm⁻². The best fit is obtained by the minimum χ^2 in the three parameters. The errors on the observed line intensities have been obtained by propagating the spectral rms with the uncertainties due to calibration (15%).

Figure 5 shows: (i) the χ^2 contour plot in the $T_{\text{kin}}-n_{\text{H}_2}$ plane obtained with the best fit of the methanol column density $N(\text{CH}_3\text{OH}) = 5 \times 10^{15}$ cm⁻² ($3-8 \times 10^{15}$ cm⁻², including uncertainties), (ii) the $\chi^2-N(\text{CH}_3\text{OH})$ distribution, where it is possible to identify the minimum χ^2 value (2.4), and (iii) the ratio between the measured and predicted intensity of the observed lines. Overall, the best fit is very good, with a reduced χ^2 equal to 1.2, which for two degrees of freedom corresponds to a probability of 30% to exceeding χ^2 (i.e. approximately the standard 1 σ). The methanol line opacities are predicted to be less than 0.1. The volume density is well constrained, $n_{\text{H}_2} = 4 \times 10^5$ cm⁻³ ($2-10 \times 10^5$ cm⁻³), in agreement with the high densities ($\geq 10^5$ cm⁻³) inferred toward the molecular cavities by Gómez-Ruiz et al. (2015) from a CS multiline analysis. The kinetic temperature is also well constrained, $T_{\text{kin}} = 90$ K (70–130 K), again in agreement with the temperatures of the B1 cavity as derived by Lefloch et al. (2012) using a CO multiline single-dish analysis. We note that a recent LVG analysis of methanol emission as observed using single-dish towards a large number of protostellar jet-driven shocks (Holdship et al. 2019) shows volume densities in the 10⁵–10⁶ cm⁻³ range and kinetic temperatures between 20 and 60 K, values consistent with what is found in L1157-B1.

The analysis described immediately above allows us to constrain the density and temperature of the methanol-emitting gas. However, in order to derive the relative abundances of methanol and acetaldehyde, we carried out a rotational diagram (RD) analysis for both molecules, where a LTE population and optically thin lines are assumed (in agreement with the NLTE LVG analysis of methanol described above). Under these assumptions, for a given molecule, the relative population distribution of all the energy levels is described by a Boltzmann temperature, that is the rotational temperature T_{rot} . As demonstrated by Lefloch et al. (2012), for example, by analysing CO emission from extended outflowing gas in L1157, the column density obtained with RD gives results that are in relatively good agreement with a NLTE LVG analysis, usually within a factor two, as is also the case for our analysis.

Figure 6 shows the RD of CH₃OH, which provides a column density of $(1.3 \pm 0.3) \times 10^{16}$ cm⁻², close to what was found using the LVG method. The rotational temperature is (10.0 ± 1.1) K, in

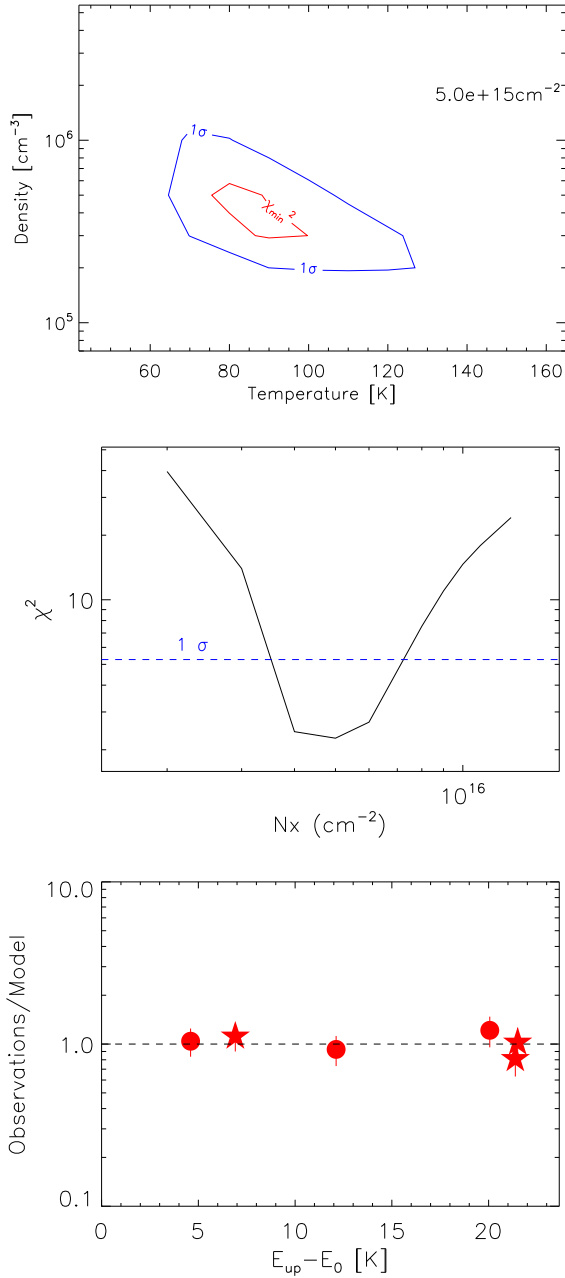


Fig. 5. *Upper panel:* density-temperature contour plot of χ^2 obtained considering the non-LTE LVG-model-predicted and observed intensity of all the A and E CH₃OH emission lines detected (beam averaged) towards the B1b clump. The best fit is obtained with $N(\text{CH}_3\text{OH}) = 5 \times 10^{15} \text{ cm}^{-2}$, $T_{\text{kin}} = 90 \text{ K}$, and $n_{\text{H}_2} = 4 \times 10^5 \text{ cm}^{-3}$. *Middle panel:* χ^2 - $N(\text{CH}_3\text{OH})$ distribution, identifying the minimum χ^2 value (2.4). *Lower panel:* ratio between observations and model predictions of the CH₃OH A (circles) and E (stars) line intensities as a function of the upper level energy of the lines.

agreement with what was found using previous interferometric and single dish measurements of CH₃OH lines in the same excitation range (12 K; [Bachiller et al. 1995](#); [Benedettini et al. 2007](#); [Codella et al. 2010](#)). Thanks to the many CH₃CHO detected lines (covering an upper level energy E_u range similar to that traced by the methanol lines) we can carry out the RD analysis also for acetaldehyde, obtaining, as for CH₃OH, a low rotational temperature, $(8 \pm 1) \text{ K}$, and a total column density $N(\text{CH}_3\text{CHO}) = (7 \pm 3) \times 10^{13} \text{ cm}^{-2}$. [Lefloch et al. \(2017\)](#) report for CH₃CHO a

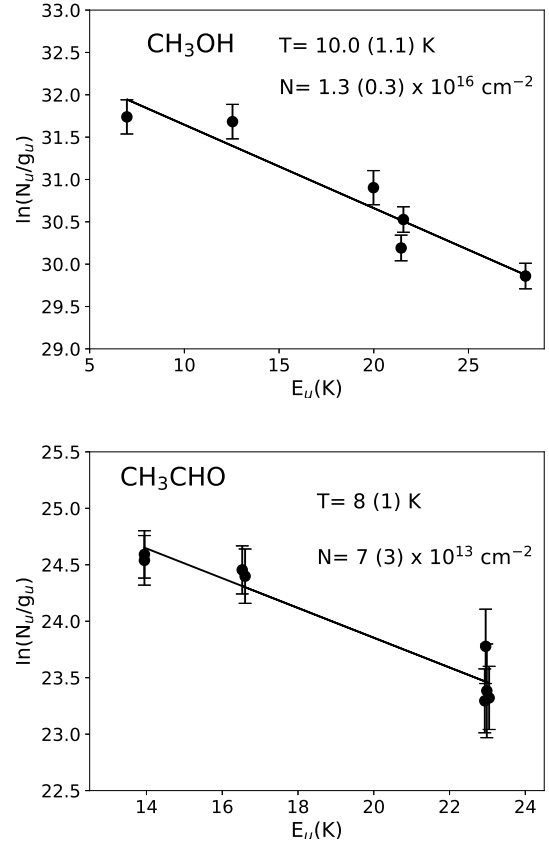


Fig. 6. Rotational diagrams for CH₃OH (*upper panel*), and CH₃CHO (*lower panel*) derived using the emission lines observed towards B1b (see Table 1 and Fig. 3). The parameters N_u , g_u , and E_{up} are the column density, the degeneracy, and the energy (with respect to the ground state of each symmetry), respectively, of the upper level. The derived values of the rotational temperature are reported in the panels.

rotational temperature of 17 K, but this was measured including transitions with E_u up to 94 K. Figure 7 shows the spatial distribution of the CH₃CHO (*upper panel*) and CH₃OH (*lower panel*) rotational temperature in the B0 and B1 regions of the L1157 outflow. We note that the CH₃CHO rotational temperature map covers a smaller portion of the L1157 region due to the weakness of the acetaldehyde lines with $E_u = 23 \text{ K}$. Considering the uncertainties (B0e and B1e: $\sim 6 \text{ K}$; B1b: $\sim 3 \text{ K}$) there is no significant difference between the rotational temperature of the two species or spatial trends throughout the L1157-B1 structure.

Finally, we computed the abundances of the various species using the values of their respective column densities divided by the H₂ column density, which was derived from the CO column density in the B1 molecular cavity ($N_{\text{CO}} = 2 \times 10^{17} \text{ cm}^{-2}$; [Lefloch et al. \(2012\)](#) assuming the standard $[\text{CO}]/[\text{H}] = 5 \times 10^{-5}$ value. In this way, we obtain methanol and acetaldehyde abundances with respect to H equal to 6.5×10^{-6} ($1.2\text{--}12 \times 10^{-6}$, considered the uncertainties) and 3.5×10^{-8} ($2\text{--}5 \times 10^{-6}$), respectively, in good agreement with that derived by CC17 for acetaldehyde ($1\text{--}3 \times 10^{-8}$) using a coarser angular resolution ($5''\text{--}6''$).

5. Discussion

5.1. The CH₃OH to CH₃CHO abundance ratio

[Codella et al. \(2017\)](#) showed that the spatial distribution of the measured abundance ratio of iCOMs can lead to very stringent

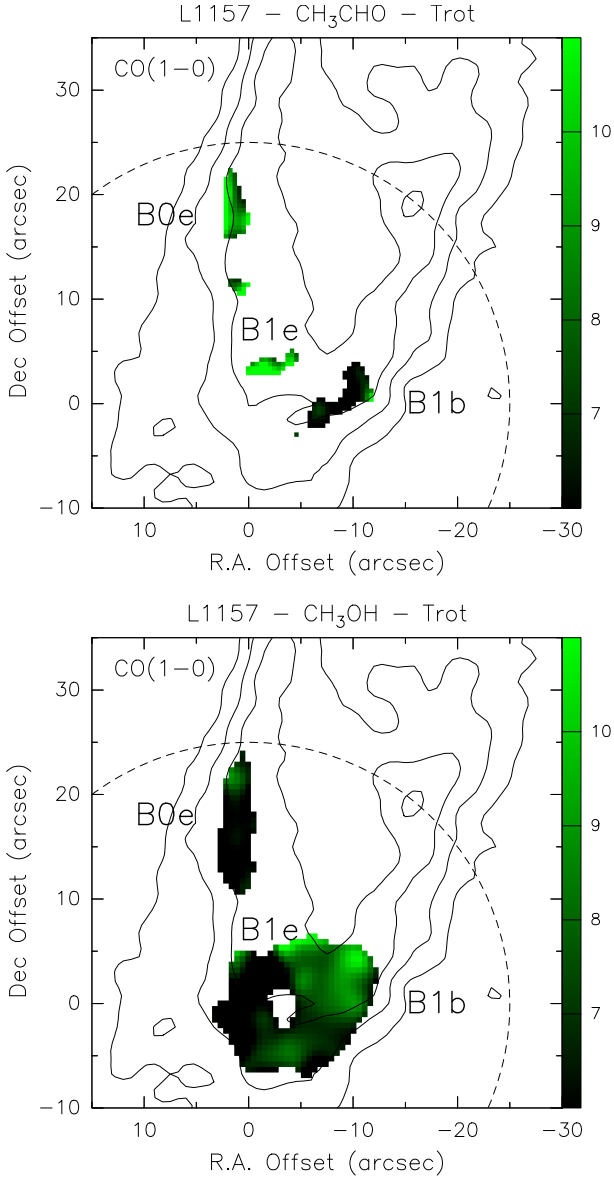


Fig. 7. CH_3CHO (upper panel) and CH_3OH (lower panel) rotational temperature (T_{rot}) map (in K scale) of the B0 and B1 region of the L1157 outflow (derived from the NOEMA images, where the emission is at least 3σ), overlaid on the CO (black contours; Gueth et al. 1996) emission map. Uncertainties on T_{rot} are ~ 3 K towards B1b, and ~ 6 K for B1e and B0. Symbols are as in Fig. 1.

constraints on how they are formed and destroyed. In particular, these latter authors showed that the formamide to acetaldehyde abundance ratio towards the L1157-B1 shock demonstrates a gas-phase formation route of formamide (e.g. Skouteris et al. 2017). Further analysis of the abundance ratio of acetaldehyde with respect to methanol is therefore worthwhile, which can be safely assumed to form on dust mantles and then injected into the gas phase. Based on the interferometric maps showing that both species are populating the same gas portion, the ratio between their column densities can be used to derive their abundance ratio. The ratio of CH_3OH to CH_3CHO is about 1.5×10^2 and 1.9×10^2 as measured towards the B0e and B1b peaks, respectively, and it varies by less than a factor of three throughout the whole B0–B1 structure, as shown in Fig. 8. Considering that the uncertainty on these measurements is about 30%, there is no evidence

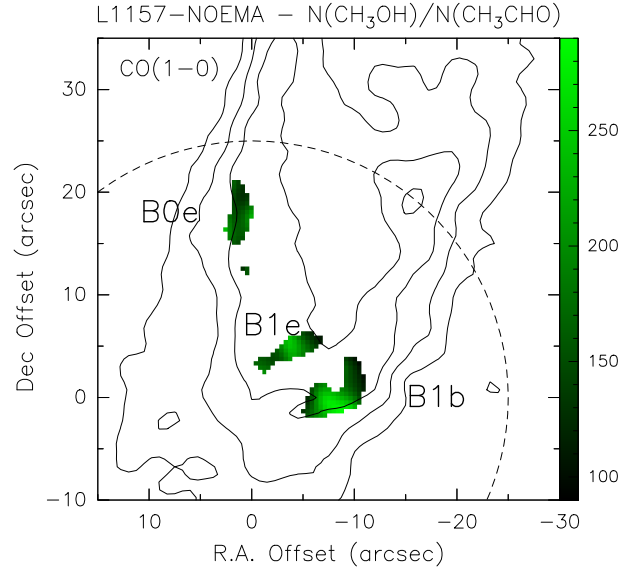


Fig. 8. Map of the ratio between the CH_3OH and CH_3CHO column density of the B0 and B1 region of the L1157 outflow (derived from the NOEMA images, where the emission is at least 3σ), overlaid on the CO (black contours; Gueth et al. 1996) emission map. Uncertainties are $\sim 30\%$ (i.e. 30–100 in absolute values). Symbols are as in Fig. 1.

for significant variation of this latter ratio in the observed region. Furthermore, these measurements are consistent with those derived using the ASAI IRAM 30 m single-dish survey ($1\text{--}3 \times 10^2$; Codella et al. 2015; Lefloch et al. 2017). We note that Holdship et al. (2019) measured $\text{CH}_3\text{OH}/\text{CH}_3\text{CHO}$ abundance ratios towards a sample of shocks using the IRAM 30 m antenna, finding higher values (by a factor of six on average): $0.4\text{--}1.3 \times 10^3$; high-spatial-resolution surveys are needed to confirm these values.

It is interesting to check whether the $\text{CH}_3\text{OH}/\text{CH}_3\text{CHO}$ abundance ratios derived for L1157-B0 and L1157-B1 are different with respect to measurements towards known Class 0 hot corinos, which rather trace the inner 100 au from the protostar where the temperature is high enough (at least 100 K) to thermally evaporate the dust-mantle products into the gas phase, or where grain mantles are sputtered by shocks (see below). If we only include in the comparison those sources imaged with high-spatial-resolution interferometric observations (to be sure of a proper comparison between the CH_3OH and CH_3CHO spatial distributions), the sample size is limited (see Table 2): HH212-mm (Lee et al. 2017, 2019; Bianchi et al. 2017; Codella et al. 2018), IRAS 16293-2422B (Jørgensen et al. 2016, 2018), NGC 1333-IRAS 4A2 (López-Sepulcre et al. 2017), Barnard 1b-S (Marcelino et al. 2018), and L483 (Jacobsen et al. 2018). Table 2 shows that the $\text{CH}_3\text{OH}/\text{CH}_3\text{CHO}$ ratio of the hot-corinos varies from ~ 90 to ~ 500 , that is, by less than a factor of six. The L1157 values fall inside this range, suggesting that the chemistry at work in the protostellar shocks, at least in the CH_3OH and CH_3CHO context, is similar to that ruling in the inner 100 au of the protostellar region. Of course, this is speculation based on one shocked region, and verification is required: (i) by increasing similar measurements towards different outflow shocks, and (ii) by investigating whether the chemical enrichment around a protostar is ruled by thermal evaporation or sputtering due to accretion shocks (see e.g. Sakai et al. 2014a,b; Lee et al. 2017, 2019; Codella et al. 2018).

Table 2. Comparison of the $\text{CH}_3\text{OH}/\text{CH}_3\text{CHO}$ abundance ratio as derived towards L1157-B1 with those measured towards Class 0 hot corinos using interferometric observations.

Object	$\text{CH}_3\text{OH}/\text{CH}_3\text{CHO}$	References
L1157-B0e	1.5×10^2	1
L1157-B1b	1.9×10^2	1
HH212-mm	$0.9\text{--}3 \times 10^2$	2, 3, 4, 5
IRAS 16293–2422B	1×10^2	6, 7
NGC 1333–IRAS 4A2	$3\text{--}6 \times 10^2$	8, 9
Barnard 1b–S	5×10^2	10
L483	2×10^2	11

References. 1. Present work; 2. Lee et al. (2017); 3. Lee et al. (2019); 4. Bianchi et al. (2017); 5. Codella et al. (2018); 6. Jørgensen et al. (2016); 7. Kristensen et al. (2017); 8. Taquet et al. (2015); 9. López-Sepulcre et al. (2017); 10. Marcelino et al. (2018); 11. Jacobsen et al. (2018).

5.2. Chemical modelling

The overlap between the spatial distribution of the CH_3OH and CH_3CHO emission suggests that either acetaldehyde or a parent molecule was previously formed on the grain mantles, such as methanol for example, and was injected into the gas because of the passage of the shock. In addition, given that the $\text{CH}_3\text{OH}/\text{CH}_3\text{CHO}$ abundance ratio is similar in the B0 and B1 knots despite them having a different kinematical age of less than about 1000 yr, if CH_3CHO is synthesised in the gas phases, the process must be relatively fast. We note that recently, Burkhardt et al. (2019) modelled the shocked chemistry in L1157-B1 finding that CH_3CHO has a post-shock chemistry due to sputtering, and additionally that a significant amount of its formation is due to gas-phase processes before 10^3 yr. The goal of this section is to explore whether gas-phase reactions are indeed fast enough to be able to reproduce the observations.

To this end, we used the same astrochemical model adopted in CC17 to analyse the formamide and acetaldehyde emission towards L1157-B1. The full details of the model can be found in CC17. Briefly, we used the time-dependent model gas-phase code MyNahoon, and adopted a two-step procedure: (1) we first compute the chemical composition of a standard molecular cloud at 10 K with a H_2 density of $2 \times 10^4 \text{ cm}^{-3}$ and initial elemental abundances as in Agundez & Wakelam (2013, their Table 3). (2) We then successively increase the gas temperature, density, and gaseous abundance of grain-mantle species to simulate the shock passage. The gas temperature and H_2 density are taken as those found by the NLTE LVG modelling, namely 90 K and $4 \times 10^5 \text{ cm}^{-3}$, respectively. For the abundances of the species injected into the gas phase at the moment of the passage of the shock, when possible, we assumed abundances (with respect to H nuclei) similar to those measured by IR observations of the interstellar dust ices (Boogert et al. 2015): CO_2 (3×10^{-5}) and H_2O (2×10^{-4} ; see also Busquet et al. 2014). Otherwise, we used the values constrained by previous studies on the L1157-B1 chemistry: OCS (2×10^{-6} ; Podio et al. 2014), NH_3 (2×10^{-5} ; Tafalla & Bachiller 1995). We varied the injected CH_3CH_2 abundance in the gas-phase acetaldehyde formation model so as to obtain the observed acetaldehyde abundance. We found it necessary to inject $\text{CH}_3\text{CH}_2/\text{H} = 8 \times 10^{-8}$. Similarly, in the grain-surface acetaldehyde formation model, we modified the acetaldehyde abundances to fit the measured value in B1b (see

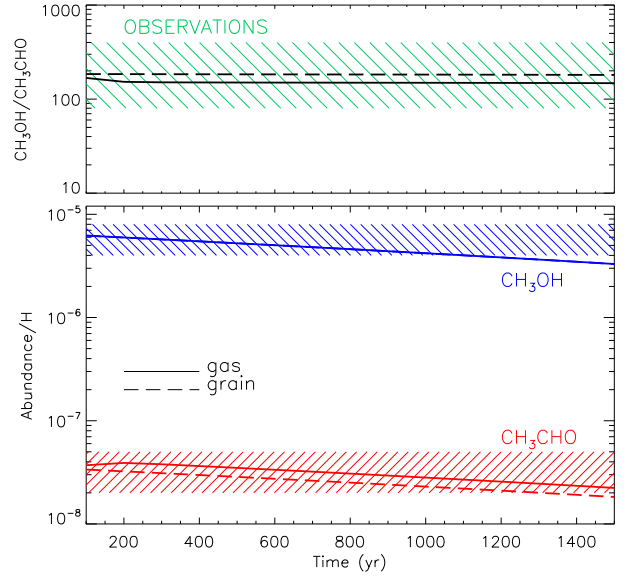


Fig. 9. Model predictions of methanol and acetaldehyde after a shock passage. *Upper panel:* $\text{CH}_3\text{OH}/\text{CH}_3\text{CHO}$ abundance ratio: solid line refers to a model where acetaldehyde is synthesised in the gas phase, whereas the dashed line refers to predictions assuming that CH_3OH is injected into the gas phase directly from the grain mantles. The hatched green zone is defined by the value (including uncertainty) measured towards the B1b peak, with a kinematical age of ≈ 1100 yr (Podio et al. 2016, revised according to the new distance of 352 pc by Zucker et al. 2019). *Lower panel:* acetaldehyde (CH_3CHO , red) and methanol (CH_3OH , blue) abundances with respect to H nuclei as a function of time from the passage of the shock. The hatched blue and red regions show the measurements towards B1b (see text).

Sect. 4.3), namely we assumed $\text{CH}_3\text{CHO}/\text{H} = 3.5 \times 10^{-8}$. Finally, in both models, methanol is assumed to be a grain-surface product and the abundance injected in the gas phase is equal to that observed, namely 6.5×10^{-6} (see Sect. 4.3).

As in CC2017, we run both the case where acetaldehyde is directly injected into the gas phase from the grain mantles and the case in which it is formed in the gas phase, mainly by the reaction of ethyl radical (CH_3CH_2) with atomic oxygen: $\text{CH}_3\text{CH}_2 + \text{O} \rightarrow \text{CH}_3\text{CHO} + \text{H}$, as suggested by Charnley et al. (2004). With respect to the chemical network used by CC2017, we included the recent theoretical study by Gao et al. (2018) on the destruction of methanol by the reaction with OH.

Figure 9 reports the model predictions as a function of time after the shock passage compared with the values measured towards B1b. As written above, if acetaldehyde is directly sputtered from the grain mantles, in order to fit the observations, its injected abundance must be equal to 3.5×10^{-8} . On the other hand, if acetaldehyde is formed via gas-phase reactions, the injected ethyl radical abundance must be 8×10^{-8} . As shown in the figure, the gas phase reactions are relatively fast at synthesising acetaldehyde: within 300 yr the acetaldehyde abundance reaches its peak, after which it slowly decreases because of the destruction from ions in the gas phase, on a timescale larger than about 2000 yr. Therefore, it is impossible to elucidate the formation route of acetaldehyde in L1157-B1: a similar study towards an even younger shock might be able to shed light on the question. Alternatively, a less dense shock region, or one that was less irradiated by cosmic rays could be good site for better constraining the acetaldehyde formation route, as the chemical evolution would be slowed down in those cases.

As a final remark, we would like to highlight the fact that, although relatively simple, the modelling described above and used in this work catches the basic processes at work after the passage of a shock and has proven to be relatively reliable for reproducing the observations (see e.g. Podio et al. 2014 and C2017). The most drastic shortcoming of our model compared to more accurate and/or sophisticated shock models is its lack of a higher temperature (up to 1000 K) over a short (a few hundred years at most) period (see e.g. Viti et al. 2011). This could be important in the presence of species whose abundances depend on reactions with activation barriers that make them inefficient at low (≤ 100 K) temperatures but efficient at higher ones. In the specific case modelled here, we verified that no similar reactions are present and that a high-temperature (1000 K) phase does not impact our results.

6. Conclusions

In the context of the IRAM NOEMA SOLIS Large Program we imaged six methanol and eight acetaldehyde emission lines toward the B0 and B1 shocks along the L1157 blueshifted outflow. The measured abundances relative to H nuclei are 6.5×10^{-6} (methanol) and 3.5×10^{-8} (acetaldehyde). The CH_3OH and CH_3CHO spatial distributions overlap well, tracing the earliest shocked regions. Comparison with astrochemical model predictions shows that CH_3CHO is either formed on the dust mantles or is quickly formed in the gas phase using simpler mantle products. Our modelling demonstrates that two species emitting lines in the same region is not enough to guarantee that they are grain-surface products: gas-phase reactions triggered by different species from the grain mantles could also display the same spatial behaviour. Detailed studies are required in order to discriminate between grain-surface and gas-phase chemistry as the major process leading to the observed species and to decipher the timescale on which this formation occurs.

For acetaldehyde in particular, at present, it is not possible to measure the abundance of one of the two parent species (ethyl radical) possibly synthesising acetaldehyde in the gas phase. The measurement of the second parent species, atomic oxygen, is challenging, as its fine structure line at $63 \mu\text{m}$ lies in a spectral window that is totally blocked by the terrestrial atmosphere. The SOFIA telescope could overcome this problem (Kristensen et al. 2017). On the other hand, although the frequencies (and spectroscopic parameters) of ethyl radicals are unknown at present, they would be a very welcome target of specific spectroscopic studies. It would then very likely be possible to search for ethyl radicals in regions where acetaldehyde is detected, and thus decipher whether or not the proposed gas-phase route is correct.

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References

Agundez, M., & Wakelam, V. 2013, *Chem. Rev.*, **113**, 8710
 Arce, H. G., Santiago-García, J., Jørgensen, J. K., Tafalla, M., & Bachiller, R. 2008, *ApJ*, **681**, L21

Bachiller, R., Liechti, S., Walmsley, C. M., & Colomer, F. 1995, *A&A*, **295**, L51
 Bachiller, R., Pérez Gutiérrez, M., Kumar, M. S. N., & Tafalla, M. 2001, *A&A*, **372**, 899
 Balucani, N., Ceccarelli, C., & Taquet, V. 2015, *MNRAS*, **449**, L16
 Burkhardt, A. M., Shingledecker, C. N., Le Gal, R., et al. 2019, *ApJ*, **881**, 32
 Belloche, A., Garrod, R. T., Müller, H. S. P., & Menten, K. M. 2014, *Science*, **345**, 1584
 Benedettini, M., Viti, S., Codella, C., et al. 2007, *MNRAS*, **381**, 1127
 Benedettini, M., Viti, S., Codella, C., et al. 2013, *MNRAS*, **436**, 179
 Bianchi, E., Codella, C., Ceccarelli, C., et al. 2017, *A&A*, **606**, L7
 Boogert, A. C. A., Gerakines, P. A., & Whittet, D. C. B. 2015, *ARA&A*, **53**, 541
 Busquet, G., Lefloch, B., Benedettini, M., et al. 2014, *A&A*, **561**, A120
 Caselli, P., & Ceccarelli, C. 2012, *A&ARv*, **20**, 56
 Caselli, P., Hartquist, T. W., & Havnes, O. 1997, *A&A*, **322**, 286
 Ceccarelli, C., Maret, S., Tielens, A. G. G. M., Castets, A., & Caux, E. 2003, *A&A*, **410**, 587
 Ceccarelli, C., Caselli, P., Fontani, F., et al. 2017, *ApJ*, **850**, 176
 Charnley, S. B. 2004, *Adv. Space Res.*, **33**, 23
 Charnley, S. B., Tielens, A. G. G. M., & Millar, T. J. 1992, *ApJ*, **399**, L71
 Codella, C., Benedettini, M., Beltrán, M. T., et al. 2009, *A&A*, **507**, L25
 Codella, C., Lefloch, B., Ceccarelli, C., et al. 2010, *A&A*, **518**, L112
 Codella, C., Fontani, F., Ceccarelli, C., et al. 2015, *MNRAS*, **449**, L11
 Codella, C., Ceccarelli, C., Caselli, P., et al. 2017, *A&A*, **605**, L3
 Codella, C., Bianchi, E., Tabone, B., et al. 2018, *A&A*, **617**, A10
 Dubernet, M. L., Alexander, M. H., Ba, Y. A., et al. 2013, *A&A*, **553**, A50
 Flower, D. R., & Pineau des Forêts, G. 1994, *MNRAS*, **268**, 724
 Fontani, F., Codella, C., Ceccarelli, C., et al. 2014, *ApJ*, **788**, L43
 Gao, L. G., Zheng, J., Fernández-Ramos, A., Truhlar, D. G., & Xu, X. 2018, *J. Am. Chem. Soc.*, **140**, 8
 Garrod, R. T., & Herbst, E. 2006, *A&A*, **457**, 927
 Garrod, R. T., Widicus Weaver, S. L., & Herbst, E. 2008, *ApJ*, **682**, 283
 Gómez-Ruiz, A. I., Codella, C., Lefloch, B., et al. 2015, *MNRAS*, **446**, 3346
 Gueth, F., Guilloteau, S., & Bachiller, R. 1996, *A&A*, **307**, 891
 Gueth, F., Guilloteau, S., & Bachiller, R. 1998, *A&A*, **333**, 287
 Gusdorf, A., Cabrit, S., Flower, D. R., & Pineau des Forêts, G. 2008a, *A&A*, **482**, 809
 Gusdorf, A., Pineau des Forêts, G., Cabrit, S., & Flower, D. R. 2008b, *A&A*, **490**, 695
 Holdship, J., Viti, S., Codella, C., et al., 2019, *ApJ*, **880**, 138
 Herbst, E., & van Dishoeck, E. F. 2009, *ARA&A*, **47**, 427
 Jacobsen, S. K., Jørgensen, J. K., Di Francesco, et al. 2018, *A&A*, **629**, A29
 Jørgensen, J. K., van der Wiel, M. H. D., Coutens, A., et al. 2016, *A&A*, **595**, A117
 Jørgensen, J. K., Müller, H. S. P., Calcutt, H., et al. 2018, *A&A*, **620**, A170
 Kristensen, L. E., Gusdorf, A., Mottram, J. C., et al. 2017, *A&A*, **601**, L4
 Lee, C.-F., Li, Z.-Y., Ho, P.-T.-P., et al. 2017, *ApJ*, **843**, L27
 Lee, C.-F., Codella, C., Li, Z.-Y., & Liu, S.-Y. 2019, *ApJ*, **873**, L63
 Lefloch, B., Cabrit, S., Busquet, G., et al. 2012, *ApJ*, **757**, L25
 Lefloch, B., Ceccarelli, C., Codella, C., et al. 2017, *MNRAS*, **469**, L73
 Lefloch, B., Bachiller, R., Ceccarelli, C., et al. 2018, *MNRAS*, **477**, 4792
 López-Sepulcre, A., Sakai, N., Neri, R., et al. 2017, *A&A*, **606**, A121
 Marcelino, N., Gerin, M., Cernicharo, J., et al. 2018, *A&A*, **620**, A80
 Mendoza, E., Lefloch, B., López-Sepulcre, A., et al. 2014, *MNRAS*, **445**, 151
 Nisini, B., Benedettini, B., Codella, C., et al. 2010, *A&A*, **518**, L120
 Pickett, H. M., Poynter, R. L., Cohen, E. A., et al. 1998, *J. Quant. Spectr. Rad. Transf.*, **60**, 883
 Podio, L., Lefloch, B., Ceccarelli, C., Codella, C., & Bachiller, R. 2014, *A&A*, **565**, A64
 Podio, L., Codella, C., Gueth, F., et al. 2016, *A&A*, **593**, L4
 Rabli, D., & Flower, D. R. 2010, *MNRAS*, **406**, 95
 Rimola, A., Taquet, V., Ugliengo, P., Balucani, N., & Ceccarelli, C. 2014, *A&A*, **572**, A70
 Ruaud, M., Loison, J. C., Hickson, K. M., et al. 2015, *MNRAS*, **447**, 4004
 Sakai, N., Sakai, T., Hirota, T., et al. 2014a, *Nature*, **507**, 78
 Sakai, N., Oya, Y., Sakai, T., et al. 2014b, *ApJ*, **791**, L38
 Schilke, P., Walmsley, C. M., Pineau des Forêts, G., & Flower, D. R. 1997, *A&A*, **321**, 293
 Skouteris, D., Vazart, F., Ceccarelli, C., et al. 2017, *MNRAS*, **468**, L1
 Skouteris, D., Balucani, N., Ceccarelli, C., et al. 2018, *ApJ*, **854**, 135
 Tafalla, M., & Bachiller, R. 1995, *ApJ*, **443**, L37
 Taquet, V., López-Sepulcre, A., Ceccarelli, C., et al. 2015, *ApJ*, **804**, 81
 Tielens, A. G. G. M., & Hagen, W. 1982, *A&A*, **114**, 245
 Vasyunin, A. I., Caselli, P., Dulieu, F., & Jiménez-Serra, I. 2017, *ApJ*, **842**, 33
 Viti, S., Jiménez-Serra, I., Yates, J., et al. 2011, *ApJ*, **740**, L3
 Watanabe, N., & Kouchi, A. 2002, *ApJ*, **571**, L173
 Zucker, C., Speagle, J. S., Schlafly, E. F., et al. 2019, *ApJ*, **879**, 125

Appendix A: The NOEMA-SOLIS CH₃OH and CH₃CHO line maps

Figures A.1 and A.2 show the images of the detected CH₃OH and CH₃CHO emission lines (see Table 1). For comparison, the

CO (1–0) (white contours; Gueth et al. 1996) and NH₂CHO (4_{1,4}–3_{1,3}) (cyan contours; Codella et al. 2017) spatial distributions are also reported.

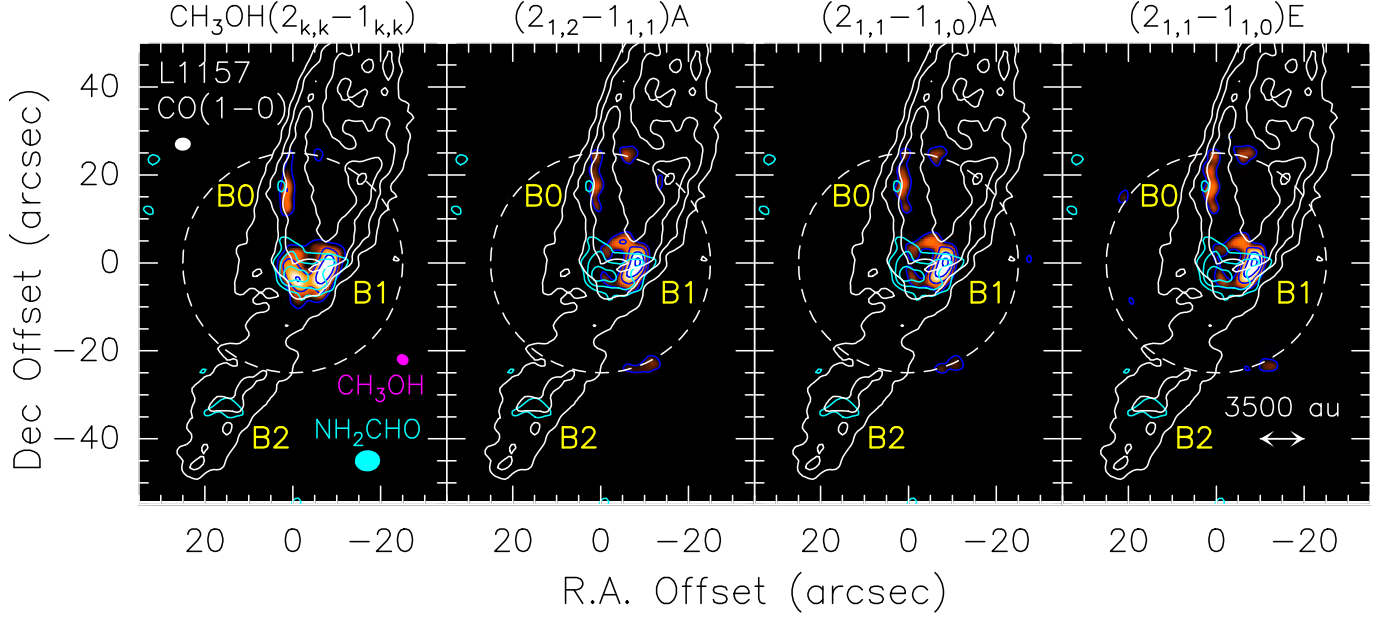


Fig. A.1. L1157 southern blueshifted lobe in CO (1–0) (white contours; Gueth et al. 1996) and NH₂CHO (4_{1,4}–3_{1,3}) (cyan contours; Codella et al. 2017). The precessing jet ejected by the central object L1157-mm (outside the frame and toward the northwest) excavated several clumpy cavities, named B0, B1, and B2, respectively. The maps are centred at $\alpha(\text{J2000}) = 20^{\text{h}}39^{\text{m}}10^{\text{s}}.2$, $\delta(\text{J2000}) = +68^{\circ}01'10''.5$ ($\Delta\alpha = +25''$ and $\Delta\delta = -63''.5$ from the L1157-mm protostar). Map (in colour scale) of the sum of the CH₃OH ($2_{-1,2}-1_{-1,1}$) E, ($2_{0,2}-1_{0,1}$) A, and ($2_{0,2}-1_{0,1}$) E emission lines (labelled $2_{k,k}-1_{k,k}$), ($2_{1,2}-1_{1,1}$) A, ($2_{1,1}-1_{1,0}$) A, and ($2_{1,1}-1_{1,0}$) E (integrated over the whole velocity range). For the CO image, the first contour and step are 6σ ($1\sigma = 0.5 \text{ Jy beam}^{-1} \text{ km s}^{-1}$) and 4σ , respectively. The first contour and step of the NH₂CHO map (cyan contours) correspond to 3σ ($15 \text{ mJy beam}^{-1} \text{ km s}^{-1}$) and 1σ , respectively. The dashed circle shows the primary beam of the CH₃OH images ($64''$). The first contour and step are 3σ ($56, 32, 45$, and $23 \text{ mJy beam}^{-1} \text{ km s}^{-1}$ for $2_{k,k}-1_{k,k}$, $2_{1,2}-1_{1,1}$ A, $2_{1,1}-1_{1,0}$ A, and $2_{1,1}-1_{1,0}$ E, respectively). Yellow labels are for the B0, B1, and B2 regions: the positions of the different clumps inside these regions are shown in Fig. 1 (see also text, and e.g. Codella et al. 2009, 2017). The cyan, white, and magenta ellipses depict the synthesised beams of the NH₂CHO ($5''.79 \times 4''.81$, PA = -94°), CO ($3''.65 \times 2''.96$, PA = $+88^\circ$), and CH₃OH ($2''.97 \times 2''.26$, PA = -155°) observations, respectively.

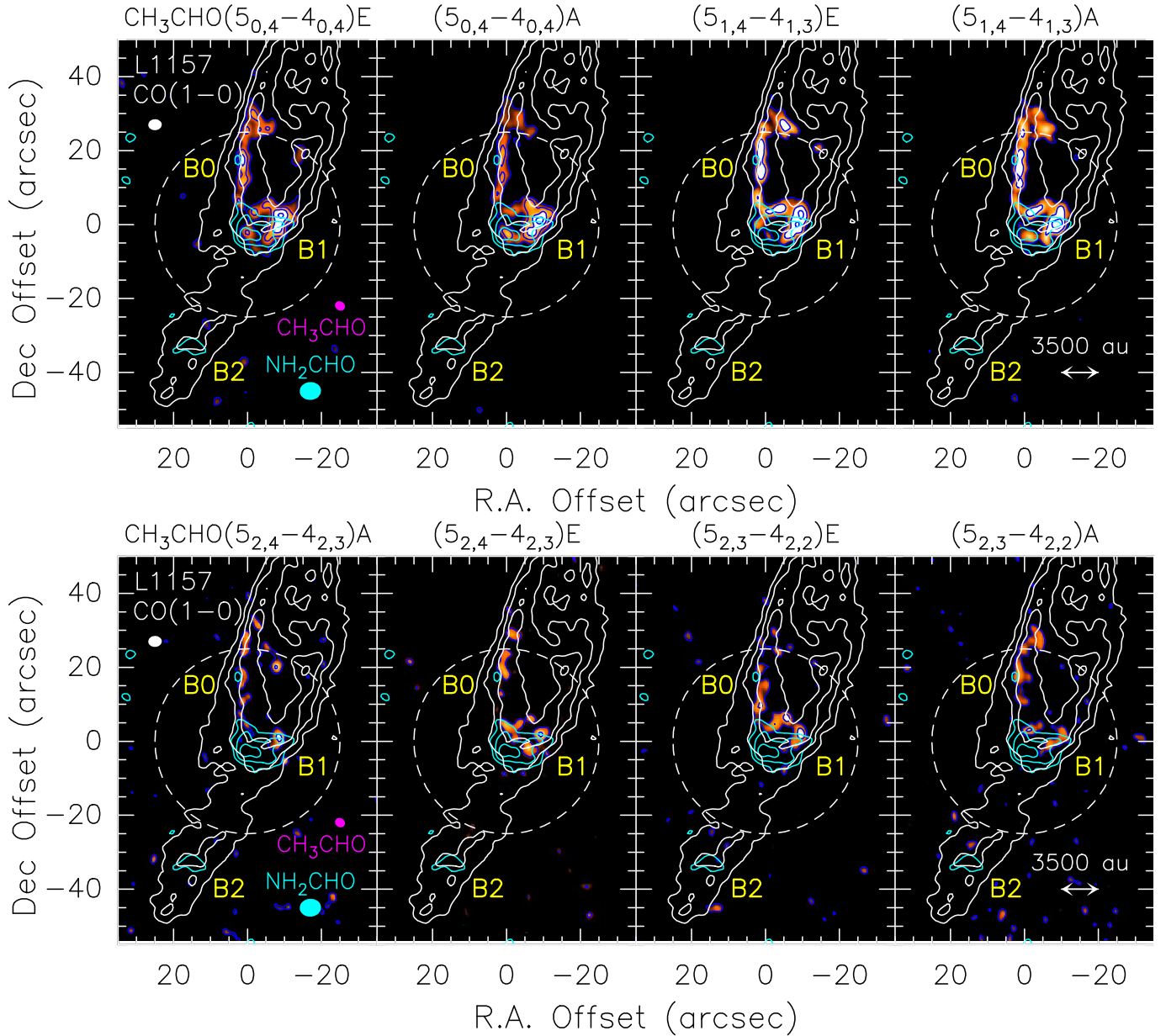


Fig. A.2. L1157 southern blueshifted lobe in CO (1–0) (white contours; Gueth et al. 1996) and NH₂CHO (4_{1,4}–3_{1,3}) (cyan contours; Codella et al. 2017). The precessing jet ejected by the central object L1157-mm (outside the frame and toward the northwest) excavated several clumpy cavities, named B0, B1, and B2, respectively. The maps are centred at $\alpha(\text{J2000}) = 20^{\text{h}}39^{\text{m}}10^{\text{s}}.2$, $\delta(\text{J2000}) = +68^{\circ}01'10''.5$ ($\Delta\alpha = +25''$ and $\Delta\delta = -63''.5$ from L1157-mm). For the CO image, the first contour and step are 6σ ($1\sigma = 0.5 \text{ Jy beam}^{-1} \text{ km s}^{-1}$) and 4σ , respectively. The first contour and step of the NH₂CHO map (cyan contours) correspond to 3σ ($15 \text{ mJy beam}^{-1} \text{ km s}^{-1}$) and 1σ , respectively. The dashed circle shows the primary beam of the CH₃CHO images ($64''$). *Upper panels:* map (in colour scale) of line emission (integrated over the whole velocity range) due to CH₃CHO ($5_{0,4}$ – $4_{0,4}$) E, ($5_{0,4}$ – $4_{0,4}$) A, ($5_{1,4}$ – $4_{1,3}$) E, and ($5_{1,4}$ – $4_{1,3}$) A, respectively. The first contour and step are 3σ : $7 \text{ mJy beam}^{-1} \text{ km s}^{-1}$ ($5_{0,4}$ – $4_{0,4}$ E, $5_{0,4}$ – $4_{0,4}$ A), and $8 \text{ mJy beam}^{-1} \text{ km s}^{-1}$ ($5_{1,4}$ – $4_{1,3}$ E, and $5_{1,4}$ – $4_{1,3}$ A). The cyan, white, and magenta ellipses depict the synthesised beams of the NH₂CHO ($5''.79 \times 4''.81$, PA = -94°), CO ($3''.65 \times 2''.96$, PA = $+88^\circ$), and CH₃CHO ($2''.97 \times 2''.26$, PA = -155°) observations, respectively). Yellow labels are for the B0, B1, and B2 regions: the positions of the different clumps inside these regions are shown in Fig. 1 (see also text, and e.g. Codella et al. 2009, 2017). *Lower panels:* same as upper panels but for the CH₃CHO ($5_{2,4}$ – $4_{2,3}$) A, ($5_{2,4}$ – $4_{2,3}$) E, ($5_{2,3}$ – $4_{2,2}$) E, and ($5_{2,3}$ – $4_{2,2}$) A. The first contour and step are 3σ : $5 \text{ mJy beam}^{-1} \text{ km s}^{-1}$ for $5_{2,4}$ – $4_{2,3}$ A,E, and $4 \text{ mJy beam}^{-1} \text{ km s}^{-1}$ for $5_{2,3}$ – $4_{2,2}$ A,E.